The Crystal Structure of Pd₆P

YVONNE ANDERSSON, VALAPHA KAEWCHANSILP,* MARIA del ROSARIO CASTELEIRO SOTO** and STIG RUNDOVIST

Institute of Chemistry, University of Uppsala, Box 531, S-751 21 Uppsala, Sweden

The crystal structure of Pd_sP has been investigated by powder diffraction methods. The unit cell is monoclinic of dimensions a=2.837 Å, b=9.441 Å, c=7.695 Å, $\beta=90.20^\circ$ and contains 12 palladium and 2 phosphorus atoms. The structure is closely related to the Re_3B -type structure.

The Pd-P equilibrium diagram has been throroughly investigated by Gullman. 1 He found two very palladium-rich intermediate phases, Pd.P and Pd.P, both being formed peritectically at 799°C and 792°C, respectively. Gullman was unable to prepare these compounds in single crystal form, and the powder diffraction patterns were too complex for indexing. Since Pd.P and Pd.P represent the most metal-rich intermediate phases known in any transition metal-phosphorus system, it would seem to be of great interest to study the properties of these phases more closely. We decided to start an X-ray crystallographic examination of the two phosphides, and in the present paper we give an account of our work on Pd,P.

EXPERIMENTAL DETAILS

Preparation and X-ray diffraction work. We prepared Pd-P samples by the silica tube technique in exactly the same manner as described by Gullman, using red phosphorus of purity higher than 99 % and palladium powder (claimed purity 99.9 %) from Johnson, Matthey & Co, Ltd., London, as starting materials. As mentioned by Gullman it is difficult to attain equilibrium conditions in palladium-rich alloys,

and in spite of long and careful heat treatments we could not obtain Pd, P samples completely free from palladium or other Pd-P phases. The alloys were very ductile, and the powders used for X-ray diffraction work were prepared by filing with fine-toothed files and sieving. To remove the effects of cold work the filings were annealed at 650 °C for one hour. The annealing time and temperature was found to be critical, since the powders sintered at higher temperatures and longer annealing times, and sharp diffraction lines were not obtained for powders annealed at lower temperatures. Attempts to obtain Pd, P crystals suitable for X-ray work were unsuccessful. The fragments extracted from the alloys invariably consisted of conglomerates of several crystallites, which were mechanically deformed and exhibited stacking faults and twinning. Annealing treatments of such fragments produced only minor improvements in the crystalline perfection.

Powder diffraction patterns were recorded in Hägg-Guinier-type cameras using $\operatorname{Cr} K\alpha_1$ or $\operatorname{Cu} K\alpha_1$ radiation and silicon (a=5.43054~Å) as internal calibration standard. Unit cell dimensions were refined by the least squares method using the local program CELNE ² on an

IBM 1800 computer.

Powder diffraction intensities were measured by two methods. In the first method the intensities were obtained by direct summation on the read-out scaler, using a Philips powder diffractometer PW 1050 with CuK radiation and a lithium fluoride monochromator between the sample and the proportional counter. In the second method Hägg-Guinier photographs, recorded in a Philips XDC 700 camera with strictly monochromatized $CrK\alpha_1$ radiation, were measured by means of an automatic drum densitometer (SAAB model 2) connected to an IBM 1800 computer in the same manner as described by Malmros and Werner.³ Evaluation of the integrated intensities was made using the program PILT * in a locally modified version.* This version included the correction factor $\{1 - \exp{-[\mu_t h/\cos{(2\theta - \phi)}]}\}, \quad (\mu_f = \text{absorption})$ $\{1-\exp{-[\mu_t h/\cos{(2\theta-\phi)}]}\}, \quad (\mu_t=\text{absorption coefficient for the film emulsion, } h=\text{thickness of}$ emulsion, θ and ϕ as in Ref. 3) allowing for the

^{*} On leave from the Department of Chemistry, Chulalongkorn University, Bangkok, Thailand.

^{**} On leave from the Department of Physics, Universidad de Oriente, Santiago de Cuba.

variation of the film blackening due to the oblique incidence of the diffracted beam.

Least squares refinements using the intensity data were performed by means of the program POWOW, originally written by the late W. C. Hamilton, Brookhaven National Laboratory, USA, and modified locally for an IBM 370/155 computer.⁵ The program permits the use of sums of intensities for unresolved overlapping reflexions as input data.

STRUCTURE ANALYSIS

Our powder diffraction data obtained for Pd₆P were in good agreement with Gullman's results. It was found, however, that some of the weaker diffraction lines listed by Gullman actually belong to the Pd₈P phase, which was present in small amounts in his Pd₆P sample.

Table 1. Powder diffraction data for $Pd_{\mathfrak{g}}P$. (Guinier-Hägg camera, $CuK\alpha_1$ radiation, internal calibration standard silicon a=5.43054 Å, intensities from film scanner.)

	$Q \times 10^5$.7	
h k l	obs	calc	- d _{obs}	$I_{ m obs}$
001		1 689		
020	4 485	4 488	4.722	-
021		6 177		
$0\ 0\ 2$	6754	$6\ 756$	3.848	-
$0\ 2\ 2$	11 244	11 244	2.982	7
110	13 543	13 547	2.717	6
īii	15 2 06	$15\ 204$	2.564)	40
111	$15\ 266$	$15\ 267$	2.559)	
$0 \ 4 \ 0$	17 947	17 951	2.360	41
041	19 664	19 640\	2.255	100
0 2 3∫	19 004	19 689)		100
Ī 1 2	$20\ 240$	$20 \ 240$	2.223	73
112	$20\ 370$	$20\ 366$	2.216)	
130	$22\ 515$	$22\ 522$	2.107	33
$\bar{1}$ 3 1	$24\ 174$	24 180	2.034	58
131	$24\ 248$	24 243	2.031	90
042	24 698	24707	2.012	11
004	$27\ 018$	$27\ 025$	1.924	3
113	28 646	28 653	1.868	9
113	$28\ 835$	$28\ 843$	1.862	8
f i 3 2	$29\ 207$	$29\ 215$	1.850	12
132	$29\ 334$	$29\ 342$	1.846	8
$0\ 2\ 4$	$31\ 512$	31 513	1.781	1
043	33 155	33 153	1.737	1
Ī 3 3		$37\ 629$		
133		37 819		
0 6 0		40 390)		
Ĩ 1 4	$40 \ 442$	$40 \ 445$	1.572	5
150		40 473)		
114	40702	40 698	1.567	4
061	$42\ 083$	$42\ 079$	1.542	4
Ī 5 1		$42\ 131$		

	¥ // 10		$ d_{ m obs}$	$I_{ m ob}$
$h \ k \ l$	obs	calc	ods	- 00
151		42 194		
044		44 976		
025	46 709	46 714	1.463	:
062		47 146)		•
$\tilde{1}$ 5 2	47 142	47 166	1.456	28
152	47 273	47 293	1.454)	-\
$\bar{1}$ $\bar{3}$ $\bar{4}$	1. 2.0	49 420	1,101,	
134		49 674		
200	49 694	49 699	1.418	20
$\tilde{2} \tilde{2} \tilde{0}$	10 001	54 187	1.110	_`
2 2 0 I 5 3ነ		55 580))	
0 6 3	55 561	55 592	1.342	
Ī i 5)	00 001	55 615)	1.012	
153	55 749	55 770	1.339	49
$\overline{2}$ $\overline{2}$ $\overline{1}$	00 110	55 812	1.000	
115	55 936	55 931	1.337	
221	00 000	55 939	1.001)	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		56 328		
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		56 582		
045	60 161	60 178	1.289	ē
006	60 795	60 806	1.282	10
222	00 133	60 816	1.202	10
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		61 070		
Ī 3 5	64 585	64 590	1.244	11
135	64 901	64 907	1.241	10
026	04 901	65 294	1.241	10
1 5 4)		67 372)	1	
170}	67 379	67 400	1.218	
064	01 318	67 415)	1.210	44
	67 647		1.216	44
154	67 647	67 625	1.210	
$\frac{2}{1} \frac{4}{7} \frac{0}{1}$		67 650)	J	
171		69 058		
171	60 100	69 121	1.202	24
$ar{2} \ 2 \ 3$	$69\ 192$	69 198	1.202	24
$ar{2}$ 4 1 2 4 1		69 276		
$\begin{array}{c} 2 & 4 & 1 \\ 2 & 2 & 3 \end{array}$	00 E04	69 402	1 100	19
4 2 3	$69\ 594$ $71\ 778$	69 578	1.199	
080		71 805	1.180	3 6
81	73 471	73 494	1.167	U
7 2		74 093 74 163		
16				
$\begin{smallmatrix} 1&7&2\\ 2&4&2 \end{smallmatrix}$	74 272	$74 220 \\ 74 280$	1.160	4
242				3
242	$74\ 512$	74 533	1.158	o
$\frac{1}{2}$ $\frac{1}{6}$ $\frac{6}{4}$	FC 400	74 543	1 144	6
204	76 462	76 470	1.144	2
204	76 953	76 977	1.140	1
82	78 542	78 561	1.128	22
46	78752	78 757	1.127	
224		80 958		
2 2 4 [7 3]		81 465		
		82 507		
55	00 200	82 542	1.100	
0 6 5	$82\ 576$	82 617}	1.100	36
4 3		82 662		
[73]		82 697J		
l 5 5		82 858		
2 4 3		83 042		
36 136		83 138 83 518		
136				

 $Q \times 10^{5}$

7

For identification purposes, revised diffraction data for Pd₅P as obtained in our study are given in Table 1.

In order to analyze the PdaP structure we started with an attempt to use single crystal methods. As mentioned above, the "crystals" available were of extremely poor quality. However, from the best crystal conglomerates that we found, oscillation films were obtained, which indicated a rotation axis of about 2.8 Å. Weissenberg recordings of the zero, first and second layer lines indicated that Pd,P might have an orthorhombic symmetry, with lengths of the remaining two axes of about 9.4 and 7.7 Å. An attempt to index the powder pattern on this basis was partially successful, but at several positions, where we expected to find only one single diffraction line, we observed pairs of closely spaced lines. We therefore tentatively assumed a monoclinic symmetry with the monoclinic angle differing only slightly from 90°. With this assumption all lines in the powder pattern could be accounted for in a very satisfactory manner. The indexed powder data are presented in Table 1.

On re-inspection of the "single crystal" films we found that the distribution of diffraction spots might well be compatible with a monoclinic symmetry. The deviations from orthorhombic symmetry were most probably masked by satellite spots, stacking fault effects, and by the very small deviation from 90° for the monoclinic angle in combination by twinning on (100).

The unit cell dimensions of Pd₆P at 24 °C as derived from the powder diffraction data were: a=2.8370(2) Å, b=9.4409(6) Å, c=7.6945(5) Å, $\beta=90.198(4)$ ° (numbers in parentheses following numerical values are the calculated standard deviations referring to the least significant digits). The standard deviations as returned by the CELNE program include no allowance for systematic errors. The absolute errors are estimated to be less than 0.04 %.

In agreement with Gullman ¹ we observed no significant variations in the cell dimensions between samples of different compositions and heat treatments. The cell volume of 206.1 ų is consistent with a cell content of 12 palladium and 2 phosphorus atoms. Both the powder and the "single crystal" diffraction data indicated face centering. Furthermore, no reflexions with l=2n+1 were observed among the (h0l) reflex-

ions. This indicates Cc or C2/c as the most probable space groups. However, among the equivalent positions in these space groups there are no sets of a multiplicity lower than fourfold, while there are only two phosphorus atoms in the unit cell. If the atoms occupy ordered positions in the structure, the space group symmetry can therefore be no higher than C2, Cm or C2/m.

The numerous overlaps in the powder diffraction patterns and the extremely poor single crystal data precluded the use of X-ray intensities for obtaining an initial structure proposal for Pd,P. An attempt was therefore made to derive the atomic arrangement by simple space considerations. Accordingly, we assumed that no Pd - Pd and Pd - P distances in Pd, P should be shorter than 2.7 Å and 2.2 Å, respectively, as inferred from the structure data known for Pd, P (Sellberg 6) and Pd, P (Rundqvist and Gullman 7). Furthermore, we assumed that the atomic arrangement was ordered. An analysis of the various possibilities showed that there was only one principal way of arranging the atoms in accordance with the assumed spatial requirements. This structure proposal had C2 symmetry, with the following approximate atomic positions: 4 Pd in 4c: $x \sim 0$, y = 0 (arbitrarily), $z \sim 0.32$; 4 Pd in 4c: $x \sim 0$, $y \sim 0.73$, $z \sim 0.19$; 2 Pd in 2b: $y \sim 0.28$; 2 Pd in 2a: y ~ 0.44 ; 2 P in 2b: $y \sim 0.58$. With this structure proposal as the starting point we proceeded to make refinements based on powder diffraction intensities.

In the powder diffractometer work, we recorded intensities up to a maximum in θ of about 45°. For diffraction angles higher than this value the overlapping becomes so extensive that we considered an extension of the intensity measurements to be of very little value. The powder sample used contained small amounts of palladium. Within the θ -range recorded, the three palladium reflexions (111), (311), and (222) overlap reflexions from Pd₆P. From measurements of the intensities of the remaining well resolved palladium reflexions we could calculate the intensities for (111), (311), and (222). In the case of (111) and (222), the calculated intensities amounted only to 4.4 and 8.6 %, respectively, of the total integrated intensities for the overlapping palladium and Pd Preflexions. For each of these two cases of overlap we subtracted the calculated palladium intensity from the total intensity and used the resulting value as the Pd₆P intensity in the subsequent calculations. The palladium reflexion (311), which overlaps

Table 2. Structure data for $Pd_{\bullet}P$ based on space group C2/c.

		\boldsymbol{x}	y	z
8 Pd(1) 4 Pd(2)	in 8 <i>f</i> in 4 <i>e</i>	0.997(5) 0	$0.133(1) \\ 0.422(2)$	0.064(2)
2 P(randomly)	in 4e	0	0.715(12)	ł

Isotropic overall temperature factor B = 0.8(3) Å².

the Pd₆P (081) reflexion, had a calculated intensity which amounted to a considerable fraction of the measured total intensity. Any correction for this overlap would therefore be very uncertain, and accordingly we omitted the Pd₆P (081) reflexion from the intensity data. The intensity material finally obtained comprised 23 data, seven of which consisting of intensities for single, non-overlapped Pd₆P reflexions, the remainder being sums of overlapping Pd₆P reflexions.

For the intensity calculations we assumed the

lithium fluoride monochromator crystal to be ideally imperfect. This leads to the expression

 $(1 + \cos^2 2\theta \cos^2 2\theta_{\rm M})/\sin^2 \theta \cos \theta$

for the Lorentz-polarization correction, where θ is the Bragg angle for the sample, and $\theta_{\rm M}$ that for the monochromator. The weights w assigned to the observed intensities in the refinement procedure were based essentially on counting statistics but modified according to the formula

$$w = [N_{\rm T} + N_{\rm B} + 0.02(N_{\rm T} - N_{\rm B})^2]^{-1}$$

where $N_{\rm T}$ is the total integrated intensity, and $N_{\rm B}$ the background intensity, in order to reduce the emphasis on the strongest intensities. In the structure factor calculations, atomic scattering factors for palladium were taken from Cromer and Waber, for phosphorus from Hanson et al. and dispersion correction from Cromer.

The refinements based on space group C2 never reached a state of complete convergence, and the parameters oscillated slightly between successive cycles. At the termination of the refinement the weighted agreement factor R_w , defined as

Table 3. Observed and calculated intensities for Pd_sP. (Diffractometer data, CuKα radiation).

h k l	$I_{\mathbf{o}}$	$I_{\mathbf{c}}$	$h \ k \ l$	I_{o}	$I_{\mathbf{c}}$	h k l	I_{o}	I_{e}
0 2 2	1 426	1 551	0 2 5)			$egin{array}{cccc} ar{1} & 7 & 1 \\ 1 & 7 & 1 \\ \hline 2 & 2 & 3 \\ \hline 2 & 4 & 1 \\ 2 & 4 & 1 \\ 2 & 2 & 3 \\ \hline 2 & 0 & 4 \\ 2 & 0 & 4 \\ 2 & 0 & 2 \\ \end{array}$		
$\begin{array}{c} 1 & 1 & 0 \\ \hline 1 & 1 & 1 \\ 1 & 1 & 1 \end{array}$	805	688	$ \begin{array}{c} 0 & 2 & 5 \\ 0 & 6 & 2 \\ \hline 1 & 5 & 2 \\ \underline{1} & 5 & 2 \\ 1 & 3 & 4 \\ 1 & 3 & 4 \end{array} $	3 648	3 444	171		
1 1 1)	5 481	5 243	1 5 2	0 040	0 111	$\frac{2}{9}$ $\frac{2}{4}$ $\frac{3}{1}$	4 016	4 536
$\alpha a \alpha$	6 415	6 284	1 2 2 1			241		
0 4 1 0 2 3 1 1 2 1 1 2 1 3 0 1 3 1 1 3 1	0 110	0 201	1 3 4	1 739	2 304	$\begin{array}{cccccccccccccccccccccccccccccccccccc$		
0 2 3	33 410	32 400	<u>2</u> 0 0)			2 0 4 (349	258
1 1 2	90 410	32 4 00	1 3 4 2 0 0 1 5 3 0 6 3 1 1 5 3 1 1 5 3 1 1 2 2 1 2 2 0 2 2 2 2 2 2 2 1 3 5 6 1 1 7 0			2041	010	200
1121	4 598	4 715	0 6 3			004	1 944	1 540
131	4 090	4 /10	153			1731		
1 3 1}	9 430	9 629	$\{ \hat{2}, \hat{2}, \hat{1} \}$	6 784	6 927	Ī 5 5		
U 4 Z)			1 1 5			046 173 1555 0243 173 1755 243 136		
004	841	759	$\frac{2}{9}$ 2 1			2 4 3		
1 1 3			202			007	2 122	1 911
$\frac{1}{1}$ $\frac{1}{3}$ $\frac{3}{2}$	5 638	5 436	0 4 5			155		
1 3 2			006	2 528	1 989	<u>2</u> 4 3		
024	277	202	2 2 2	2 020	1 909	1 3 6		
043	280	431	2 2 2J			1 3 6j		
114			1 3 5	3 075	2 830			
150	1 542	1 410	0 2 6	0010	2 000			
114			$\overline{1}$ $\overline{5}$ $\overline{4}$					
0 0 4 1 1 3 1 3 2 1 3 2 1 0 2 4 0 4 3 0 6 0 1 1 4 1 5 0 1 1 5 1 1 5 1	4.46		170	0.016				
151	446	444	$ \begin{array}{c c} 0 & 6 & 4 \\ 1 & 5 & 4 \end{array} $	3 916	4 235			
191)			$\begin{bmatrix} 1 & 5 & 4 \\ 2 & 4 & 0 \end{bmatrix}$					

 $R_w = \sum w||\mathbf{I}_{\mathbf{o}}| - k|I_{\mathbf{o}}||/\sum |I_{\mathbf{o}}||$

(where k is the scale factor and $I_{\rm o}$ and $I_{\rm c}$ denote intensity values for separate, resolved reflexions or sums of non-resolved, overlapping reflexions) oscillated between 0.11 and 0.12. The good agreement between observed and calculated intensities substantiated the essential features of the proposed structure. It was furthermore observed that the magnitude of the calculated intensities for all (h0l) reflexions with l odd was far below the limit of detection in the powder diffractometer measurements.

A study of the correlation matrix revealed strong correlations between some of the parameters refined. This indicates that the oscillatory behaviour of the refinement might be due to the fact that the structure proposed very nearly conforms to an atomic arrangement of higher symmetry. A description in terms of C2/c symmetry would involve one eightfold and one fourfold palladium position, and one fourfold position randomly occupied by two phosphorus atoms. This would lead to a reduction in the number of positional parameters from eight in C2 symmetry to five in C2/c symmetry.

A refinement based on C2/c symmetry converged satisfactorily to a final R_w of 0.113. The corresponding value for the unweighted agreement factor R was 0.058. The sign of the x parameter for the eightfold palladium position remained ambiguous, however, due to the fact that every pair of corresponding (hkl) and $(\bar{h}kl)$ reflexions was overlapping in the powder diffractometer data. The higher resolution in the Hägg-Guinier films afforded a possibility of resolving this ambiguity. A refinement based

Table 4. Observed and calculated intensities for $Pd_{\bullet}P$. (Film scanner data, $CrK\alpha_1$ radiation).

h k l	I_{o}	$I_{\mathbf{c}}$	$h \ k \ l$	I_{o}	$I_{ m c}$
0 2 2	35	30	130	156	146
$\begin{array}{c} 1 & 1 & 0 \\ \overline{1} & 1 & 1 \end{array}$	$\begin{array}{c} 21 \\ 84 \end{array}$	$\begin{array}{c} 22 \\ 70 \end{array}$	$\begin{bmatrix} \bar{1} & 3 & 1 \\ 1 & 3 & 1 \end{bmatrix}$	261	240
111	67	62	042	44	51
040	181	175	Ī13	46	45
041	470	400	113	49	43
0 2 3	470	490	$rac{1}{1} rac{1}{3} rac{2}{2}$	51	47
112	182	182	132	51	50
112	181	188			

on film scanner intensity data as obtained from $\operatorname{Cr} K\alpha_1$ powder films was therefore carried out. The final R_w and R values for this refinement were both 0.058, and the value obtained for the x parameter turned out to be insignificantly different from zero.

The final structure data as obtained from the C2/c refinements are given in Table 2. Lists of observed and calculated intensities are given in Tables 3 and 4.

DESCRIPTION AND DISCUSSION OF THE Pd.P STRUCTURE

The structure of Pd₆P is shown in projection on the (100) plane in Fig. 1. Calculated interatomic distances are given in Table 5.

The structure can conveniently be described in terms of triangular prismatic building blocks. Four Pd(1) and two Pd(2) atoms are situated at the corners of a slightly deformed trigonal prism. The phosphorus atoms are situated at the centres of every second Pd, prism. The prisms are stacked into a closely packed network, where each palladium atom coordinates 12-15 palladium and phosphorus neighbours. In addition to the six palladium atoms at the corners of the prism, each phosphorus atom has three more remote palladium neighbours situated outside the quadrilateral faces of the prism. The augmented trigonal prismatic ninecoordination for phosphorus is very common with metal-rich transition metal phosphide structures and has been discussed at length earlier.12,13 In particular, this coordination oc-

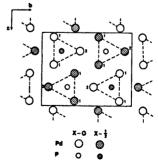


Fig. 1. The crystal structure of Pd₀P projected on (100). Only half of the phosphorus positions as indicated in the figure are occupied by phosphorus atoms.

Table 5. Interatomic distances (Å) in Pd_6P . Distances up to 3.8 Å are included. When greater than one, the number of equivalent distances from a central atom to its neighbours precedes the notation for the neighbouring atoms.

P Pd(1) Pd(1) P Pd(2) Pd(2) 2Pd(1) Pd(2) Pd(2) Pd(2) Pd(2) Pd(2)	2.16(5) 2.16(5) 2.69(2) 2.80(2) 2.81(6) 2.82(2) 2.83(2) 2.83(2) 2.84(0) 2.84(1) 2.86(1) 2.86(2) 3.09(2)	Pd(2)	P 2Pd(1) 2Pd(2) 2Pd(2) 2Pd(1) 2Pd(1) 2Pd(1) 2Pd(1) 2Pd(2) Pd(2)	2.83(2) 2.84(0) 2.84(1) 2.86(1) 3.09(2) 2.16(5) 2.16(5) 2.42(10) 2.76(12)
			Pd(2) 2Pd(1) 2P	

curs ^{6,7} in the two palladium phosphides Pd_{4,8}P and Pd₃P.

As regards the twinning observed for the Pd_sP crystals it is evident from the structural results that twinning on (100) requires very small distortions of the structure at the twin boundaries. The stacking faults may be connected with the distribution of the two phosphorus atoms among the four positions available. The distribution might have a certain degree of order, and it is conceivable that Pd.P might ideally crystallize with C2 symmetry as we assumed in our preliminary structure analysis. In the real structure, stacking faults may occur among the sheets of filled and unfilled triangular prisms traversing the structure parallel to (001), and the diffraction data would accordingly correspond to C2/c symmetry.

The atomic arrangement in $\mathrm{Pd_6P}$ resembles very closely that occurring in the boride $\mathrm{Re_3B}$. This compound crystallizes with the orthorhombic Cmcm symmetry. The unit cell contains twelve rhenium and four boron atoms, and the unit cell dimensions: a=2.890 Å, b=9.313 Å, c=7.258 Å, are not much different from those of $\mathrm{Pd_6P}$. A projection on (100) of the $\mathrm{Re_3B}$ structure would look very nearly the same as Fig. 1, with rhenium atoms at the palladium positions and boron atoms filling all triangular $\mathrm{Re_6}$ prisms.

The Re₃B-type metal skeleton is apparently very stable with respect to changes in the nonmetal sublattice. In the so-called "filled" Re₃B-type structure, as investigated and discussed thoroughly by Boller and Nowotny, 15-19 the triangular prismatic holes as well as octahedral holes, which also occur between the metal atoms, are filled with non-metal atoms. In contrast, Pd₅P exhibits an Re₃B-type metal sublattice with none of the octahedral, and only half of the triangular prismatic holes, filled.

Acknowledgements. The authors thank Mr. J. Gullman and Mr. N.-O. Ersson for valuable help in computer programming and handling the powder diffractometer and the film scanner. V. K. and M.R.C.S. are indebted to the International Seminar in Chemistry, Uppsala, and the Swedish International Development Authority (SIDA) for financial support. The financial support for this work provided by the Swedish Natural Science Research Council is gratefully acknowledged.

REFERENCES

- Gullman, L.-O. J. Less-Common Metals 11 (1966) 157.
- Ersson, N.-O. Institute of Chemistry, Uppsala 1973. Unpublished.
- Malmros, G. and Werner, P.-E. Acta Chem. Scand. 27 (1973) 493.
- Hägg, G. and Regnström, G. Ark. Kemi Mineral. Geol. A 18 (1944) No. 5.
- 5. Gullman, J. Institute of Chemistry, Uppsala 1973. Unpublished.
- 6. Sellberg, B. Acta Chem. Scand. 20 (1966) 2179.
- Rundqvist, S. and Gullman, L.-O. Acta Chem. Scand. 14 (1960) 2246.
- Azaroff, L. V. Acta Crystallogr. 8 (1955) 701.
 Cromer, D. T. and Waber, J. T. Acta Crystallogr. 18 (1965) 104.
- Hanson, H. P., Herman, F., Lea, J. D. and Skillman, S. Acta Crystallogr. 17 (1964)
- 11. Cromer, D. T. Acta Crystallogr. 18 (1965) 17.
- 12. Rundqvist, S. Ark. Kemi 20 (1962) 67.
- 13. Lundström, T. Ark. Kemi. 31 (1969) 227.
- Aronsson, B., Bäckman, M. and Rundqvist, S. Acta Chem. Scand. 14 (1960) 1001.
- 15. Boller, H. and Nowotny, H. Monatsh. Chem. 98 (1967) 2127.
- Boller, H. and Nowotny, H. Monatsh. Chem. 99 (1968) 721.
- Nowotny, H., Boller, H. and Beckmann, O. J. Solid State Chem. 2 (1970) 462.
- Boller, H. Monatsh. Chem. 102 (1971) 431.
 Boller, H. Monatsh. Chem. 104 (1973) 545.

Received March 29, 1974.

Acta Chem. Scand. A 28 (1974) No. 7