

# Stability Constants for Chloride Complexes of Copper(II) in Sulphuric Acid Solution

MARIANNE OHLSON and NILS-GÖSTA VANNERBERG

Department of Inorganic Chemistry, Chalmers University of Technology and University of Göteborg,  
P.O. Box, S-402 20 Göteborg 5, Sweden

Spectrophotometric methods have been used to determine the stability constants of copper(II) chloride complexes in an ionic medium of 9 M sulphuric acid at 25 °C. The total molarity was held constant at 9 M.

Preliminary constants were obtained by graphical methods and were then refined with the generalized least squares program "LETAGROP". The experimental data could best be explained in terms of the following equilibria and stability constants:



The errors given correspond to an error of  $3\sigma$ , where  $\sigma$  is the standard deviation in  $\log \beta$ .

The formation of complexes between copper(II) ions and chloride ions in aqueous solution has mainly been studied by spectrophotometric methods, but a few investigators have used ion exchange or freezing point methods. The formation of complexes between Cu(II) ions and chloride ions cannot be investigated potentiometrically using a copper amalgam electrode or a silver chloride electrode, as both these electrodes reduce Cu(II) to Cu(I) in the presence of chloride ions.

The values of the stability constants (*cf.* Table 1) determined in the various investigations differ from each other, owing to different ionic media and methods of measurement. Some of the constants have been extrapolated to zero ionic strength.<sup>1,8,9,11</sup> From the values given in Table 1, it is evident that the complexes are very weak, the stability constant,  $\beta_1$ , for the

reaction  $\text{Cu}^{2+} + \text{Cl}^- \rightleftharpoons \text{CuCl}^+$  having an approximate value of 1.<sup>1-6,8,11</sup> Some authors have found a higher value for  $\beta_1$ .<sup>7,8,10</sup> This can be attributed to a more concentrated ionic medium, to the neglection of the second complex, or to the method of calculation. The uncertainty of the stability constant,  $\beta_2$ , for the reaction  $\text{Cu}^{2+} + 2\text{Cl}^- \rightleftharpoons \text{CuCl}_2^-$  is larger than that of  $\beta_1$ , but its value is probably less than 1.<sup>1,2</sup>

It has been claimed that copper(II) forms four mononuclear complexes with chloride ions, namely  $\text{CuCl}^+$ ,  $\text{CuCl}_2^-$ ,  $\text{CuCl}_3^{2-}$  and  $\text{CuCl}_4^{3-}$ .<sup>1,6,10</sup> On account of the weak complexity, it has, however, proved difficult to determine the stability constants of the higher complexes. Using spectrophotometric measurements, with different metal chlorides as ionic media, Bjerrum<sup>1</sup> was, however, able to determine approximate values for  $\beta_3$  and  $\beta_4$ . The values determined by Morris and Short<sup>10</sup> differ, however, considerably from those determined by Bjerrum.

In this work, the copper(II) chloride system has been studied spectrophotometrically in 9 M sulphuric acid at 25 °C, sulphuric acid being chosen as medium since it seemed likely that the copper(II) complexes would be stronger in 9 M sulphuric acid than in dilute aqueous solution. Sulphuric acid medium has, moreover, industrial interest, since anhydrous chlorides of the transition metals can be obtained by the distillation of such solutions.

Table 1. Survey of reported values for the stability constants of copper(II) chloride complexes.

Author	Ref.	Method	Medium	Temp. °C	$\beta_1$ (M <sup>-1</sup> )	$\beta_2$ (M <sup>-2</sup> )	$\beta_3$ (M <sup>-3</sup> )	$\beta_4$ (M <sup>-4</sup> )
Bjerrum	1	spectr.	extrap. to 0	22	1	0.2	$8 \times 10^{-3}$	$8 \times 10^{-5}$
McConnell and Davidsson	2	spectr.	1 M HClO <sub>4</sub>	25.2	1.3	0.3		
Näsänen	3	spectr.	extrap. to 0	25	1.2			
Kruh	4	spectr.	1 M HClO <sub>4</sub>	22	0.27			
Lister and Rosenblum	5	spectr.	2 M NaClO <sub>4</sub>	25	1.22			
Andreev and Sapozhinkova	6	spectr.	HCl		1			
Libus	11	spectr.	extrap. to 0	25	1.63			
Faucherre and Crego	7	fp	sat. KNO <sub>3</sub>		2.70			
Kenttämäa	8	fp	sat. KClO <sub>3</sub>		1.50			
		fp	sat. KClO <sub>4</sub>		4.60			
			extrap. to 0		8.90			
Tre'millon	9	ion exch.	1.5 M NaNO <sub>3</sub>		0.4			
Morris and Short	10	ion exch.	0.69 M HClO <sub>4</sub>	20	9.6	4.92	3.52	1.0
Ohlson and Vannerberg		spectr.	9 M H <sub>2</sub> SO <sub>4</sub>	25	10.7	39.6	59	163

## EXPERIMENTAL

*Chemicals and analysis.* Stock solutions of copper(II) sulphate were prepared by dissolving copper(II) sulphate (Merck *p.a.*) in 9 M sulphuric acid and the copper content was determined electrogravimetrically.<sup>13</sup> Sulphuric acid was prepared by dilution of conc. sulphuric acid (Merck *p.a.*) and its concentration was calculated from the experimental density,<sup>14</sup> determined with an areometer, graduated from 1.470 to 1.520 g cm<sup>-3</sup> (accuracy  $\pm 0.001$  g cm<sup>-3</sup>). Sodium chloride (Merck *p.a.*) was dried at 110 °C and weighed.

The light absorption measurements in the range 350–385 nm were made with a Beckman spectrophotometer, Model DU-2. The measurements in the ultraviolet range 260–300 nm were made on a Gilford 240 spectrophotometer. Matched quartz cells of path lengths 0.1, 0.2, 0.5, 1.0 cm were employed, these being calibrated before use. During the measurements, the sample compartment was thermostated to  $25.0 \pm 0.1$  °C.

Four series of solutions of copper(II) chloride were prepared by adding accurately weighed sodium chloride to a solution of copper(II) sulphate. In each series, the total copper ion concentration ( $B$ ) was kept constant ( $B$ : 0.005, 0.010, 0.020, 0.050 M), while the total chloride concentration ( $A$ ) was varied between 0.004 and 0.600 M. The total molarity was held constant at 9 M by the addition of sulphuric acid.

In one and the same series of measurements,  $B$  and  $l$  were kept constant and in the different series of measurements the product  $lB$  was kept constant.<sup>15</sup>

The wave lengths employed, the numbers of solutions and the numbers of measurements are given in Table 2.

Table 2. The wave lengths used, the numbers of solutions and the chloride concentrations.

Wave lengths (nm)	Number of solutions	Number of measured values	Chloride concentration (M)
260, 265, 270, 275, 280, 290, 300	63	354	0.004–0.064
350, 355, 360, 365, 370, 375, 380, 385	113	658	0.0192–0.600

## LIST OF SYMBOLS

- $A$  total concentration of chloride ions,  $\text{Cl}^-$   
 $a$  free      »      »      »      »  
 $B$  total      »      » copper ions,  $\text{Cu}^{2+}$   
 $b$  free      »      »      »      »  
 $A_s$  absorbance  
 $l$  optical path length  
 $\varepsilon$  apparent molar absorptivity  
 $\beta_i$  equilibrium constant for the reaction  
 $\text{Cu}^{2+} + i\text{Cl}^- \rightleftharpoons \text{CuCl}_i^{(2-i)+}$   
 $\varepsilon_i$  molar absorptivity for the complex  
 $\text{CuCl}_i^{(2-i)+}$   
 $n$  mean ligand number

## MEASUREMENTS

The measurements were performed spectrophotometrically, the method of corresponding solutions<sup>1,15-18</sup> being used. In order to find suitable wave lengths, spectra were recorded for copper chloride solutions with constant  $B$  and varying  $A$ . The spectra showed absorption maxima at 250, 375, and 800–900 nm (Figs. 1, 2). The absorption of the  $\text{Cu}^{2+}$  ion, which probably exists as a sulphate complex in the sulphuric acid solution, increases with decreasing wave length in the UV range, and shows an absorption maximum at 810 nm (Fig. 1). The absorption bands of the complexes are displaced towards longer wave lengths with increasing chloride concentrations, their band

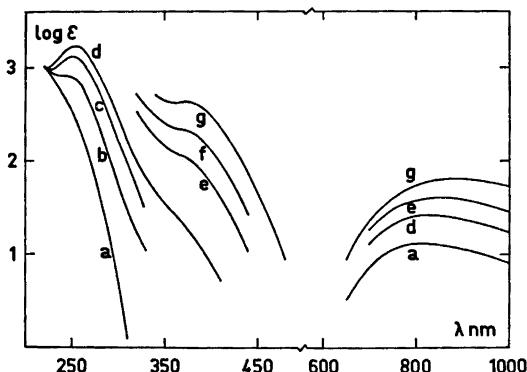


Fig. 1.  $\log \varepsilon$  as a function of  $\lambda$  for copper(II) chloride solutions.  $B=0.010 \text{ M}$  and the following values of  $A$  were used: a. 0 M; b. 0.024 M; c. 0.060 M; d. 0.100 M; e. 0.300 M; f. 0.500 M; g. 1.000 M.

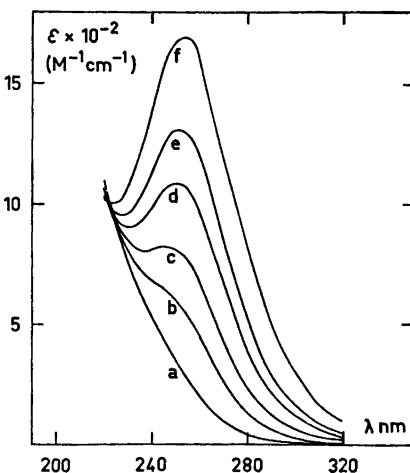


Fig. 2.  $\varepsilon$  data as a function of  $\lambda$  for solutions with  $B=0.010 \text{ M}$  and different values of  $A$ : a. 0 M; b. 0.012 M; c. 0.024 M; d. 0.044 M; e. 0.060 M; f. 0.100 M.

widths also increasing considerably. Solutions with low  $A$  show one isosbestic point at 224 nm.

The majority of the measurements (658 values) have been carried out in the range 350–385 nm, where the higher complexes can be detected. Another advantage of this range is that the  $\text{Cu}^{2+}$  ion does not absorb there. The remaining 354 measurements were made in the 260–300 nm range (Table 2). Only the stability constants of the two first complexes,  $\text{CuCl}^+$  and  $\text{CuCl}_2$ , could be determined from measurements in the latter range, owing to the strong absorption of both the  $\text{Cu}^{2+}$  ion and the copper(II) chloride complexes. No measurements have been carried out in the 800–900 nm range, because of the small absorptivities of the complexes. The absorption of chloride ions is negligible.

## TREATMENT OF THE DATA

The absorbance,  $A_s$ , of a solution is the product of the apparent molar absorptivity,  $\varepsilon$ , the optical path length,  $l$ , and the total concentration of the absorbing substance,  $B$ ,

$$A_s = l\varepsilon B = l \sum_{i=0}^N \varepsilon_i [\text{BA}_i] = l \sum_{i=0}^N \varepsilon_i \beta_i b a^i \quad (1)$$

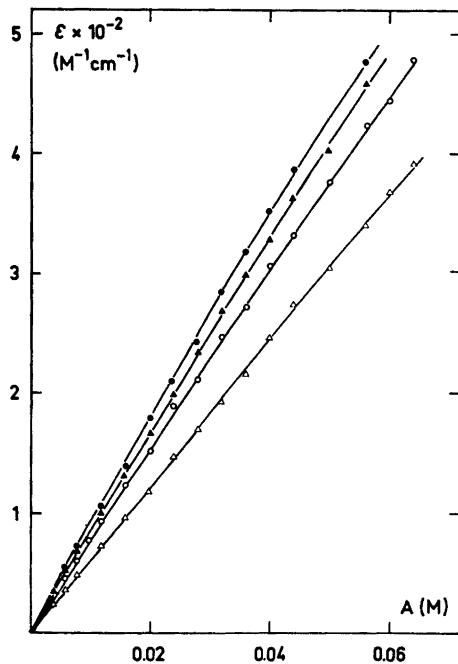


Fig. 3. The function  $\epsilon(A)$  at different values of  $B$  at 280 nm.  $\triangle B=0.05$  M;  $\circ B=0.02$  M;  $\bullet B=0.01$  M;  $\blacksquare B=0.005$  M.

where  $\epsilon_0, \epsilon_1, \dots, \epsilon_N$  are the molar absorptivities for  $\text{Cu}^{2+}, \text{CuCl}^+, \dots, \text{CuCl}_{N(2-N)}^+$ , respectively.

The apparent molar absorptivity is defined as

$$\epsilon = \frac{\sum_{i=0}^N \epsilon_i [BA_i]}{B} = \left[ \sum_{i=0}^N \epsilon_i \beta_i a^i \right] / \left[ 1 + \sum_{i=1}^N \beta_i a^i \right] \quad (2)$$

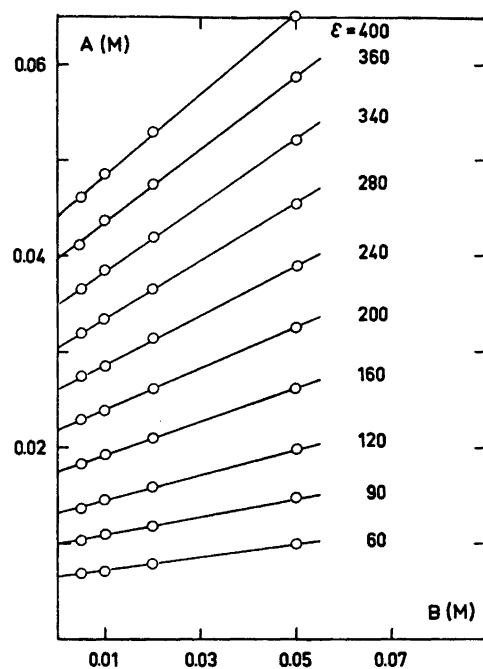


Fig. 4.  $A$  versus  $B$  at different  $\epsilon$  at 280 nm.

From eqn. (2) it is seen that  $\epsilon$  is solely a function of  $a$ , if only mononuclear complexes are present. The mean ligand number  $\bar{n}$  is defined as

$$\bar{n} = \frac{A - a}{B} = \left[ \sum_{i=1}^N i \beta_i a^i \right] / \left[ 1 + \sum_{i=1}^N \beta_i a^i \right] \quad (3)$$

and  $\bar{n}$  is thus only a function of  $a$ , this gives

$$A = a + \bar{n}B \quad (4)$$

Table 3. Corresponding values of  $B$ ,  $A$ ,  $a$  and  $\bar{n}$  obtained from the  $\epsilon(A)$ -curve for  $\lambda=280$  nm.

$B$	$A$ M				$a$	$-\log a$	$\bar{n}$
	0.005	0.01	0.02	0.05			
60	0.0068	0.0071	0.0079	0.0099	0.0064	2.19	0.07
90	0.0103	0.0108	0.0117	0.0148	0.0098	2.01	0.10
120	0.0137	0.0145	0.0158	0.0198	0.0132	1.88	0.13
160	0.0183	0.0192	0.0211	0.0262	0.0174	1.76	0.18
200	0.0229	0.0239	0.0262	0.0325	0.0219	1.66	0.21
240	0.0274	0.0286	0.0314	0.0389	0.0261	1.58	0.26
280	0.0319	0.0335	0.0366	0.0454	0.0304	1.52	0.30
340	0.0388	0.0411	0.0448	0.0554	0.0373	1.43	0.36
360	0.0411	0.0437	0.0475	0.0587	0.0398	1.40	0.38
400	0.0460	0.0488	0.0531	0.0654	0.0447	1.35	0.41

Table 4. Corresponding values of  $\log a$  and  $\bar{n}$ , obtained from  $\varepsilon(A)$ -curves for 15 different wave lengths, followed by  $(\bar{n}_{\text{calc}} - \bar{n}) \times 10^3$ , which has been calculated with the "LETAGROP VRID" program.

$\lambda \text{ nm}$	$\lambda \text{ nm}$	$\lambda \text{ nm}$	$\lambda \text{ nm}$
385 -0.9340 C. 93000 4.99	-0.5260 1.90000 -1.29	290 -2.1760 0.04600 1.73	
-0.8780 C. 99000 45.53	-0.4980 1.96000 23.89	-2.0260 0.08000 3.31	
-0.8330 1.11000 13.55	-0.4740 2.04000 16.18	-1.9600 C. 0.11000 0.38	
-0.7930 1.17000 36.20	-0.4500 2.11000 23.39	-1.9000 0.12000 6.21	
-0.7610 1.33000 23.49	-0.4280 2.20000 2.93	-1.8240 0.14500 7.81	
-0.7300 1.43000 11.17	-0.4060 2.30000 17.19	-1.7750 0.16800 2.13	
-0.6990 1.53000 60.17	-0.3840 2.41000 76.73	-1.7280 C. 1.87000 1.48	
-0.6630 1.69000 42.58	-0.3670 2.48000 83.17	-1.6880 0.21000 4.45	
-0.5530 1.71000 6.89	-0.3470 2.55000 39.87	-1.6520 0.22500 2.86	
-0.5750 1.78000 25.61	365 -1.1690 0.60000 -0.19	-1.6160 0.24100 -1.03	
-0.5410 1.86000 6.15	-0.1090 0.70000 -4.27	-1.5880 C. 0.26500 -10.24	
-0.5110 1.92000 24.13	-0.0270 0.76000 26.78	-1.5560 0.27700 -4.32	
-0.4840 2.00000 27.10	-0.9710 0.82000 53.34	-1.5300 0.29500 -6.91	
-0.4580 2.08000 28.23	-0.8910 0.10100 1.38	-1.5020 0.31000 4.43	
-0.4340 2.15000 33.92	-0.8270 0.19000 54.80	-1.4790 C. 0.34000 -13.35	
-0.4100 2.20000 60.09	-0.7770 0.28000 23.61	-1.4570 0.34200 -10.50	
-0.3890 2.29000 36.90	-0.7230 0.40000 35.49	-1.4270 0.35100 6.25	
-0.3630 2.36000 33.65	-0.6700 0.47000 7.11	-1.4090 0.37900 -8.22	
-0.3390 2.44000 -0.18	-0.6410 0.55000 -5.59	280 -2.2480 0.05800 -1.35	
-0.3110 2.55000 -39.34	-0.6060 0.65000 -23.30	-2.1920 0.07000 -2.74	
380 -1.2030 0.57000 -8.75	-0.5750 0.76000 -5.60	-2.1220 0.08000 -1.25	
-1.1550 C. 66000 -37.66	-0.5440 0.84000 4.94	-2.0610 C. 0.09000 0.29	
-1.0930 0.72000 -25.60	-0.5160 0.90000 28.93	-2.0100 0.10000 1.20	
-1.0410 0.79000 -23.66	-0.4910 0.20000 5.45	-1.9620 0.11000 2.62	
-0.9730 C. 88000 -9.89	-0.4670 0.08000 0.03	-1.8880 0.13100 4.07	
-0.9120 0.94000 33.42	-0.4440 0.21000 12.31	-1.8120 C. 0.15000 6.89	
-0.8690 1.06000 -7.46	-0.4210 0.20000 25.14	-1.7660 0.17500 C. 75	
-0.8280 1.12000 13.16	360 -1.3280 0.47000 -32.53	-1.7080 0.19100 3.44	
-0.7910 1.20000 16.50	-1.2640 C. 52000 -24.51	-1.6500 0.21300 5.35	
-0.7590 1.27000 11.02	-1.2040 0.51000 -26.75	-1.6200 C. 0.23400 3.39	
-0.7330 1.35000 -19.13	-1.1690 0.40000 -20.19	-1.5830 0.25500 2.48	
-0.7050 1.43000 -51.51	-1.1300 0.67000 -23.15	-1.5480 0.27600 1.34	
-0.6830 1.52000 -57.47	-1.0660 C. 72000 19.34	-1.5170 0.30000 -3.91	
-0.6610 1.63000 -58.59	-1.0070 0.81000 6.79	-1.4840 C. 0.32400 -6.69	
-0.6450 1.74000 -70.51	-0.9590 0.86000 32.93	-1.4550 0.34000 -2.90	
-0.5710 1.80000 -34.08	-0.9210 0.92000 37.53	-1.4280 0.35900 -2.48	
-0.5390 1.88000 -20.20	-0.8870 0.00000 18.76	-1.4000 0.37500 2.71	
-0.5100 1.93000 17.17	-0.8540 0.06000 21.43	-1.3760 0.39700 -0.23	
-0.4830 2.00000 30.20	-0.8230 0.11000 33.36	-1.3500 0.41400 4.38	
-0.4590 2.10000 5.09	-0.7970 0.15000 7.65	275 -2.2480 0.06500 1.36	
-0.4360 2.19000 -12.41	-0.7480 0.30000 6.06	-2.1980 0.07800 2.73	
-0.4140 2.24000 7.38	-0.7030 0.37000 42.82	-2.0950 0.08300 0.47	
-0.3930 2.33000 -15.82	-0.6670 0.47000 23.27	-2.0200 0.10000 -1.03	
-0.3740 2.41000 -19.95	-0.6320 0.58000 18.53	-1.9470 0.11400 2.44	
-0.3540 2.45000 -11.99	-0.6000 0.68000 3.44	-1.8870 C. 0.13200 1.00	
-0.3350 2.54000 -42.09	-0.5700 0.77000 -1.19	-1.8350 0.15000 C. 84	
375 -1.1970 0.58000 -12.10	-0.5410 C. 85000 3.85	-1.7890 0.16800 -3.00	
-1.0910 0.69000 7.06	-0.5140 0.92000 15.00	-1.7460 0.18600 -4.76	
-1.0160 C. 83000 -26.84	-0.4890 0.20000 11.62	-1.6690 C. 0.21400 0.15	
-0.9530 0.87500 27.87	-0.4640 0.06000 29.42	-1.6030 C. 0.24800 -1.26	
-0.9070 0.98000 2.34	-0.4410 0.21000 51.78	-1.5740 0.26200 0.46	
-0.8830 1.06000 4.01	-0.4180 0.21000 74.67	-1.5200 0.30200 -7.78	
-0.8250 1.11000 29.27	355 -1.2490 0.50800 14.53	270 -1.5100 0.05000 -1.95	
-0.7660 1.31000 29.70	-1.1820 C. 59400 1.53	-1.4700 0.06300 -2.53	
-0.7140 1.44000 -53.92	-1.0300 0.74000 13.44	-1.2130 0.06300 1.15	
-0.6710 1.58000 -86.38	-0.9390 0.76000 21.23	-1.2120 0.07200 5.52	
-0.6400 1.65000 -7.79	-0.6510 C. 86300 38.52	-2.0030 C. 0.09900 3.79	
-0.5940 1.71000 -9.71	-0.9520 0.88700 17.54	-1.9010 0.12500 3.95	
-0.5620 1.77000 22.03	-0.9170 0.95700 7.56	-1.8170 C. 0.15300 2.17	
-0.5330 1.83000 47.70	-0.8830 0.10000 26.18	-1.7490 0.18000 0.06	
-0.5040 1.91000 55.50	-0.8520 0.10700 68.34	-1.6850 C. 0.21200 -5.12	
-0.4790 2.00000 42.62	-0.8260 0.11900 18.23	-1.6290 0.24000 -6.61	
-0.4450 2.11000 4.51	-0.8000 0.16200 29.28	-1.5790 0.26700 -7.32	
-0.4340 2.20000 -16.08	350 -1.4780 C. 21300 -2.97	-1.5540 C. 0.27900 -5.16	
-0.4130 2.26000 -9.45	-1.6070 0.23400 10.64	-1.5330 0.29600 -1.91	
-0.3920 2.29000 27.36	-1.5480 0.26100 16.34	265 -2.1010 0.07800 4.55	
-0.3730 2.34000 -52.22	-1.4110 0.33300 9.66	-2.0000 C. 0.10000 3.48	
-0.3490 0.80000 1.32	-1.3570 0.42400 13.75	-1.9140 0.12100 4.29	
-0.3160 C. 0.22000 3.04	-1.3160 0.46800 11.54	-1.7770 0.16000 9.38	
-0.2900 0.20000 -25.95	-1.2510 0.47100 0.33	-1.7200 0.19200 -0.22	
-0.2610 0.17000 -2.14	-1.2530 0.50000 8.48	-1.6720 C. 0.23400 -17.23	
-0.2360 0.13000 7.23	-1.2530 0.50000 8.48	-1.5540 C. 0.27900 -5.16	
-0.1960 1.21000 -10.21	300 -2.1900 0.06300 4.57	260 -2.3040 0.05100 1.22	
-0.1760 1.28000 -14.70	-2.0730 0.07900 8.99	-2.1650 0.06500 6.48	
-0.1470 C. 1.33000 5.47	-1.9120 C. 11700 8.84	-2.0660 0.08200 7.29	
-0.1710 1.38000 3.67	-1.7170 0.19100 2.03	-1.9800 C. 0.10300 5.20	
-0.1690 1.46000 -24.98	-1.6500 0.22100 2.10	-1.9400 C. 0.11500 3.27	
-0.1672 1.50000 -9.54	-1.5950 C. 0.25200 1.67	-1.9030 C. 0.12200 6.38	
-0.1613 1.53000 9.69	-1.5330 C. 27600 6.67	-1.8690 C. 0.12700 11.39	
-0.1617 1.64000 -3.68	-1.4700 0.31000 5.46	-1.8350 C. 0.14000 9.16	
-0.1610 1.70000 -19.36	-1.4170 0.36100 3.71		
-0.1595 1.75000 -24.22	-1.3840 0.39200 -1.68		
-0.1554 1.86000 15.44	-1.3520 C. 0.41200 4.68		

Solutions with the same  $\varepsilon$  have the same value of  $\bar{n}$  and  $a$ , and such solutions are known as corresponding solutions.<sup>1,15-18</sup>

The function  $\varepsilon(A)^{16}$  was plotted for three or four given values of the total concentration  $B$ . From these curves pairs of values  $(B, A)$ , corresponding to selected  $\varepsilon$ , were obtained.

If  $A$  was plotted against  $B$  at a constant value of  $\varepsilon$ , a straight line was obtained, indicating that only mononuclear complexes

were present,  $\bar{n}$  being obtained from the slope and  $a$  from the intercept on the  $A$ -axis. Measurements for  $\lambda = 280$  nm illustrate this procedure (cf. Figs. 3, 4 and Table 3).

The stability constants were then calculated from the  $\bar{n}(a)$  values by several different methods (cf. Table 4 and Fig. 6).

CALCULATIONS BASED ON THE DATA  
IN THE LOW UV RANGE

(5)

*Curve fitting.* In order to determine values of  $\beta_1$  and  $\beta_2$ , the experimental curve in Fig. 5 was fitted to the following normalized curve:

where  $u = \beta_2^{\frac{1}{2}}a$  and  $R = \beta_1\beta_2^{-\frac{1}{2}}$ .<sup>10</sup> Functions  $\bar{n}(\log u)_R$  were drawn for different constant  $R$  values and compared with the experimental ones.

Table 5. Spectrophotometric data in the low UV range for 40 solutions. *A* and *B* are given for each solution, followed by  $\epsilon$ ,  $\epsilon_{\text{calc}}$  and  $\epsilon_{\text{calc}} - \epsilon$  for 7 wave lengths. The concentrations are expressed in M and the apparent absorptivities in  $M^{-1} \text{ cm}^{-1}$ . Missing data is indicated by -1.

0.004000	0.005000								
15.3°C	15.3°C	C. CCC	34.300	34.516	C.216	81.300	80.787	-0.513	
118.700	118.968	0.268	168.500	169.606	1.106	230.500	231.706	1.206	
304.300	304.348	0.048							
C. CCC	0.005000								
20.3°C	15.93C	-C.37C	43.700	43.307	-0.393	100.600	98.428	-2.172	
143.500	142.974	-0.526	201.730	200.827	-C.873	272.400	270.532	-1.868	
352.3CC	348.822	-3.178							
0.008000	0.005000								
24.680	24.736	C.056	52.150	52.316	0.166	116.900	116.075	-0.825	
165.100	166.758	1.058	230.800	231.474	0.674	307.500	308.259	0.759	
352.300	351.910	0.340							
0.010000	0.005000								
29.800	29.713	-0.097	61.600	61.522	-C.076	134.400	133.714	-0.686	
152.100	150.315	-1.685	264.400	261.500	-2.84C	347.000	344.925	-2.075	
435.3CC	423.456	-2.144							
0.012000	0.005000								
34.800	34.820	C.020	71.000	70.908	-0.052	151.100	151.332	0.232	
214.4CC	213.64C	-C.76C	252.400	251.094	-1.306	380.300	380.562	C.263	
473.290	473.720	C.52C							
0.016000	0.005000								
44.760	45.451	0.691	89.910	90.152	0.242	185.200	186.467	1.267	
255.5CC	255.591	-0.308	349.000	348.558	-0.444	447.1CC	448.9C3	1.8C4	
-1.000	-1.000	C.0							
0.020000	0.005000								
51.5CC	50.540	-1.360	113.600	109.923	-3.477	-1.CCC	-1.CCC	C.0	
-1.000	-1.000	C.0	-1.000	-1.000	0.0	-1.000	-1.000	0.0	
C. CCC	0.005000								
67.6CC	68.066	0.446	130.500	130.110	-0.390	257.700	256.055	-1.644	
350.2CC	348.631	-1.565	458.6CC	457.371	-1.229	572.600	574.722	2.122	
-1.000	-1.000	0.0							
C. CCC	0.005000								
76.720	75.76C	1.16C	148.800	150.619	1.819	288.300	290.385	2.086	
-1.000	-1.000	C.0	-1.CCC	-1.CCC	C.0	-1.000	-1.000	0.0	
C. CCC	0.005000								
92.900	91.801	-1.159	173.000	171.365	-1.635	-1.000	-1.000	0.0	
-1.000	-1.000	0.0	-1.000	-1.000	C.0	-1.000	-1.000	0.0	
0.034000	0.010000								
15.400	14.928	-C.562	34.210	33.63C	-L.57C	79.8CC	78.984	-0.816	
117.4CC	116.502	-1.298	165.400	166.382	0.982	227.4CC	227.674	-0.126	
255.610	255.718	C.11E							
0.036000	0.010000								
19.680	19.214	-0.466	42.250	41.954	-C.356	97.2CC	95.742	-1.278	
135.4CC	135.334	-0.266	194.610	196.112	1.512	265.000	264.694	-C.306	
341.210	342.148	C.44E							
0.008000	0.010000								
25.540	25.754	-0.106	50.400	50.484	0.084	112.400	112.520	0.12C	
160.5CC	160.565	1.485	222.000	225.346	3.346	298.400	300.746	2.346	
379.200	383.289	4.485							
C. CCC	0.010000								
32.1CC	33.282	-0.11E	68.400	68.097	-0.303	146.600	146.092	-C.508	
207.2CC	206.723	-0.477	263.400	262.361	-1.039	369.800	370.062	0.262	
455.6CC	461.875	2.275							
C. CCC	0.010000								
42.150	43.335	1.165	84.890	86.346	1.456	176.400	179.614	3.214	
-1.000	-1.000	C.0	-1.CCC	-1.CCC	C.0	-1.000	-1.000	0.0	
C. CCC	0.010000								
53.100	53.133	G.742	1C5.000	1C5.122	0.122	211.900	213.013	1.113	
292.7CC	292.849	1.149	389.100	390.6C7	1.7C7	495.6CC	498.31C	2.710	
-1.CCC	-1.CCC	0.0							
0.024000	0.010000								
64.700	64.709	0.709	122.800	124.329	1.525	244.300	246.225	1.925	
323.800	336.158	2.398	439.000	442.308	3.365	-1.CCC	-1.CCC	C.0	
0.028000	0.010000								
76.240	75.899	-0.341	144.100	143.821	-C.215	279.600	279.196	-C.404	
377.3CC	377.721	0.422	492.400	492.241	-0.159	-1.000	-1.000	C.0	
-1.CCC	-1.CCC	0.0							
0.032000	0.010000								
85.400	87.118	-1.171	165.010	163.701	-1.299	314.300	311.879	-2.42C	
420.500	418.412	-2.487	542.200	540.467	-1.713	-1.000	-1.000	0.0	
-1.000	-1.000	C.0							
0.036000	0.010000								
1CC.CCC	55.007	-3.553	183.600	183.721	C.121	345.400	344.235	-1.165	
458.CCC	458.268	C.265	-1.CCC	-1.CCC	C.0	-1.000	-1.000	0.0	
-1.000	-1.000	0.0							
C. CCC	0.020000								
18.100	17.985	-C.115	39.260	39.628	C.368	90.750	91.100	0.350	
133.800	133.030	-0.77C	166.9CC	167.53C	1.03C	253.400	254.542	1.142	
331.3CC	330.531	-0.469							
C. CCC	0.020000								
21.890	22.070	0.18C	44.0CC	47.332	3.312	106.000	106.363	0.363	
153.100	153.106	0.596	212.700	214.678	1.978	285.4CC	287.630	2.231	
342.4CC	342.343	1.743							
0.020000	0.020000								
26.640	26.299	-0.35G	55.010	55.20E	C.15E	122.700	121.656	-1.043	
174.4CC	174.235	-0.1565	241.100	241.052	-0.04E	321.3CC	319.971	-1.320	
406.5CC	405.154	-1.44E							

Table 5. Continued.

0.012000	0.020000										
31.079	30.638	-0.432	63.480	63.243	-0.237	138.700	136.971	-1.729			
155.500	154.641	-0.859	268.200	267.057	-1.143	353.200	351.584	-1.616			
442.000	440.554	-1.666									
0.016000	0.020000										
39.840	39.482	-0.158	80.050	79.750	-0.360	168.500	167.629	-0.671			
225.700	235.045	-0.655	320.800	317.977	-2.022	415.600	412.698	-2.901			
512.100	515.857	-2.233									
0.020000	0.020000										
46.550	46.145	0.555	96.570	96.770	0.200	197.600	198.279	0.679			
273.700	274.882	1.182	367.800	367.476	-0.324	471.400	471.114	-0.286			
575.100	575.173	0.073									
C.024000	0.020000										
55.440	56.973	-0.467	116.400	114.226	-2.174	-1.000	-1.000	0.0			
-1.000	-1.000	0.0	-1.CCC	-1.CCC	0.0	-1.000	-1.000	0.0			
-1.000	-1.000	0.0									
C.028000	0.020000										
66.370	66.115	0.745	132.100	132.051	-0.049	257.400	259.340	1.941			
352.100	352.775	0.675	462.200	462.358	0.156	581.500	580.377	-1.122			
694.900	695.864	0.964									
C.C32CCCC	C.C20000										
79.920	79.528	-0.392	151.700	150.182	-1.518	292.300	289.662	-2.638			
399.500	390.801	-4.199	-1.000	-1.CCC	0.0	-1.000	-1.000	0.0			
-1.CCC	-1.000	0.0									
0.026000	0.020000										
88.550	90.171	1.621	165.700	168.561	2.861	317.800	319.792	1.993			
425.900	428.198	2.298	550.990	551.999	1.100	680.500	680.366	-0.133			
-1.000	-1.CCC	0.0	C.C								
0.030000	0.020000										
12.296	12.100	0.086	28.540	28.684	C.344	46.850	46.240	0.390			
103.900	103.124	-0.276	147.330	148.839	1.539	203.500	205.698	2.150			
273.500	274.358	C.658									
0.006000	0.050000										
15.300	15.357	0.097	34.260	34.699	0.425	81.236	81.159	-0.071			
118.900	115.477	0.577	168.000	170.272	2.272	229.400	232.530	3.138			
301.700	305.303	3.603									
C.CC8000	0.050000										
16.640	16.521	-0.119	40.440	40.644	0.204	94.360	93.131	-1.229			
136.000	135.791	-0.209	151.990	151.516	0.516	258.800	258.994	0.195			
335.900	335.627	-0.273									
C.C12CCCC	C.C50000										
25.052	25.052	-0.578	52.600	52.905	0.305	117.600	117.216	-0.386			
164.600	164.598	-6.312	223.300	223.427	C.137	309.600	310.662	1.062			
393.300	394.573	1.273									
C.C14CCCC	C.C50000										
32.440	31.935	-0.505	65.930	65.627	-C.303	142.100	141.461	-0.639			
202.600	200.596	-2.094	277.500	274.600	-2.852	362.400	360.713	-1.687			
453.100	451.316	-1.684									
0.C20000	C.C50000										
39.290	39.141	-0.149	77.720	78.769	1.C45	162.700	165.834	3.134			
230.100	232.695	2.595	311.700	315.036	3.336	-1.CCC	-1.CCC	0.0			
0.024000	0.050000										
47.400	46.645	-0.645	94.380	92.294	-2.C65	191.900	190.305	-1.595			
216.700	216.568	-1.132	358.600	354.727	-4.073	459.400	456.161	-3.239			
555.100	556.564	-C.554									
0.C28000	0.050000										
54.760	54.422	-J.338	106.800	106.168	-0.632	215.100	214.845	-0.255			
295.600	296.159	1.200	394.200	393.688	-0.512	501.900	501.653	-0.247			
-1.000	-1.000	C.0									
0.C32000	0.050000										
62.520	62.450	-C.CTC	119.700	120.357	0.657	239.300	239.427	0.127			
324.900	327.574	2.675	430.CCC	431.927	1.927	-1.000	-1.000	0.0			
-1.000	-1.000	0.0									
C.C36CCCC	C.C50000										
69.260	70.704	1.324	132.500	134.829	2.329	-1.000	-1.000	0.0			
-1.000	-1.000	C.0	-1.CCC	-1.CCC	C.C	-1.000	-1.000	0.0			
-1.CCC	-1.000	0.0									

The best fit was obtained when  $R=1.5$  and  $\log u=1$  and  $\log a=1.83$  (cf. Fig. 5). This gave  $\beta_1=10.1 \text{ M}^{-1}$  and  $\beta_2=45.7 \text{ M}^{-2}$ .

#### "LETAGROP" CALCULATIONS

The experimental data from the low UV-range were also processed with the spectrophotometric version "SPEFO" of the "LETAGROP" program.<sup>22</sup>  $U$  is the error squares sum, defined as  $U=\sum(\varepsilon_{\text{calc}}-\varepsilon)^2$ .

The "best values" for  $\beta_1$  and  $\beta_2$  obtained were:

$$\beta_1=11.60 \pm 0.22 \text{ M}^{-1}$$

$$\beta_2=39.17 \pm 6.21 \text{ M}^{-2}$$

and  $U=490.2$  for 231 values for  $A < 0.04 \text{ M}$ . The errors given are  $\sigma$ , where  $\sigma$  is the standard deviation in  $\beta$ . The corresponding  $\varepsilon_1$  and  $\varepsilon_2$

values are given in Table 7 and Fig. 8. These  $\varepsilon_1$  and  $\varepsilon_2$  values were also confirmed by graphical methods. In Table 5 experimental and calculated data are presented.

#### CALCULATIONS BASED ON THE COMPLETE DATA SET

*Fronæus' method.*<sup>17,20</sup> The function used was

$$\bar{n} = \frac{d\phi_0/\phi_0}{da/a} = \frac{d \ln \phi_0}{d \ln a} \quad (6)$$

Integrating (6) between the limits 0 and  $a_i$ , gives

$$\ln \phi_0(a_i) = \int_0^{a_i} \bar{n} d \ln a \quad (7)$$

$\phi_0$  is determined from eqn. (7).

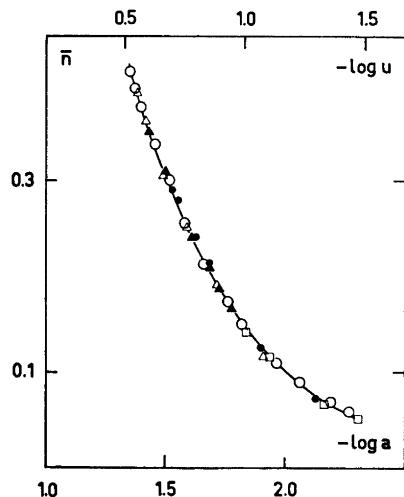


Fig. 5.  $\bar{n}$  as a function of  $\log a$  in the UV range.  $\triangle$  300 nm;  $\blacktriangle$  290 nm;  $\circ$  280 nm;  $\bullet$  270 nm;  $\square$  260 nm.

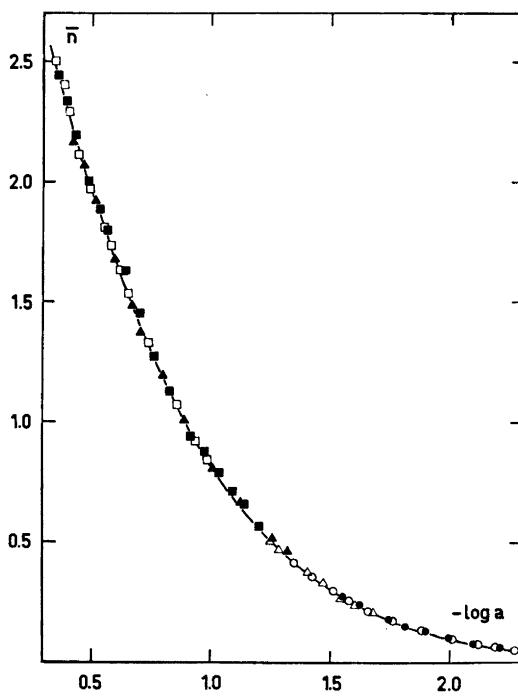


Fig. 6. The function  $\bar{n}(\log a)$  at different wave lengths.  $\circ$  280 nm;  $\bullet$  300 nm,  $\triangle$  350 nm;  $\blacktriangle$  360 nm;  $\square$  370 nm;  $\blacksquare$  380 nm.

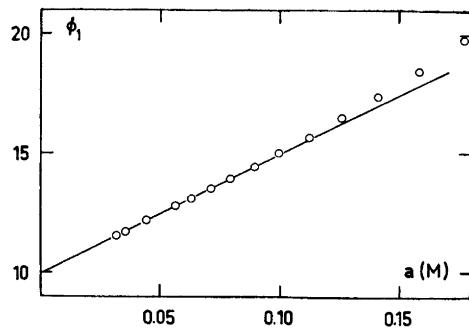


Fig. 7a.  $\phi_1$  as a function of  $a$ .

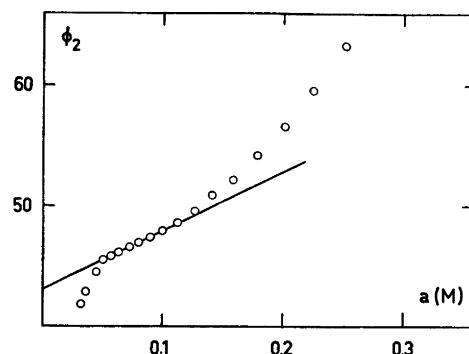


Fig. 7b.  $\phi_2$  as a function of  $a$ .

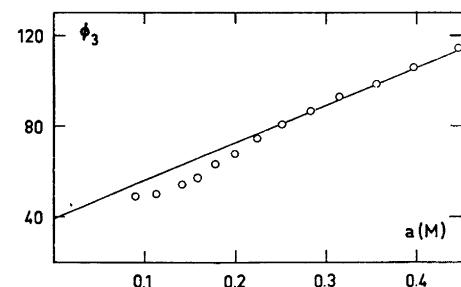


Fig. 7c.  $\phi_3$  as a function of  $a$ .

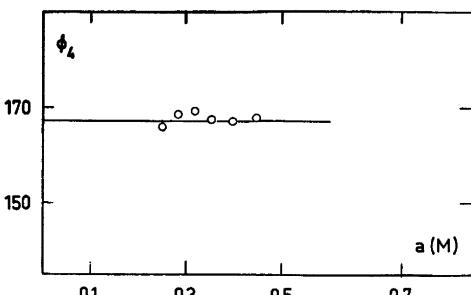


Fig. 7d.  $\phi_4$  as a function of  $a$ .

Table 6. The intercepts and the slopes of the different graphical functions for the calculation of the  $\beta$ -values.

	Function	Intercept	Slope	$\beta_1$ (M <sup>-1</sup> )	$\beta_2$ (M <sup>-2</sup> )	$\beta_3$ (M <sup>-3</sup> )	$\beta_4$ (M <sup>-4</sup> )
All data	$\phi_1$	10	50				
	$\phi_2$	43	49				
	$\phi_3$	39	167	10	43	39	167
	$\phi_4$	167	—				
Data in the range 350–385 nm	$\phi_1$	11.5	42.5				
	$\phi_2$	40.5	43				
	$\phi_3$	46	177	11.5	40.5	46	177
	$\phi_4$	177	—				
All data	F <sub>1</sub>	10	46				
	F <sub>2</sub>	36	75				
	F <sub>3</sub>	41.5	161	10	36	42	161
	F <sub>4</sub>	161	—				
All data	G <sub>2</sub>	43	38	11	38	43	—
	G <sub>3</sub>	43	11				

$$\phi_1 = \frac{\phi_0 - 1}{a} = \beta_1 + \beta_2 a \left( + \sum_{i=3}^N \beta_i a^{i-1} \right) \quad (8)$$

$$F_1 = \frac{\bar{n}}{(1-\bar{n})a} = \beta_1 + \beta_2 \frac{2-\bar{n}}{1-\bar{n}} a +$$

$$\sum_{i=3}^N \frac{i-\bar{n}}{1-\bar{n}} \beta_i a^{i-1} \quad (9)$$

At very low values of  $a$ , the first two complexes,  $BA$  and  $BA_2$ , predominate, and the function  $\phi_1(a)$  becomes a straight line of slope  $\beta_2$  and with intercept  $\beta_1$ . A value of any constant  $\beta_i$  may be obtained by using a generalization of eqn. (8).

The disadvantage of this method is that any errors in the values of  $\beta_1, \dots, \beta_{i-1}$  will accumulate in the value of  $\beta_i$ .

In Figs. 7a–d, the different functions  $\phi_i(a)$  have been plotted, and the slopes and intercepts of these four functions are listed in Table 6. The approximate values of the stability constants obtained were:

$$\begin{aligned} \beta_1 &= 10 \text{ M}^{-1} & \beta_3 &= 39 \text{ M}^{-3} \\ \beta_2 &= 43 \text{ M}^{-2} & \beta_4 &= 167 \text{ M}^{-4} \end{aligned}$$

The values of the constants based on the 350–385 nm data, only, were somewhat different. This can be attributed to the selection of the tail of the  $\bar{n}(\log a)$  curve. The values obtained were:

$$\begin{aligned} \beta_1 &= 11.5 \text{ M}^{-1} & \beta_3 &= 46 \text{ M}^{-3} \\ \beta_2 &= 40.5 \text{ M}^{-2} & \beta_4 &= 177 \text{ M}^{-4} \end{aligned}$$

(cf. Table 6).

Rossotti's method.<sup>21</sup> The function

may be derived from eqn. (3). The plot of  $\bar{n}(1-\bar{n})/a$  against  $(2-\bar{n}) a/(1-\bar{n})$  becomes a straight line of slope  $\beta_2$  and intercept  $\beta_1$ , as  $a \rightarrow 0$ .

A value of any constant  $\beta_i$  may be obtained by using a generalization of eqn. (9).

The slopes and the intercepts of the four functions  $F_i$  are given in Table 6. The following values of the stability constants were obtained:

$$\begin{aligned} \beta_1 &= 10 \text{ M}^{-1} & \beta_3 &= 42 \text{ M}^{-3} \\ \beta_2 &= 36 \text{ M}^{-2} & \beta_4 &= 161 \text{ M}^{-4} \end{aligned}$$

The reciprocal method of Rossotti.<sup>21</sup> The constants may also be obtained by extrapolation to  $a^{-1}=0$ , using the function

$$G_N = \frac{\bar{n}a^{-N}}{N-\bar{n}} = \beta_N + \beta_{N-1} \frac{N-1-\bar{n}}{N-\bar{n}} a^{-1} + \sum_{i=1}^{N-2} \frac{i-\bar{n}}{N-\bar{n}} \beta_i a^{i-N} \quad (10)$$

In this case no value can be obtained for  $\beta_4$  (cf. Table 6). The values obtained were:

$$\begin{aligned} \beta_1 &= 11 \text{ M}^{-1} & \beta_3 &= 43 \text{ M}^{-3} \\ \beta_2 &= 38 \text{ M}^{-2} & & \end{aligned}$$

Table 7. Molar absorptivities,  $\varepsilon_i$ , in  $M^{-1} \text{cm}^{-1}$  calculated with the "LETAGROP" program. The errors given are  $\sigma$ , where  $\sigma$  is the standard deviation in  $\varepsilon$ .

$\lambda \text{ nm}$	$\varepsilon_0$	$\varepsilon_1$	$\varepsilon_2$	$\varepsilon_3$	$\varepsilon_4$
260	210.9	2417 $\pm$ 10	1754 $\pm$ 199	—	—
265	150.6	2055 $\pm$ 7	2164 $\pm$ 111	—	—
270	105.4	1597 $\pm$ 7	2926 $\pm$ 100	—	—
275	70.3	1190 $\pm$ 5	3083 $\pm$ 67	—	—
280	45.6	847 $\pm$ 5	2913 $\pm$ 64	—	—
290	17.7	390 $\pm$ 4	2276 $\pm$ 53	—	—
300	6.6	195 $\pm$ 2	1434 $\pm$ 28	—	—
350	—	8.8 $\pm$ 0.7	171 $\pm$ 3	222 $\pm$ 18	513 $\pm$ 15
355	—	5.9 $\pm$ 0.6	157 $\pm$ 2	190 $\pm$ 16	503 $\pm$ 13
360	—	4.1 $\pm$ 0.6	145 $\pm$ 2	141 $\pm$ 12	533 $\pm$ 8
365	—	2.5 $\pm$ 0.4	129 $\pm$ 2	122 $\pm$ 13	534 $\pm$ 9
370	—	1.0 $\pm$ 0.4	114 $\pm$ 2	110 $\pm$ 11	536 $\pm$ 7
375	—	0.9 $\pm$ 0.4	98 $\pm$ 2	85 $\pm$ 12	549 $\pm$ 8
380	—	0.6 $\pm$ 0.3	84 $\pm$ 2	62 $\pm$ 12	550 $\pm$ 8
385	—	0.5 $\pm$ 0.3	71 $\pm$ 2	47 $\pm$ 13	530 $\pm$ 9

Table 8. Spectrophotometric data in the range 350–385 nm for 11 solutions and 8 wave lengths.  $A$  and  $B$  are given for each solution, followed by  $\varepsilon$ ,  $\varepsilon_{\text{calc}}$ , and  $\varepsilon_{\text{calc}} - \varepsilon$ . The concentrations are expressed in  $M$  and the apparent absorptivities in  $M^{-1} \text{cm}^{-1}$ .  $\beta_1 = 10.7 \text{ M}^{-1}$  and  $\beta_2 = 39.6 \text{ M}^{-2}$  were kept constant during the calculation.

0.019200	0.010000							
3.460	3.176	-0.024	2.550	2.556	0.004	2.050	2.085	0.035
1.450	1.668	0.018	1.25C	1.317	0.067	1.100	1.101	0.001
0.850	0.882	0.032	0.650	0.714	0.644			
C.0256CC	0.010000							
4.750	4.769	C.019	3.850	3.914	0.064	3.300	3.241	-0.059
2.600	2.647	0.047	2.100	2.150	C.05C	1.800	1.797	-0.003
1.400	1.448	0.048	1.100	1.169	0.069			
C.032CC0	0.010CCC							
6.750	6.551	-0.155	5.65C	5.453	-0.197	4.750	4.561	-0.189
3.850	3.776	-0.074	3.150	3.123	-0.027	2.650	2.610	-0.040
2.120	2.111	-0.039	1.700	1.702	0.002			
0.038400	0.010CCC							
8.000	8.053	-0.117	7.250	7.132	-0.112	6.050	6.014	-0.036
5.45C	5.028	-0.022	4.250	4.212	-0.038	3.600	3.520	-0.080
2.500	2.553	-C.C47	2.300	2.298	-0.002			
0.020400	0.020000							
3.400	3.057	0.017	2.440	2.454	C.014	2.CCC	2.000	0.000
1.600	1.557	-C.C03	1.300	1.257	-0.043	1.100	1.052	-0.048
0.850	0.842	-0.008	0.700	0.682	-0.018			
0.027200	0.020000							
4.540	4.600	0.060	3.690	3.768	0.078	3.040	3.116	0.076
2.540	2.541	C.CC1	2.040	2.059	0.019	1.700	1.721	0.021
1.400	1.386	-0.014	1.150	1.120	-0.03C			
0.034000	0.020000							
6.400	6.331	-C.069	5.300	5.263	-0.037	4.450	4.397	-0.053
3.400	3.357	-0.113	3.100	3.002	-0.098	2.550	2.509	-0.041
2.100	2.026	-0.072	1.750	1.635	-0.115			
C.4CECCC	0.020000							
8.230	8.218	-0.112	6.55C	6.506	-0.084	5.890	5.813	-0.077
4.890	4.855	-0.035	4.140	4.061	-0.075	3.440	3.394	-0.046
2.750	2.750	-0.040	2.240	2.215	-0.025			
0.020000	C.050CCC							
2.050	2.168	0.118	1.660	1.709	0.049	1.270	1.374	0.104
1.030	1.075	0.045	0.830	0.822	-0.008	0.640	0.687	0.047
C.54C	0.547	0.007	0.440	0.444	0.004			
0.040000	C.050000							
5.690	5.841	0.151	4.760	4.828	C.078	3.930	4.032	0.102
3.24C	3.322	0.082	2.650	2.731	0.081	2.160	2.283	0.123
1.700	1.644	C.C74	1.470	1.487	0.017			
0.040000	0.050000							
10.330	10.536	0.206	8.760	8.939	0.175	7.430	7.574	0.144
6.300	6.38C	C.C80	5.310	5.396	0.086	4.430	4.508	0.078
3.590	3.661	0.071	2.90C	2.946	0.046			

### THE "LETAGROP" CALCULATIONS

The values obtained by the different graphical methods were refined using the computer program "LETAGROP VRID".<sup>23</sup> The "best values" for  $\beta_1$ ,  $\beta_2$ ,  $\beta_3$ , and  $\beta_4$  obtained were:

$$\beta_1 = 10.69 \pm 0.25 \text{ M}^{-1} \quad \beta_3 = 58.53 \pm 8.37 \text{ M}^{-3}$$

$$\beta_2 = 39.60 \pm 2.10 \text{ M}^{-2} \quad \beta_4 = 163.3 \pm 9.66 \text{ M}^{-4}$$

and  $U = \sum(\bar{n}_{\text{calc}} - \bar{n})^2 = 0.13946$ . 246  $\bar{n}(\log a)$ -values were used. The errors given are  $\sigma$ , where  $\sigma$  is the standard deviation in  $\beta$ .

The original experimental data in the range 350–385 nm were also processed with the spectrophotometric version "SPEFO" of the "LETAGROP" program.<sup>22</sup>  $U$  is the error squares sum, defined as  $U = \sum(\varepsilon_{\text{calc}} - \varepsilon)^2$ .

It appeared to be somewhat hazardous to determine the stability constants with the "SPEFO" program, owing partly to the large number of parameters and partly to the low molar absorptivity of CuCl<sup>+</sup>.

*Table 9.* Spectrophotometric data in the range 350–385 nm for 74 solutions.  $A$  and  $B$  are given for each solution, followed by  $\varepsilon$ ,  $\varepsilon_{\text{calc}}$ , and  $\varepsilon_{\text{calc}} - \varepsilon$  for 8 wave lengths. The concentrations are expressed in M, the apparent absorptivities in M<sup>-1</sup> cm<sup>-1</sup>.  $\beta_1 = 10.7$  M<sup>-1</sup>,  $\beta_2 = 39.6$  M<sup>-2</sup>,  $\beta_3 = 59$  M<sup>-3</sup> and  $\beta_4 = 163$  M<sup>-4</sup> were kept constant during the calculation. Missing data is indicated by -1.

0.240003	0.005000								
-1.000	-1.000	0.0	-1.000	-1.000	0.0	99.000	98.885	-0.115	0.3
-1.003	-1.000	0.0	85.000	84.918	-0.082	-1.000	-1.000	-1.000	
-1.000	-1.323	0.0	-1.000	-1.000	0.0				
0.270000	0.005000								
-1.000	-1.000	0.0	-1.000	-1.000	0.0	115.800	115.122	-0.678	
108.000	107.238	-3.792	101.000	100.337	-0.663	93.800	93.281	-0.519	
86.400	85.727	-0.673	-1.000	-1.000	0.0				
0.285000	0.005000								
-1.000	-1.000	0.0	-1.000	-1.000	0.0	123.300	123.244	-0.356	
115.300	115.138	-0.162	108.100	108.140	0.040	101.100	100.949	-0.151	
93.300	93.117	-0.153	84.300	84.684	0.184				
0.300000	0.005000								
-1.000	-1.323	0.0	-1.000	-1.000	0.0	131.400	131.338	-0.062	
123.400	123.068	-0.332	116.300	115.970	-0.330	109.400	108.679	-0.721	
101.000	100.657	-0.343	91.800	91.791	-0.009				
0.315000	0.005000								
-1.000	-1.000	0.0	-1.000	-1.000	0.0	138.700	139.383	0.683	
130.300	130.976	0.676	123.100	123.801	0.701	115.700	116.443	0.743	
107.100	108.229	1.129	98.300	98.975	0.675				
0.330000	0.005000								
-1.000	-1.000	0.0	-1.000	-1.000	0.0	146.400	147.360	0.960	
137.800	138.841	1.041	130.400	131.611	1.211	123.400	124.216	0.816	
114.600	115.635	1.035	105.300	106.209	1.209				
0.345000	0.005000								
-1.000	-1.000	0.0	-1.000	-1.000	0.0	154.600	155.253	0.653	
146.603	146.645	0.045	139.000	139.379	0.379	132.000	131.976	-0.224	
123.200	123.452	0.252	113.200	113.468	0.268				
0.360000	0.005000								
-1.000	-1.000	0.0	-1.000	-1.000	0.0	161.800	163.047	1.247	
153.700	154.370	0.670	146.100	147.087	0.987	139.300	139.721	0.401	
130.300	131.056	0.756	119.930	120.729	0.829				
0.375000	0.005000								
-1.000	-1.000	0.0	-1.000	-1.000	0.0	171.300	170.727	-0.573	
163.200	162.002	-1.198	-1.000	-1.000	0.0	-1.000	-1.000	0.0	
139.600	138.627	-1.173	128.800	127.972	-0.828				
0.390000	0.005000								
-1.000	-1.323	0.0	-1.000	-1.000	0.0	178.500	178.283	-0.217	
169.800	169.527	-0.273	162.610	162.258	-0.342	155.600	154.973	-0.627	
146.200	146.147	-0.053	135.600	135.178	-0.422				
0.405000	0.005000								
-1.000	-1.000	0.0	-1.000	-1.000	0.0	185.300	185.706	0.406	
176.500	176.935	0.435	169.700	169.693	-0.007	162.400	162.488	0.088	
153.100	153.599	0.499	142.300	142.330	0.030				
0.076800	0.010000								
24.000	23.761	-0.239	21.233	20.948	-0.252	18.400	18.561	0.161	
16.000	16.182	0.182	-1.000	-1.000	0.0	-1.000	-1.000	0.0	
-1.000	-1.000	0.0	-1.000	-1.000	0.0				
0.089600	0.010000								
22.600	22.641	0.041	26.200	26.293	0.093	23.100	23.395	0.295	
20.203	20.534	0.234	17.600	17.873	0.273	-1.000	-1.000	0.0	
-1.000	-1.000	0.0	-1.000	-1.000	0.0				
0.102400	0.010000								
36.000	35.816	-0.184	32.100	31.926	-0.174	28.400	28.515	0.115	
25.000	25.175	0.175	22.000	22.081	0.081	19.000	19.339	0.339	
16.400	16.682	0.282	-1.000	-1.000	0.0				
0.115200	0.010000								
41.803	42.235	0.435	37.300	37.804	0.504	33.600	33.883	0.283	
29.700	30.375	0.375	26.290	26.561	0.361	22.900	23.364	0.464	
19.800	20.256	0.456	16.900	17.447	0.567				
0.128000	0.010000								
48.600	48.600	0.060	43.750	43.890	0.140	39.600	39.472	-0.128	
35.300	35.211	-0.089	31.400	31.295	-0.105	27.600	27.651	0.051	
24.000	24.097	0.097	20.700	20.853	0.153				
0.140800	0.010000								
55.400	55.659	0.259	50.300	50.155	0.155	45.300	45.258	-0.042	
40.600	40.561	-0.039	36.200	36.265	0.065	32.200	32.191	-0.009	
28.203	28.200	-0.000	24.300	24.517	0.217				
0.153600	0.010000								
62.500	62.602	0.102	56.250	56.573	0.323	50.800	51.220	0.420	
45.800	46.108	0.308	41.400	41.454	0.054	36.800	36.972	0.172	
32.400	32.555	0.155	28.200	28.435	0.235				
0.166400	0.010000								
69.800	69.665	-0.135	63.150	63.121	-0.029	57.400	57.340	-0.060	
51.500	51.515	-0.085	47.200	46.847	-0.353	42.200	41.979	-0.221	
37.400	37.153	-0.247	32.830	32.599	-0.201				
0.179200	0.010000								
77.000	76.824	-0.176	70.000	69.777	-0.223	63.800	63.599	-0.201	
57.800	57.725	-0.075	52.800	52.428	-0.372	47.400	47.201	-0.199	
42.200	41.983	-0.217	37.230	37.000	-0.203				
0.192000	0.010000								
83.603	84.358	0.458	76.000	76.522	0.522	69.500	69.980	0.483	
63.300	63.761	0.461	57.700	58.181	0.481	52.200	52.621	0.421	
46.800	47.032	0.232	41.300	41.625	0.325				
0.204600	0.010000								
90.600	91.347	0.547	83.050	83.336	0.286	75.800	76.466	0.666	
69.600	69.948	0.328	63.620	64.089	0.489	57.800	58.226	0.426	
52.000	52.284	0.264	46.000	46.461	0.461				
0.217603	0.010000								
97.800	98.672	0.872	89.700	90.203	0.503	82.500	83.040	0.540	
75.800	76.207	0.407	69.700	70.135	0.435	-1.000	-1.000	0.v	
57.500	57.724	0.224	51.300	51.494	0.194				

Table 9. Continued.

0.230400	0.010003	-1.000	0.0	-1.000	-1.000	3.0	88.900	89.487	0.787	
81.800	82.583	0.783	0.534	75.500	74.301	0.601	-1.000	-1.000	0.3	
62.800	63.536	0.534	0.534	56.100	56.706	0.606				
0.240000	0.010000	-1.000	0.0	-1.000	-1.000	3.0	93.600	94.709	-0.891	
87.800	87.418	-0.382	0.301	81.600	80.995	-0.005	74.500	74.446	-0.354	
67.900	67.646	-0.254	0.402	60.700	60.724	0.024				
0.243200	0.010000	-1.000	0.0	-1.000	-1.000	0.0	96.000	96.389	0.389	
88.000	89.339	1.039	0.200	82.200	82.571	0.371	75.600	75.970	0.370	
68.800	69.101	0.301	0.301	61.600	62.082	0.482				
0.250000	0.010000	-1.000	0.0	-1.000	-1.000	0.0	102.400	103.132	0.732	
92.200	95.559	0.359	0.400	88.400	88.927	0.527	-1.000	-1.000	0.0	
74.400	75.003	0.402	0.402	67.300	67.604	0.304				
0.268800	0.010000	-1.000	0.0	-1.000	-1.000	0.0	-1.000	-1.000	0.3	
-1.000	-1.000	0.0	0.0	-1.000	-1.000	0.0	87.900	88.401	0.501	
80.600	81.021	0.421	0.421	73.100	73.254	0.154				
0.270000	0.010303	-1.000	0.0	-1.003	-1.000	0.0	111.100	110.535	-0.565	
103.100	102.744	-0.356	0.510	96.100	95.957	-0.143	89.500	88.993	-0.507	
82.100	81.590	-0.510	0.200	74.200	73.789	0.410				
0.285000	0.011303	-1.000	0.0	-1.000	-1.000	0.0	119.200	118.479	-0.721	
111.800	111.335	-0.317	0.220	103.600	103.356	-0.044	96.600	96.640	-0.160	
89.000	88.783	-0.220	0.600	80.600	80.361	-0.039				
0.300000	0.010000	-1.000	0.0	-1.000	-1.000	0.0	127.000	126.614	-0.586	
119.203	118.240	-0.960	0.800	111.600	111.201	-0.599	104.600	103.987	-0.633	
96.400	96.375	-0.325	0.600	87.600	87.453	-0.147				
0.315000	0.010000	-1.000	0.0	-1.000	-1.000	0.0	133.600	134.319	0.719	
125.200	125.395	0.795	0.0	-1.000	-1.000	0.0	112.100	111.547	-0.553	
-1.000	-1.000	0.0	0.0	95.100	94.439	-0.661				
0.330000	0.010000	-1.003	0.0	-1.000	-1.000	0.0	142.500	142.175	-0.325	
133.800	133.226	0.026	0.200	127.600	126.329	-0.471	118.400	119.155	0.755	
110.900	110.800	-0.026	0.100	101.600	101.494	-0.306				
0.345000	0.010000	-1.000	0.0	-1.000	-1.000	0.0	158.200	157.674	-0.526	
148.800	149.043	0.243	0.243	141.600	141.770	-0.030	134.600	134.369	-0.031	
126.500	125.805	-0.695	0.695	116.200	115.714	-0.486				
0.375000	0.010000	-1.000	0.0	-1.000	-1.000	0.0	164.900	165.288	0.388	
156.703	156.595	-0.105	0.490	149.200	149.311	0.111	142.100	141.934	-0.166	
133.300	132.958	-0.042	0.200	122.700	122.634	0.134				
0.390000	0.010000	-1.000	0.0	-1.000	-1.000	0.0	173.100	172.795	-0.305	
162.600	164.060	1.460	0.200	157.000	156.778	-0.222	150.500	149.447	-1.053	
141.400	140.678	-0.722	0.600	130.600	129.937	-0.663				
0.081600	0.020000	23.203	23.174	-0.026	20.500	20.416	-0.084	17.900	18.081	0.181
-1.000	-1.300	0.0	0.0	-1.000	-1.000	0.0	-1.000	-1.000	0.0	
0.095200	0.020000	29.103	28.992	-0.108	25.800	25.701	-0.099	22.700	22.859	0.159
19.800	20.050	0.250	0.250	17.300	17.437	0.137	-1.000	-1.000	0.0	
-1.000	-1.000	0.0	0.0	-1.000	-1.000	0.0				
0.108000	0.020000	31.400	31.121	-0.279	31.500	31.291	-0.293	27.800	27.937	0.137
24.500	24.649	0.149	0.149	21.600	21.602	0.002	18.700	18.912	0.212	
16.200	16.304	0.104	0.104	-1.000	-1.000	0.0				
0.122600	0.020000	41.200	41.513	0.313	36.930	37.141	0.241	33.200	33.277	0.077
29.400	29.520	0.120	0.200	26.000	26.052	0.352	22.800	22.904	0.104	
19.703	19.846	0.146	0.146	16.800	17.086	0.286				
0.136000	0.022003	47.600	48.126	0.526	42.800	43.215	0.415	38.700	38.851	0.151
34.500	34.638	0.138	0.138	30.600	30.765	0.165	27.000	27.170	0.170	
23.500	23.564	0.164	0.164	20.200	20.467	0.267				
0.149000	0.020000	54.929	-0.171	49.800	49.481	-0.319	45.000	44.634	-0.366	
40.400	39.982	-0.418	0.400	35.725	36.000	-0.275	31.900	31.697	-0.233	
28.000	27.751	-0.249	0.200	24.303	24.115	-0.185				
0.163200	0.020000	62.203	61.891	-0.309	56.200	55.914	-0.286	51.000	50.607	-0.393
46.000	45.536	-0.464	0.400	41.200	40.917	-0.283	36.800	36.475	-0.325	
32.500	32.101	-0.399	0.346	28.300	28.325	-0.275				
0.176800	0.020000	69.200	68.985	-0.215	62.600	62.489	-0.111	56.900	56.748	-0.152
51.800	51.280	-0.520	0.400	46.800	46.323	-0.477	41.900	41.491	-0.409	
37.300	36.703	-0.597	0.300	32.600	32.191	-0.409				
0.192400	0.020000	76.100	0.188	69.200	69.185	-0.015	63.000	63.041	0.041	
57.200	57.998	-0.002	0.100	51.700	51.928	0.220	46.500	46.731	0.231	
41.200	41.548	0.346	0.346	36.700	36.602	-0.098				
0.204000	0.020000	84.600	83.479	-0.521	76.400	75.981	-0.419	69.900	69.466	-0.434
63.700	63.275	-0.425	0.000	-1.000	-1.000	0.0	52.400	52.182	-0.218	
-1.000	-1.000	0.0	0.0	41.248	41.248	0.552				
0.217600	0.020000	90.900	90.835	-0.065	83.200	82.857	-0.343	76.000	76.008	0.008
69.600	69.491	-0.109	0.100	63.600	63.670	0.070	57.700	57.827	0.127	
51.900	51.939	0.309	0.309	46.400	46.115	-0.285				
0.231200	0.020000	98.500	98.237	-0.263	89.800	89.794	-0.006	82.500	82.648	0.148
75.800	75.832	0.332	0.332	70.000	69.773	-0.227	65.700	65.650	-0.050	
57.400	57.396	-0.004	0.004	51.600	51.190	-0.410				

Table 9. Continued.

0.244800	0.020000							
-1.000	-1.000	0.0	-1.000	-1.000	0.0	93.500	89.373	-1.133
83.400	82.279	-1.121	76.600	76.006	-0.594	70.300	69.634	-0.666
63.900	63.066	-0.834	57.100	56.455	-0.645			
0.258400	0.020000							
-1.000	-1.000	0.0	-1.000	-1.000	0.0	97.000	96.158	-0.842
89.500	88.816	-0.684	82.900	82.354	-0.546	76.103	75.760	-0.340
69.500	68.900	-0.600	62.400	61.695	-0.505			
0.272300	0.020000							
-1.000	-1.000	0.0	-1.000	-1.000	0.0	-1.000	-1.000	0.0
96.500	95.425	-1.075	89.400	88.797	-0.603	82.500	82.011	-0.469
75.400	74.811	-0.519	67.800	67.490	-0.310			
0.285900	0.020003							
-1.000	-1.000	0.0	-1.000	-1.000	0.0	-1.000	-1.000	0.0
-1.000	-1.000	0.0	95.600	95.318	-0.282	88.400	88.369	-0.031
80.800	80.989	0.189	73.400	73.224	-0.176			
0.360000	0.020030							
169.000	167.532	-1.468	156.800	155.561	-1.239	146.300	147.408	1.108
139.400	138.888	-0.512	132.000	131.657	-0.343	124.100	124.262	0.162
115.600	115.881	0.281	105.700	106.252	0.552			
0.390000	0.020000							
181.700	182.766	1.066	169.300	170.199	0.899	162.200	162.223	0.023
152.600	153.552	0.953	145.400	146.271	0.871	137.600	138.881	1.281
128.700	130.448	1.548	118.603	119.957	1.357			
0.420000	0.020000							
-1.000	-1.000	0.0	-1.000	-1.000	0.0	177.700	176.491	-1.009
168.900	167.949	-0.960	161.300	160.667	-0.633	154.500	153.367	-1.133
144.900	144.557	-0.343	133.700	133.654	-0.046			
0.140000	0.050000							
38.200	38.183	-0.017	34.000	34.091	0.091	33.600	30.490	-0.110
27.000	26.974	-0.026	23.700	23.721	0.021	20.700	20.809	0.109
17.800	17.983	0.183	-1.000	-1.000	0.0			
0.120000	0.050000							
30.600	30.431	-0.169	27.200	27.012	-0.188	23.700	24.047	0.347
20.800	21.123	0.323	18.000	18.406	0.406	-1.000	-1.000	0.0
-1.000	-1.000	0.0	-1.000	-1.000	0.0			
0.100000	0.050003							
23.400	23.199	-0.210	20.433	20.431	0.031	17.800	18.094	0.294
-1.000	-1.000	0.0	-1.000	-1.000	0.0	-1.000	-1.000	0.0
-1.000	-1.000	0.0	-1.000	-1.000	0.0			
0.160000	0.050000							
46.200	46.362	0.162	41.900	41.592	-2.308	37.600	37.358	-0.242
33.700	33.265	-0.435	29.400	29.497	0.097	26.200	26.018	-0.182
22.500	22.633	0.130	19.300	19.549	0.249			
0.180000	0.050000							
55.100	54.894	-0.206	49.600	49.449	-0.151	45.003	44.604	-0.396
40.400	39.955	-0.445	36.000	35.700	-0.300	31.900	31.673	-0.227
28.000	27.730	-0.270	24.200	24.096	-0.104			
0.200000	0.050000							
64.400	62.710	-0.380	57.900	57.607	-0.293	52.800	52.185	-0.115
47.100	47.006	-0.091	42.600	42.299	-0.301	38.000	37.754	-0.246
33.500	33.271	-0.229	29.200	29.082	-0.118			
0.220000	0.050000							
73.200	72.786	-0.414	66.400	66.620	-0.381	60.400	60.361	-0.339
54.400	54.392	-0.008	49.800	49.266	-0.534	44.400	44.237	-0.163
39.400	39.238	-0.162	34.700	34.495	-0.205			
0.240000	0.050000							
81.200	82.045	0.845	73.603	74.643	1.043	67.800	68.199	0.399
61.600	62.073	0.473	56.600	56.569	-0.031	51.200	51.099	-0.101
45.700	45.610	-0.090	40.600	40.320	-0.280			
0.260000	0.050003							
92.400	91.600	-0.547	83.803	83.435	-0.365	77.000	76.560	-0.440
70.200	70.018	-0.182	64.400	64.176	-0.224	58.200	58.308	0.108
52.400	52.361	-0.039	46.700	46.533	-0.167			
0.280000	0.050000							
101.000	100.968	-0.032	92.300	92.358	0.058	84.700	85.112	0.412
77.600	78.192	0.592	71.500	72.051	0.551	65.400	65.833	0.433
59.200	59.462	0.262	53.100	53.105	0.005			
0.300000	0.050000							
-1.000	-1.000	0.0	-1.000	-1.000	0.0	93.900	93.819	-0.081
86.200	86.563	0.360	79.800	80.161	0.361	73.000	73.640	0.640
66.400	66.878	0.478	59.500	60.007	0.507			
0.320000	0.050000							
-1.000	-1.000	0.0	-1.000	-1.000	0.0	-1.000	-1.000	0.0
96.100	95.519	0.389	88.300	88.468	0.168	81.400	81.692	0.292
74.300	74.157	0.275	66.700	67.204	0.504			
0.340000	0.050000							
-1.000	-1.000	0.0	-1.000	-1.000	0.0	-1.000	-1.000	0.0
-1.000	-1.000	0.0	105.000	105.533	0.533	97.700	98.383	0.683
-1.000	-1.000	0.0	-1.000	-1.000	0.0			
0.420000	0.050000							
-1.000	-1.000	0.0	-1.000	-1.000	0.0	147.600	147.488	-0.112
138.700	138.967	0.267	131.900	131.736	-0.164	124.400	124.341	-0.059
116.100	115.957	-0.143	106.700	106.325	-0.375			
0.440000	0.050003							
-1.000	-1.000	0.0	-1.000	-1.000	0.0	155.900	156.392	0.492
147.600	147.773	0.173	141.600	140.503	-1.097	134.000	133.101	-0.899
125.500	124.556	-0.942	115.500	114.523	-0.977			
0.460000	0.050000							
-1.000	-1.000	0.0	-1.000	-1.000	0.0	165.400	165.217	-0.193
157.500	156.525	-0.975	141.000	140.000	0.3	141.700	141.862	0.162
132.900	133.187	0.287	123.200	122.767	-0.433			
0.480000	0.050000							
-1.000	-1.000	0.0	-1.000	-1.000	0.0	-1.000	-1.000	0.0
164.800	165.233	0.400	157.100	157.920	0.820	149.600	150.598	0.998
-1.000	-1.000	0.0	130.400	131.027	0.627			
0.500000	0.050000							
-1.000	-1.000	0.0	-1.000	-1.000	0.0	183.400	182.544	-0.856
174.900	173.778	-1.122	166.900	166.522	-0.378	159.000	159.281	0.281
150.400	150.417	-0.283	-1.000	-1.000	0.0			

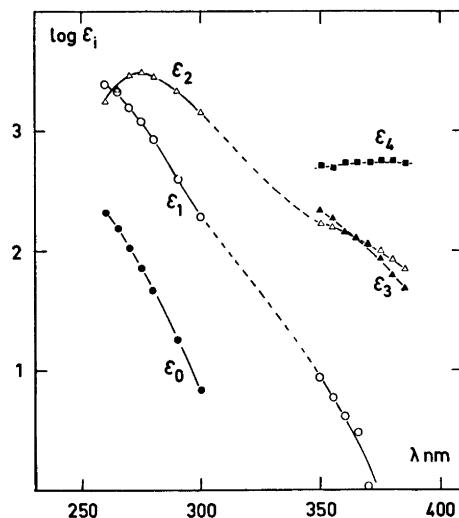


Fig. 8. The  $\log \varepsilon_i$  values obtained from the "LETAGROP" program.  $\varepsilon_0$  measured directly.

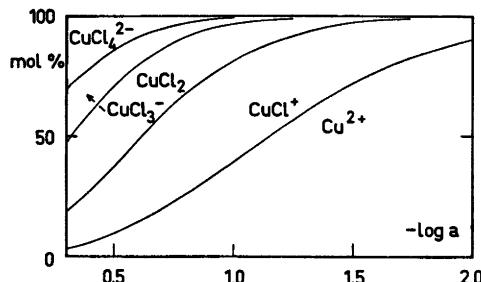


Fig. 9. The distribution of copper(II) chloride complexes as a function of  $\log a$  for  $B = 0.010$  M.

The values of the stability constants were determined from the  $\bar{n}(\log a)$ -curve by means of the program "LETAGROP VRID".<sup>23</sup> The values of the constants were then inserted in the "SPEFO" version to calculate the  $\varepsilon_i$  values. The program requires that  $\varepsilon_1$  is determined. By inserting the values of  $\beta_1$  and  $\beta_2$  and processing the experimental data obtained at low chloride ion concentration ( $A < 0.06$  M), it was possible to obtain approximate values for  $\varepsilon_1$  in the range 350–385 nm.  $\varepsilon_1$  was then fixed, and  $\varepsilon_2$ ,  $\varepsilon_3$ , and  $\varepsilon_4$  were calculated (cf. Table 7). All the  $\varepsilon_i$  values were then fixed, and the stability constants were calculated with the "SPEFO" program in the range 370–385 nm (cf. Table 10). The values for  $\beta_1$ ,  $\beta_2$ ,  $\beta_3$  and  $\beta_4$  obtained were:

$$\begin{array}{ll} \beta_1 = 10.43 \pm 0.42 \text{ M}^{-1} & \beta_3 = 56.54 \pm 2.13 \text{ M}^{-3} \\ \beta_2 = 39.13 \pm 0.92 \text{ M}^{-2} & \beta_4 = 160.0 \pm 4.86 \text{ M}^{-4} \end{array}$$

and  $U = 67.55$  for 274 values for  $A \leq 0.5$  M. The errors given are  $\sigma$ , where  $\sigma$  is the standard deviation. The errors are probably underestimated since it was necessary to hold the  $\varepsilon_i$  values constant during the calculation. The experimental data, which were processed with the "SPEFO" program are presented in Tables 8 and 9.

## RESULTS AND DISCUSSION

The refinement of the constants thus gave the following values which are regarded as being the "best values":

$$\begin{array}{ll} \log \beta_1 = 1.03 \pm 0.03 & \log \beta_3 = 1.77 \pm 0.19 \\ \log \beta_2 = 1.60 \pm 0.07 & \log \beta_4 = 2.21 \pm 0.08 \end{array}$$

Table 10. Survey of the values for the stability constants,  $\beta$ , obtained by the different methods of calculation.

Calculation method	$\beta_1$ (M <sup>-1</sup> )	$\beta_2$ (M <sup>-2</sup> )	$\beta_3$ (M <sup>-3</sup> )	$\beta_4$ (M <sup>-4</sup> )
Curve fitting	10.1	45.7	—	—
LETAGROP "SPEFO"	$11.6 \pm 0.2$	$39.6 \pm 6.2$	—	—
Fronæus' method	10	43	39	167
	11.5	41	46	177
Rosotti's method	10	36	42	161
Rosotti's reciprocal method	11	38	43	—
LETAGROP VRID	$10.7 \pm 0.3$	$39.6 \pm 2.1$	$59 \pm 8$	$163 \pm 10$
LETAGROP "SPEFO"	$10.4 \pm 0.4$	$39.1 \pm 0.9$	$57 \pm 2$	$160 \pm 5$

These were obtained from the "LETAGROP VRID" calculations based on the  $\bar{n}(\log \alpha)$  data (cf. Table 10).

The errors given are  $3\sigma$ , where  $\sigma$  is the standard deviation in  $\log \beta$ . The third complex shows a greater error than the others. In Fig. 9 a distribution curve has been drawn from the above values of the constants. The narrow band of existence for the third complex shows that it only exists in comparatively small quantities.

The recorded spectra show absorption bands at 250, 375, and 800–900 nm. According to Andreev *et al.*<sup>6</sup> and Eswein *et al.*<sup>12</sup> the absorption band at 250 nm indicates the formation of the lower complexes at these low concentrations. The absorption bands at 375 nm and 800–900 nm should indicate  $\text{CuCl}_4^{2-}$ . The fourth complex also absorbs in the low UV range.

The different  $\epsilon_i$  values have been calculated with the "LETAGROP" program (cf. Fig. 8). The maximum for  $\text{CuCl}_2$  is at 275 nm, which is in accordance with Andreev *et al.* The molar absorptivity,  $\epsilon_4$ , for  $\text{CuCl}_4^{2-}$  at 375 nm is about  $530 \text{ M}^{-1} \text{ cm}^{-1}$  which agrees with the findings of Eswein *et al.*<sup>12</sup> The value of  $\epsilon_2$  (approximately  $100 \text{ M}^{-1} \text{ cm}^{-1}$ ) is also similar to the value calculated by Eswein *et al.*

The coordination of the  $\text{Cu}^{2+}$  ion in aqueous solution is tetragonally distorted octahedral, and of  $\text{CuCl}_4^{2-}$  probably distorted tetrahedral. As can be seen from Fig. 8, the absorption maximum at 375 nm is characteristic for the  $\text{CuCl}_4^{2-}$  complex, only. Thus there is evidence that this complex has a different symmetry from the lower ones. However, it should be borne in mind that, although only  $\text{CuCl}_4^{2-}$  has an absorption maximum at 375 nm, the absorptivities of the other complexes cannot be neglected.

It is obvious that the stepwise stability constants in sulphuric acid solution are greater by a factor of 10 than those obtained in perchlorate medium. Without a knowledge of the entropies and heats of formation of the complexes in the perchlorate and sulphuric acid media, it is not worthwhile to speculate on the reason for this.

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