NMR Studies of Lignins. 7. 1 H NMR Spectroscopic Investigation of the Distribution of *Erythro* and *Threo* Forms of β -*O*-4 Structures in Lignins

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The distribution of *erythro* and *threo* forms of β -O-4 structures in spruce and birch lignin was elucidated by ¹H NMR examination of lignin samples (methylated and/or acetylated) and stereoisomers of 11 lignin model compounds of the β -O-4 type. It was found that similar amounts of *erythro* and *threo* forms of such structures were present in spruce lignin while *erythro* forms dominated in birch lignin.

Formulas A-D and A'-D' exemplify structural elements in lignin (acetylated) with phenylpropane units attached to each other by β -O-4 linkages. Such linkages constitute very roughly half of the intermonomer connections in lignins. Structural elements A-D and A'-D' can have the erythro or the threo configuration. ¹³C NMR studies suggest that about equal amounts of the two diastereomeric forms are present in spruce lignin.^{1,2} Similar results could be derived from ¹H NMR studies.^{3,4} Applied to birch lignin, the same approach indicated a rather large proportion of erythro forms of β-O-4 structures.³ Interestingly, recent ¹³C NMR studies² suggested a prevalence of erythro forms of β -O-4 structures in beech lignin, the composition of which resembles that of birch lignin.

The conclusions drawn in earlier 1H NMR studies about the distribution of *erythro* and *threo* forms of β -O-4 structures in lignins were primarily based on studies of the signal from H_{α} in acetate derivatives. $^{3.4}$ We have now been able to corroborate the earlier results by extended model

Spruce lignin consists almost entirely of guaiacyl units. This implies that the main types of β-O-4 structures which have to be considered are the erythro and threo forms of structures A and A'. Since birch lignin is composed of nearly equal amounts of guaiacyl units and syringyl units, one has to take into account a substantially larger number of different β -O-4 structures, namely the erythro and threo forms of structures A-D and A'-D'. The model compound data given in Table 1 cover most of these structural possibilities. It can be concluded from data given in Table 1 that the signal from H_a is shifted 0.03–0.06 δ units upfield when a p-acetoxy group is replaced by a p-alkoxy group. A similar shift is observed when going from threo to erythro (Table 1 and Refs. 5 and 6). Application of these generalizations together with the data given in Table 1 makes it possible to derive δ values for the positions of the

compound studies and examinations of lignin derivatives prepared by methylation of phenolic groups and subsequent acetylation. The 1H NMR data for lignin model compounds of the β -O-4 type given in this paper (Table 1) also provide a good basis for the stereochemical assignment of such compounds.

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Table 1. NMR data for lignin model compounds of the β -O-4 type (solvent, CDCl₃).

				8/p.p.m. vs. TMS (J/Hz)	(ZH/C)		
Compound	ř	H _e	Η(γ ₁)	H(γ ₂)	Ϊ	осн³	СН,СО
1a (erythro)	6.02 (5.1)	4.70	4.23 (4.0, 11.8)	4.42 (5.9, 11.8)		3.79, 3.86, 3.87	2.02, 2.07
1a (threo)	6.08 (7.0)	4.65	4.01 (5.7, 11.8)	4.28 (4.0, 11.8)		3.83, 3.86, 3.87	2.00, 2.01
	6.08 (5.4)	4.67	4.25 (4.0, 11.8)	4.46 (5.9, 11.8)		3.77, 3.81	2.02, 2.09, 2.30
1b (threo)	6.12 (6.4)	4.63	4.06 (5.8, 11.9)	4.31 (4.5, 11.9)		3.81, 3.82	1.99, 2.05, 2.30
1c (erythro) ^b	6.01 (5.3)	4.69	4.21 (3.9, 11.9)	4.41 (5.9, 11.9)	5.02		
1c (threo) ^b	6.07 (6.8)	4.63	4.01 (5.8, 12.0)	4.27 (4.1, 12.0)	5.04		
$\overline{}$	6.11 (6.3)	4.62	4.06 (5.7, 11.8)	4.31 (4.6, 11.8)	5.03		1.99, 2.05, 2.09, 2.29
	(9.9) 60.9	4.50	3.93 (4.6, 11.9)	4.32 (4.5, 11.9)	2.29	3.84°	1.99°
	6.12 (6.2)	4.47	3.97 (4.4, 11.8)	4.36 (5.2, 11.8)	2.30		2.00, 2.02, 2.32
2c (threo)	6.12 (6.2)	4.53	3.98 (4.6, 11.8)	4.36 (4.8, 11.8)	5.05	3.77, 3.80	2.00, 2.01, 2.10, 2.32
	6.00 (5.7)	4.69	4.27 (4.3, 11.7)	4.44 (5.8, 11.7)			2.04, 2.09
	6.05 (6.5)	4.63	4.05 (5.9, 11.8)	4.31 (4.3, 11.8)			2.01, 2.05
	6.05 (5.7)	4.67	4.29 (4.1, 12.0)	4.46 (5.4, 12.0)			2.04, 2.09, 2.32
3b (threo)	6.09 (6.2)	4.62	4.09 (5.7, 11.8)	4.33 (4.6, 11.8)		3.80°, 3.81	2.00, 2.06, 2.32
	6.04 (5.7)	4.66	4.28 (4.0, 12.0)	4.45 (5.5, 12.0)	5.05		
	6.08 (6.2)	4.61	4.08 (5.7, 11.8)	4.32 (4.5, 11.8)	5.03	3.81⁴	2.00, 2.07, 2.10, 2.32
4 (threo)	6.15 (6.2)	4.52	•	(4.8,	5.02	3.77°, 3.81	1.99, 2.00, 2.10, 2.30

*Multiplet. Prom analysis of a diastereomeric mixture; data for methoxyl and/or acetyl groups are lacking. 6H. 49H.

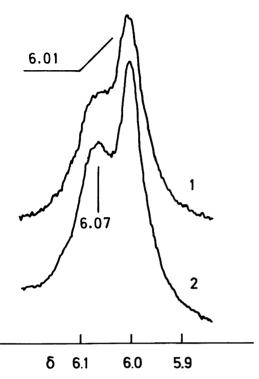


Fig. 1. Range δ 5.9–6.1 of the 1H NMR spectrum of acetylated birch lignin. Spectrum 2 appeared on irradiation at δ 4.6.

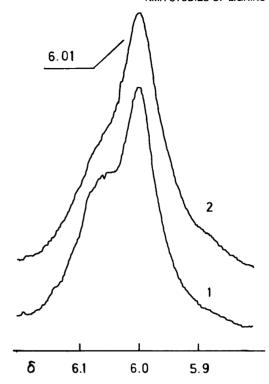


Fig. 2. Range δ 5.9–6.1 of the 'H NMR spectrum of methylated and acetylated birch lignin (Spectrum 2). For comparison, the corresponding range of the spectrum of acetylated birch lignin is included in the figure (Spectrum 1).

signals from H_{α} in all the structural elements A-D and A'-D'.

Fig. 1 shows the region of the ¹H NMR spectrum of acetylated birch lignin in which signals from H_a in β -O-4 structures are located (the highest point of the peak appears at δ 6.01; the complete spectrum is given in Ref. 7). Decoupling by irradiation at δ 4.6 (the position for H_B in β -O-4 structures, cf. Table 1) results in changes of the peak at δ 6.01; it is resolved into two peaks (δ 6.01 and δ 6.07) separated by a shallow minimum at δ 6.04 (Fig. 1). Peak positions (cf. Table 1) together with the decoupling experiments show that the lignin peak at $\delta \approx 6$ is due to signals from H_{α} in β -O-4 structures. The signal from H_{α} in all the threo forms (and some erythro forms) of structures A-D and A'-D' should be located above δ 6.04; this could be concluded from the data in Table 1 together with applications of the

generalizations discussed above. Accordingly, only signals from *erythro* forms will appear below δ 6.04. The highest point (δ 6.01) of the lignin peak is found below this δ value. It follows from what has been said above that this suggests a predominance of *erythro* forms of β -O-4 structures in birch lignin. Additional support for this conclusion comes from the fact that the spectrum of acetylated birch lignin exhibits a peak at δ 4.437 which could be attributed to H_{γ} in *erythro* forms of β -O-4 structures (Table 1).

Erythro forms of β -O-4 structures with p-acetoxy groups (formulas A'-D') do not contribute to the fraction of the lignin peak located below δ 6.04, since such groups cause a downfield shift of the signal from H_{α} to values above δ 6.04. This is apparently a drawback in estimates of the threo/erythro ratio, although the amount of units with phenolic groups (or p-acetoxy groups) is rather

A' $R = COCH_3$, R' = R'' = HB R = C in adjacent unit, $R' = R'' = OCH_3$ B' $R = COCH_3$, $R' = R'' = OCH_3$ C R = C in adjacent unit, $R' = OCH_3$, R'' = HC' $R = COCH_3$, $R' = OCH_3$, R'' = H

 $R = COCH_3$, R' = H, $R'' = OCH_3$

R = C in adjacent unit, R' = H, R'' = OCH,

small in birch lignin (about $18 \%^7$). To eliminate this source of error we methalited the phenolic groups prior to acetylation. The types of β -O-4 structures in this derivative are limited to those represented by formulas A-D. As could be expected, the fraction of the lignin peak which is located below δ 6.04 is larger in this derivative (Fig. 2). This strengthens the arguments for a predominance of *erythro* forms of β -O-4 structures in birch lignin. It should, however, be noted that the peak still exhibits an inflexion at about δ 6.07 which indicates that the amount of *threo* forms present is not negligible.

The distribution of threo and erythro forms in spruce lignin has earlier been discussed on the basis of the contributing signals from H_{α} in structures A and A'. We examined methylated and acetylated spruce lignin. In this derivative only structures of type A have to be considered. The lignin peak from H_{α} in β -O-4 structures ($\delta \approx 6.05$) was found to consist of approximately equal contributions from erythro ($\delta(H_{\alpha})$ 6.01) and threo ($\delta(H_{\alpha})$ 6.07) forms which is in agreement with earlier results.

The stereochemical assignment of the *erythro* and *threo* forms of compounds *1a* and *1b* has been firmly established in earlier work (*cf.* Ref.

$$\gamma$$
 CH_2OCOCH_3
 β
 HC
 OCH_3
 OCH_3
 OR

1a. R = CH₃, R' = R" = R"' = H
1b. R = COCH₃, R' = R" = R" = H
1c. R = CH₃, R' = R" = H, R"" = CH₂OCOCH₃
1d. R = COCH₃, R' = R" = H, R"" = CH₂OCOCH₃
2a. R = CH₃, R' = R" = OCH₃, R" = CH₃
2b. R = COCH₃, R' = R" = OCH₃, R"" = CH₃
2c. R = COCH₃, R' = R" = OCH₃, R"" = CH₂OCOCH₃
3a. R = CH₃, R' = OCH₃, R" = R"" = H
3b. R = COCH₃, R' = OCH₃, R" = H, R"" = H
3c. R = COCH₃, R' = OCH₃, R" = H, R"" = CH₃OCOCH₂
4. R = COCH₃, R' = H, R" = OCH₃, R"" = CH₂OCOCH₃

8). ¹H NMR spectral comparisons with these compounds could therefore be used for the assignment of the *threo* and *erythro* forms of the other compounds in Table 1. The stereochemical relationships have actually already previously been elucidated for several of the compounds by the same method.^{5,6} A modern high frequency spectrometer was used in the present work. This considerably facilitates comparative studies of the current type, since the spectra are well resolved (the improved resolution of the signals from H_y is particularly notable) and very accurate spectral data can be obtained.

Additional conclusions regarding 1H NMR spectral properties of β -O-4 model compounds could be drawn from the data in Table 1. An exchange of the hydrogen at C-1' for a substituent more resembling a lignin side-chain (CH₂O-COCH₃) leaves the signals from H_{α} , H_{β} , and H_{γ} practically unaffected. A replacement of the

same hydrogen atom for a methyl group causes a slight upfield shift (0.06 δ units) of the signal from $H_{\textrm{B}}$.

Experimental

Lignin model compounds were prepared according to procedures described in Refs. 6, 9 and 10 or modifications of them.¹¹

Methylation of lignin from birch (Betula verrucosa). The procedure described by Fenn and Kirk¹² was adapted. Milled wood lignin from birch (0.63 g), K₂CO₃ (0.48 g), and methyl iodide (0.21 ml) in DMF (5 ml) were stirred magnetically overnight (argon atmosphere). Methyl iodide (0.21 ml) was then added and stirring continued for 3 h. The DMF solution was dripped into 300 ml of an aqueous solution of 1 % Na₂SO₄ acidified with a few drops of acetic acid (magnetic stirring). The precipitate formed was centrifuged off and dried. The product was methvlated a second time according to the same procedure. 1H NMR examination of the acetate derivative of the product from two methylations showed that the methylation of phenolic groups was almost complete [the peak due to phenolic acetate ($\delta \approx 2.3$) was practically absent].

¹H NMR spectra were recorded on a 270 MHz

instrument working in the pulse Fourier mode (Bruker WH 270). Chloroform-d was used as solvent (internal reference, TMS). Temperature, 300 K. Number of scans (lignin spectra), 1000.

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