Short Communications

A Convenient Synthesis of Pyrrole-2,5-dicarboxaldehyde

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The title compound (4) was needed as starting material for the synthesis of pyrroles formed in the Maillard reaction. Several methods for preparing 4 from pyrrole have been described. Their most important characteristics are:

Unfortunately, the most promising of these methods 3 in our hands gave <10% yield of 4 , as shown by GC analysis of the crude reaction mixture. We now report the synthesis of 4 in 38% yield by the four-step sequence shown in Scheme 1.

The base-catalyzed reaction with ethyl cyanoacetate is a well-known method for protecting pyrrolecarboxaldehydes, 6 and the sequence $1\rightarrow2\rightarrow3$ was described recently. 7 Since there is no need for chromatography or for purification of the intermediates 1-3, the present synthesis of 4 is simple and may be performed on a large scale.

ArH
$$POCl_{3} \downarrow HCONMe_{2}$$

$$ArCHO$$

$$ArCHO$$

$$2$$

$$Et_{2}NH \downarrow NCCH_{2}CO_{2}Et$$

$$ArCH=C-CN \quad \frac{POCl_{3}}{HCONMe_{2}} \quad OHC$$

$$2$$

$$Ar = 2-pyrrolyl$$

$$ArCH=C+CN \quad \frac{CO_{2}Et}{HCONMe_{2}} \quad \frac{CO_{2}Et}{H}$$

Scheme 1.

Experimental. Compounds $1,^8$ 2 and 3^7 were prepared according to the literature (2 and 3 are probably the E isomers 9). However, 1 was neither distilled nor crystallized but converted directly to 2 (154 g, 81 % calc. on pyrrole) after evaporation of the solvents. A more detailed procedure for the preparation of 3 is given below (cf. also Ref. 8).

Ethyl α-cyano-5-formyl-2-pyrroleacrylate (3). Phosphoryl chloride (150 g, 90 ml, 0.98 mol) was added over 20 min to N,N-dimethylformamide (72 g, 76 ml, 0.98 mol), stirred and kept at 10-20 °C by cooling with an ice-salt bath. After stirring for another 15 min without cooling, 1,2-dichloroethane (450 ml) was added. The stirring and cooling were continued while a suspension of ethyl α-cyano-2-pyrroleacrylate (2, 154 g, 0.81 mol) in 1,2-dichloroethane (675 ml) was added over 30 min at ca. 5 °C. The mixture was then refluxed for 15 min (HCl evolution!). Aqueous 4.0 M sodium acetate (1.25 l) was added over ca. 5 min at 25-30 °C to the vigorously stirred mixture, which was then refluxed for another 15 min. Crystallization overnight vielded 3 (141 g, 80%).

Pyrrole-2,5-dicarboxaldehyde (4). Aqueous 3 M sodium hydroxide (600 ml) was refluxed for 2 h with 3 (32.7 g, 0.150 mol), which dissolved within a few minutes. The solution was acidified below 20 °C with 2 M sulfuric acid to pH 4.5 and then extracted with ethyl acetate $(500 + \bar{1}00 \text{ ml})$. Rotary evaporation of the extract and crystallization of the residue from water (100 ml) yielded nearly pure but discoloured 4 (9.8 g), m.p. 117 – 121 °C. Processing of the mother liquor raised the yield to 10.9 g (59%; 38% calc. on pyrrole). Sublimation in vacuo or recrystallization from water or toluene in the presence of charcoal yielded pure 4, m.p. 123-124 °C (sealed tube, lit.⁵ 122-123 °C). The mass, IR and ¹H NMR spectral data agreed with those reported.2,4,5 At half neutralization of 4 with aqueous 0.1 M sodium hydroxide, pH 10.0 was observed, indicating that p K_a is close to 10.

Acknowledgements. We thank Mr. Lars Adolfsson for skilled technical assistance and Dr. Per-Åke Pernemalm and Professor Olof Theander for their kind interest in the present work. This work was supported by a grant from the Swedish Board for Technical Development.

- Olsson, K., Pernemalm, P.-Å. and Theander, O. Prog. Food Nutr. Sci. 4 (1980). In press.
- Cresp, T. M. and Sargent, M. V. J. Chem. Soc. Perkin Trans 1 (1973) 2961.
- Bergman, J., Renström, L. and Sjöberg, B. Tetrahedron 36 (1980) 2505.
- 4. Severin, T. and Ipach, I. Chem. Ber. 108 (1975) 1768.
- 5. Loader, C. E. and Anderson, H. J. Synthesis (1978) 295.
- Fischer, H. and Weiss, B. Ber. Dtsch. Chem. Ges. 57 (1924) 602; Paine, J. B., III, Woodward, R. B. and Dolphin, D. J. Org. Chem. 41 (1976) 2826.
- 7. Olsson, K. and Pernemalm, P.-Å. Acta Chem. Scand. B 33 (1979) 125.
- Silverstein, R. M., Ryskiewicz, E. E. and Willard, C. Org. Synth. Coll. Vol. 4 (1963) 831.
- Phillips, W. M. and Currie, D. J. Can. J. Chem. 47 (1969) 3137; Currie, D. J., Lough, C. E., McClusky, F. K. and Holmes, H. L. Can. J. Chem. 47 (1969) 3147.

Received March 16, 1981.