

## Refinement of the Crystal Structure of Potassium Barium Hexathionate

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The crystal structure of potassium barium hexathionate,  $K_2Ba(S_6O_6)_2$ , has been refined by full-matrix least squares for 2375 independent non-zero reflections. The data were collected by means of a single-crystal diffractometer using  $MoK\alpha$  radiation (Nb-filtered). The refinement converged at a conventional  $R$  value of 0.020.

The space group is  $P2/c$  (No. 13) with two formula units in a unit cell of dimensions  $a = 11.591(4)$  Å,  $b = 10.835(5)$  Å,  $c = 9.145(3)$  Å,  $\beta = 111.93(4)^\circ$ .

The six-membered sulphur chain of the hexathionate anion has the *cis-cis* rotational isomeric form. The bond lengths and angles, from one end of the chain to the other, are 2.119(1), 2.042(2), 2.056(2), 2.039(2), 2.110(1) Å, and 101.25(5) $^\circ$ , 110.20(5) $^\circ$ , 108.99(5) $^\circ$ , 99.95(5) $^\circ$ . The SSS/SSS dihedral angles are 109.4 $^\circ$ , 89.0 $^\circ$ , and 106.3 $^\circ$ .

The crystal structures of two salts of hexathionic acid have been published, both in 1965. In one of these salts,  $K_2Ba(S_6O_6)_2$ ,<sup>1</sup> the anion occurs in the *cis-cis* rotational isomeric form, whereas in the other one,  $[Co(en)_2Cl_2]_2S_6O_6 \cdot H_2O$ ,<sup>2</sup> the anion has the *trans-trans* form. In the latter structure there was a tendency of variation in the lengths of the three middle S-S bonds, although the differences might not have been significant. Mainly to decide this point new sets of data have been recorded on a single-crystal diffractometer and refined by a full-matrix least squares program. The refinements and results are described in the present and a following paper.<sup>3</sup>

### EXPERIMENTAL

The measurements were carried out on a Siemens automatic single-crystal diffractometer using  $MoK\alpha$  radiation (Nb-filtered) and a scintillation counter.

The crystals of  $K_2Ba(S_6O_6)_2$  were prepared by Foss and Palmork.<sup>4</sup> The sample used had been kept for thirteen years in a refrigerator, and showed no signs of decomposition. The crystal used for measurements of unit cell dimensions and intensities was a prism along the  $c$  axis, bounded by (100), (110), and (110), and terminated by (111). It was mounted with the  $c$  axis approximately parallel to the  $\phi$  axis of the diffractometer, and oriented by measurements of  $\theta, \chi$ , and  $\phi$  angles of six non-coplanar reciprocal vectors.

The five-value procedure and  $\theta - 2\theta$  scan technique, with scan width of  $0.70^\circ$ , were used. The maximum scan time per degree was 24 sec.

Two reflections, 12 0 0 and 0 10 0, were measured two times each at intervals of 50 reflections. The intensities of these reflections decreased by 6 % during the data collection, and were used to bring the net intensities to a common scale.

Reflections were measured up to  $\theta = 28^\circ$ , and only 215 of the 2590 independent reflections within this region were found to have net intensities below three times its standard deviation. These reflections were assigned an intensity equal to this limit, and labelled as unobserved.

Beside Lorentz and polarization corrections, the intensities were also corrected for absorption. The linear absorption coefficient for  $\text{MoK}\alpha$  is  $34.7 \text{ cm}^{-1}$ . The crystal used had a length along  $c$  of 0.330 mm, and the distances between the (100), between the (110), and between the (110) boundary faces were 0.087, 0.130, and 0.130 mm, respectively. The number of Gaussian grid points used along the  $a$ ,  $b$ , and  $c$  axes were 6, 6, and 12, respectively. The absorption factors, by which the intensities were multiplied, varied from 1.29 to 1.51.

The scattering factor curves used where those listed in *International Tables for X-Ray Crystallography*,<sup>5</sup> except for barium ion, for which the curve given by Thomas and Umeda<sup>6</sup> was used. The curves for barium, potassium, and sulphur were corrected for anomalous dispersion using the values given by Cromer,<sup>7</sup> and taking the amplitude as the corrected value.

The refinement was carried out with a full-matrix least squares program minimizing the function

$$r = \sum W(|F_o| - K|F_c|)^2$$

The intensity data were eventually corrected for secondary extinction with a program written by K. Åse of this Institute.

For further details concerning the data collection and the programs used, see Ref. 8.

The unit cell dimensions, calculated by means of a least squares program using the  $\theta$  angles ( $\theta = 21 - 28^\circ$ ) of 15 reflections measured on the diffractometer, are  $a = 11.591(4)$  Å,  $b = 10.835(5)$  Å,  $c = 9.145(3)$  Å,  $\beta = 111.93(4)^\circ$ .

The space group is  $P2/c$  (No. 13) with two  $\text{K}_2\text{Ba}(\text{S}_6\text{O}_6)_2$  units per unit cell.<sup>4</sup>

#### REFINEMENT

The least squares refinement was started using the positional parameters given by Foss and Johnsen,<sup>1</sup> and individual isotropic thermal parameters that were the averaged values of those given for the two projections. Refinement on scale factor, positional parameters, and isotropic thermal parameters resulted in an  $R$  value ( $\sum |F_o| - |F_c| / \sum |F_o|$ ) of 0.095. On using anisotropic thermal parameters for barium, potassium, and sulphur the  $R$  value was reduced to 0.031, and with anisotropic, thermal parameters also for the oxygen atoms, the refinement converged at  $R = 0.022$ .

For the reflections with highest intensities, the observed structure factors were considerable lower than the calculated ones. Secondary extinction corrections were therefore carried out. The expression given by Zachariasen<sup>9</sup> was used. With the absorption term equal to one,  $C$  was found to be  $6.0 \times 10^{-7}$ . The observed values of the structure factors now corresponded well to the calculated ones also for the strongest reflections.

The final  $R$  value was 0.020, with unobserved reflections included when  $|F_c|$  is greater than the observable limit. There were no shifts of the parameters in the last refinement cycle.

A final difference electron density map showed no peaks higher than 0.4 e/Å<sup>3</sup>.

Table 1. Atomic coordinates for potassium barium hexathionate. Origin at a centre of symmetry. Standard deviations are given in parentheses.

	<i>x</i>	<i>y</i>	<i>z</i>
Ba	0	0.01396(2)	$\frac{1}{4}$
K(1)	$\frac{1}{2}$	0	$\frac{1}{2}$
K(2)	0	0.44987(8)	$\frac{1}{4}$
S(1)	0.13933(6)	0.27403(5)	0.04470(7)
S(2)	0.33447(7)	0.28355(8)	0.16708(10)
S(3)	0.34815(9)	0.42062(8)	0.32588(9)
S(4)	0.40178(8)	0.34854(8)	0.54946(9)
S(5)	0.24695(8)	0.30389(7)	0.59363(10)
S(6)	0.25238(5)	0.10984(6)	0.57818(7)
O(1)	0.10136(21)	0.39894(17)	-0.00500(25)
O(2)	0.13175(21)	0.18877(19)	-0.07952(24)
O(3)	0.08578(18)	0.22999(17)	0.15501(23)
O(4)	0.36031(19)	0.06818(23)	0.70720(24)
O(5)	0.13276(16)	0.07734(17)	0.58622(20)
O(6)	0.25668(16)	0.07982(17)	0.42586(21)

Table 2. Thermal parameters expressed in the form  $\exp[-2\pi^2(h^2a^{-2}U_{11} + \dots + 2hka^{-1}b^{-1}U_{12} + \dots)]$ . All values have been multiplied by  $10^4$ . Standard deviations are given in parentheses.

	$U_{11}$	$U_{22}$	$U_{33}$	$U_{12}$	$U_{23}$	$U_{13}$
Ba	196(1)	202(1)	182(1)	0	0	77(1)
K(1)	284(5)	1270(12)	402(6)	169(6)	-366(7)	5(5)
K(2)	575(6)	254(4)	331(5)	0	0	156(4)
S(1)	300(3)	200(3)	274(3)	-33(3)	7(2)	88(3)
S(2)	302(4)	574(5)	549(5)	-72(4)	-40(4)	153(4)
S(3)	647(6)	387(4)	423(5)	-270(4)	39(4)	41(4)
S(4)	521(5)	554(5)	416(5)	-327(4)	56(4)	-15(4)
S(5)	653(6)	303(4)	574(5)	-119(4)	-124(4)	285(5)
S(6)	197(3)	279(3)	223(3)	-30(2)	32(2)	57(2)
O(1)	621(16)	250(10)	480(14)	56(10)	127(10)	145(12)
O(2)	560(15)	395(11)	446(14)	-124(11)	-172(10)	231(12)
O(3)	374(12)	302(9)	392(12)	-47(8)	38(8)	180(10)
O(4)	306(12)	676(15)	359(13)	2(11)	193(11)	-18(10)
O(5)	230(9)	369(10)	324(11)	-61(8)	7(8)	139(9)
O(6)	326(11)	335(10)	297(10)	-19(8)	-21(8)	167(9)

Tables 1 and 2 give the final atomic parameters with standard deviations from least squares. The oxygen atoms are numbered in an order different from that published earlier,<sup>1</sup> and have now numbers in accordance with those in the telluropentathionates described in Refs. 8 and 10.

As seen from Table 2 there is an extremely high value of  $U_{22}$  for K(1) compared to all other thermal parameters. This agrees with what was found by the refinement of the two-dimensional film data through difference maps. In the  $hk0$  zone the value of  $B$  in  $\text{\AA}^2$  was  $2.5 + 10.0 \cos^2 \phi$  for K(1), where  $\phi$  is the angle between the normal of the reflecting plane and the direction of maximum vibration of the atom; this direction being parallel to the  $b$  axis.<sup>1</sup>

The observed structure factors and those calculated from the final atomic parameters are listed in Table 3.

Table 3. Observed and calculated structure factors ( $\times 10$ ) for potassium barium hexathionate. A minus sign on  $F(O)$  indicates an unobserved reflection.

H	K	L	$F(O)$	$F(C)$	H	K	L	$F(O)$	$F(C)$	H	K	L	$F(O)$	$F(C)$	H	K	L	$F(O)$	$F(C)$					
3	0	0	1361	1347	-9	0	8	195	190	5	1	2	654	-579	-8	1	6	709	-822					
4	0	0	246	274	-8	0	8	369	366	6	1	2	274	-272	-7	1	6	506	-522					
5	0	0	286	321	-7	0	8	620	628	7	1	2	575	-585	-6	1	6	235	-253					
6	0	0	710	739	-6	0	8	513	529	8	1	2	93	-87	-5	1	6	416	-415					
7	0	0	725	641	-5	0	8	152	164	9	1	2	633	-632	-4	1	6	136	-139					
8	0	0	1835	1781	-4	0	8	889	875	10	1	2	419	-431	-3	1	6	856	-865					
9	0	0	760	749	-3	0	8	849	849	11	1	2	262	-266	-2	1	6	81	-86					
10	0	0	-120	4	-2	0	8	1109	1143	12	1	2	223	-207	-1	1	6	621	-625					
11	0	0	130	154	-1	0	8	580	580	13	1	2	304	-361	0	1	6	924	-937					
12	0	0	167	169	-1	0	8	115	115	14	1	2	59	-56	1	1	6	400	-626					
13	0	0	215	181	1	0	8	673	681	14	1	3	115	-110	-1	1	6	61	-40					
14	0	0	296	282	2	0	8	740	739	13	1	3	249	-254	3	1	6	542	-535					
-14	0	0	2461	-451	3	0	8	-55	-25	-12	1	3	228	223	4	1	6	380	-382					
-13	0	0	158	-159	4	0	8	689	703	-11	1	3	141	142	5	1	6	292	-289					
-12	0	0	533	-534	5	0	8	461	460	-10	1	3	52	-57	6	1	6	264	-262					
-11	0	0	2129	-1313	6	0	8	536	535	-9	1	3	438	448	7	1	6	417	-418					
-10	0	0	368	-120	10	0	8	308	-304	-8	1	3	592	594	0	1	6	248	-246					
-9	0	0	753	-754	-11	0	10	172	-169	-7	1	3	-44	-18	9	1	6	474	-468					
-8	0	0	61	-10	10	0	10	454	-451	-6	1	3	-693	-14	1	1	7	-56	-28					
-7	0	0	65	-74	0	0	10	467	-462	-5	1	3	137	-130	1	1	7	326	332					
-6	0	0	1723	-1483	-6	0	10	251	-254	1	3	213	223	12	1	7	17	14	2	0	1575			
-5	0	0	1036	-1035	-7	0	10	506	-502	-3	1	3	123	149	-11	1	7	-54	10	6	2	0	1008	
-4	0	0	747	-780	-6	0	10	532	-530	-2	1	3	244	236	-10	1	7	396	404	7	2	0	349	
-3	0	0	2650	-2815	-5	0	10	776	-771	-1	1	3	74	-84	-9	1	7	96	-89	8	2	0	762	
-2	0	0	1188	-1213	-4	0	10	169	-177	0	1	3	896	905	-8	1	7	330	-340	9	2	0	318	
-1	0	0	522	516	-3	0	10	294	-299	1	1	3	-42	-44	-7	1	7	231	-223	10	2	0	692	
0	0	0	206	204	-2	0	10	514	-528	2	1	3	691	-701	-6	1	7	-47	-8	11	2	0	653	
1	0	0	213C	-215B	-1	0	10	564	-568	3	1	3	261	269	1	1	7	761	751	12	2	0	353	
2	0	0	1243	-1247	-1	0	10	454	-457	4	1	3	503	509	1	1	7	312	312	13	2	0	247	
3	0	0	223	-224	1	0	10	353	-366	5	1	3	438	423	-2	1	7	244	244	14	2	0	323	
4	0	0	727	-711	2	0	10	363	-366	6	1	3	230	-239	2	1	7	77	77	13	2	1	168	
5	0	0	2128	-2137	3	0	10	511	-506	7	1	3	140	140	-1	1	7	299	301	-13	2	1	-171	
6	0	0	1163	-1166	-7	0	12	239	-222	8	1	3	152	152	0	1	7	458	-469	-12	2	1	-99	
7	0	0	199	184	-6	0	12	245	-238	9	1	3	226	-221	1	1	7	413	-415	-11	2	1	199	
8	0	0	59	-77	-5	0	12	251	-253	10	1	3	-57	-65	2	1	7	520	531	-10	2	1	354	
9	0	0	1664	-1074	-4	0	12	568	-553	11	1	3	-55	-27	3	1	7	206	211	-9	2	1	168	
10	0	0	553	-537	1	0	10	849	849	12	1	3	182	191	4	1	7	76	75	-8	2	1	185	
11	0	0	238	-232	2	0	10	537	-526	-15	1	4	84	85	5	1	7	61	67	-7	2	1	67	
12	0	0	151	-141	3	0	10	511	-511	-14	1	4	310	310	6	1	7	205	205	-5	2	1	858	
13	0	0	474	-445	4	0	10	454	-455	13	1	4	330	334	7	1	7	217	203	5	2	1	344	
14	0	0	149	132	5	0	10	214	-227	-12	1	4	410	407	8	1	7	-58	-40	-4	2	1	876	
14	0	0	527	515	6	0	10	482	480	-11	1	4	282	300	-14	1	8	431	430	-3	2	1	973	
15	0	0	356	400	7	1	0	566	574	-1	1	4	540	549	-13	1	8	273	275	-2	2	1	1157	
16	0	0	89	81	8	1	0	913	929	-9	1	4	377	377	-12	1	8	328	339	-1	2	1	593	
11	0	0	309	315	9	1	0	193	193	-8	1	4	526	533	-11	1	8	234	243	0	2	1	847	
-10	0	0	1768	1800	10	1	0	273	271	-7	1	4	184	187	-10	1	8	481	483	1	2	1	169	
-9	0	0	745	763	-9	1	0	361	363	3	1	4	852	852	246	2	2	1	470	462	1	2	1	150
-8	0	0	1179	1179	11	0	0	464	427	-5	1	4	804	835	-9	1	8	197	182	3	2	1	145	
-7	0	0	226	234	12	0	0	199	199	-4	1	4	778	785	-7	1	8	328	328	4	2	1	174	
-6	0	0	1114	1114	14	1	0	272	265	-3	1	4	384	372	-6	1	8	739	744	3	2	1	63	
-5	0	0	1118	-1127	-14	1	1	118	-107	-2	1	4	1361	1340	-5	1	8	467	456	6	2	1	443	
-4	0	0	194	209	-13	1	1	220	-225	-1	1	4	317	-285	-4	1	8	329	332	7	2	1	304	
-3	0	0	501	505	-12	1	1	298	-213	0	1	4	566	577	-3	1	8	404	406	8	2	1	484	
-2	0	0	1529	-1897	-11	1	1	52	-42	1	1	4	409	428	-2	1	8	614	618	9	2	1	413	
-1	0	0	1481	1564	-10	1	1	144	136	2	1	4	207	220	-1	1	8	58	54	10	2	1	268	
0	0	0	745	763	-9	1	0	369	-368	3	1	4	852	872	0	1	8	415	416	11	2	1	198	
1	0	0	123	126	-8	1	1	54	-55	4	1	4	680	681	1	1	8	222	228	12	2	1	96	
2	0	0	1571	-1571	-7	1	1	695	-696	5	1	4	180	189	-9	1	8	547	547	13	2	1	80	
3	0	0	467	-500	-6	1	1	703	-707	7	1	4	148	149	-9	1	8	350	341	-14	2	1	641	
4	0	0	578	579	-4	1	1	699	724	8	1	4	334	326	5	1	8	305	310	-12	2	2	203	
5	0	0	848	856	-3	1	1	273	266	9	1	4	253	248	6	1	8	430	427	-11	2	2	146	
7	0	0	1017	1008	-2	1	1	899	878	10	1	4	315	322	-13	1	8	558	-35	-10	2	2	468	
8	0	0	237	224	-1	1	1	532	-503	11	1	4	279	274	-12	1	9	57	-26	-9	2	2	729	
9	0	0	64	-60	-6	0	1	491	-461	-15	1	5	104	-97	-1	1	9	177	176	-8	2	2	542	
10	0	0	560	553	1	1	1	180	187	-14	1	5	82	79	-8	1	9	269	274	-5	2	2	1189	
-15	0	0	6	345	309	2	1	1	1452	-1469	-13	1	5	82	79	-1	1	9	208	-274	-5	2	2	1356
-14	0	0	6	343	322	-14	1	1	203	-202	-2	1	5	155	153	-3	1	9	214	-244	-5	2	2	1163
-13	0	0	6	347	-163	-14	1	1	295	-295	1	1	5	63	63	-4	1	9	-58	-34	-6	2	2	525
-12	0	0	6	350	-568	6	1	1	148	-120	-9	1	5	331	343	-5	1	9	229	-320	-5	2	2	497
-11	0	0	6	361	-400	7	1	1	334	-343	-8	1	5	241	-244	-4	1	9	213	-211	-1	2	2	2421
-10	0	0	6	379	-529	-12	1	1	55	-62	1	1	5	57	-56	-4	1	9	225	-267	9	2	2	1351
-9	0	0	6	386	-524	9	1	1	243	-243	-6	1	5	475	-468	-11	1	9	73	-82	1	2	2	467
-8	0	0	6	391	-584	10	1	2	596	-593	3	1	5	3										

Table 3. Continued.

H	K	L	F(I0)	F(C)	H	K	L	F(I0)	F(C)	H	K	L	F(I0)	F(C)	H	K	L	F(I0)	F(C)
-3	2	3	458	-479	-2	2	7	137	-155	0	3	1	-41	3	-2	3	5	65	-48
-2	2	3	578	-566	-1	2	7	221	221	1	3	1	101.9	1007	-1	3	5	279	-295
-1	2	2	355	357	0	2	7	90	84	2	3	1	364	-379	0	3	5	63	52
0	2	2	116	-114	1	2	7	165	-161	3	3	1	361	-353	1	3	5	229	-231
1	2	2	750	760	2	2	7	149	-145	4	3	1	361	-565	2	3	5	873	-874
2	2	2	228	217	3	2	7	107	-103	5	3	1	361	-560	3	3	5	218	-248
3	2	2	239	-256	4	2	7	77	-61	6	3	1	429	-433	4	3	5	318	-323
4	2	2	239	283	5	2	7	187	-181	7	3	1	452	-457	5	3	5	612	-430
5	2	2	311	-312	6	2	7	269	268	8	3	1	482	-58	6	3	5	51	-30
6	2	2	477	11	7	2	7	-57	26	9	3	1	316	312	7	3	5	217	214
7	2	2	188	184	8	2	7	56	15	10	3	1	100	-108	8	3	5	55	-12
8	2	2	-51	34	-14	2	8	500	503	11	3	1	417	-422	9	3	5	209	-205
9	2	2	237	234	-13	2	8	402	405	12	3	1	112	-127	10	3	5	251	-243
10	2	2	206	-199	-12	2	8	458	466	13	3	1	66	-78	-14	3	6	330	-327
11	2	2	525	-41	-11	2	8	289	281	-14	3	2	266	-277	-13	3	6	469	-469
12	2	2	116	14	-10	2	8	360	360	-15	3	2	296	-325	-12	3	6	280	270
-15	2	4	214	219	-9	2	8	123	123	-16	3	2	424	-420	-11	3	6	370	-353
-14	2	4	253	248	-8	2	8	612	615	-17	3	2	638	-638	-10	3	6	346	-335
-13	2	4	454	454	-7	2	8	502	507	-18	3	2	51	-30	-9	3	6	443	-449
-12	2	4	793	753	-6	2	8	999	900	-19	3	2	581	-579	-8	3	6	212	-202
-11	2	4	438	446	-5	2	8	912	897	-20	3	2	287	-285	-7	3	6	622	-630
-12	2	4	694	714	-4	2	8	367	362	-21	3	2	542	-550	-6	3	6	220	-211
-9	2	4	442	447	-3	2	8	-48	9	-6	3	2	743	-740	-5	3	6	949	-931
-8	2	4	328	327	-2	2	8	656	657	-5	3	2	916	-901	-4	3	6	658	-669
-7	2	4	664	659	-1	2	8	291	287	-4	3	2	748	-740	-3	3	6	218	7
-4	2	4	641	650	0	2	8	544	546	-3	3	2	749	-740	-2	3	6	349	554
-5	2	4	238	258	1	2	8	728	746	-2	3	2	551	-550	-1	3	6	445	-452
-4	2	4	1774	1746	2	2	8	662	646	-1	3	2	293	-285	0	3	6	296	-304
-3	2	4	1172	1154	3	2	8	486	480	0	3	2	71	-78	1	3	6	701	-703
-2	2	4	856	852	4	2	8	427	406	1	3	2	1616	-1603	2	3	6	297	-300
-1	2	4	778	767	5	2	8	73	48	2	3	2	101	-108	3	3	6	748	-743
0	2	4	1392	1356	6	2	8	356	361	3	3	2	1077	-1080	4	3	6	213	-214
1	2	4	719	723	-13	2	9	-59	70	4	3	2	544	-543	5	3	6	411	-415
2	2	4	1112	1132	-12	2	9	113	-114	5	3	2	274	-282	6	3	6	193	-190
3	2	4	145	145	-11	2	9	145	-140	6	3	2	553	-552	3	3	6	319	-315
4	2	4	557	556	-10	2	9	100	-95	7	3	2	424	-418	0	3	6	299	-300
5	2	4	571	579	-9	2	9	97	92	8	3	2	301	-298	9	3	6	247	-242
6	2	4	540	544	-8	2	9	164	-163	9	3	2	579	-591	-14	3	7	58	43
7	2	4	-30	16	-7	2	9	-52	0	10	3	2	322	-320	-13	3	7	214	211
8	2	4	794	759	-6	2	9	100	-95	11	3	2	474	-468	12	3	7	149	142
9	2	4	487	480	-5	2	9	156	163	12	3	2	145	-136	-11	3	7	54	-31
10	2	4	456	435	-4	2	9	-51	31	-14	3	3	-55	35	-10	3	7	112	83
11	2	4	377	369	-3	2	9	313	-319	-13	3	3	304	314	-9	3	7	575	591
-15	2	5	133	-119	-2	2	9	190	192	-12	3	3	342	339	-8	3	7	428	425
-14	2	5	150	-150	-1	2	9	52	52	-13	3	3	340	-340	-7	3	7	8	2
-13	2	5	62	71	0	2	9	70	-270	-10	3	3	340	-146	-7	3	7	51	30
-12	2	5	60	-86	1	2	9	67	61	-9	3	3	87	100	-5	3	7	266	259
-11	2	5	133	-139	2	2	9	124	-113	-8	3	3	192	-188	-4	3	7	47	47
-10	2	5	266	-263	3	2	9	153	158	-7	3	3	801	797	-3	3	7	45	72
-9	2	5	160	164	4	2	9	-57	37	-6	3	3	579	573	-2	3	7	313	314
-8	2	5	221	219	5	2	9	124	-110	-5	3	3	141	-147	-1	3	7	770	772
-7	2	5	302	308	-12	2	10	271	-258	-4	3	3	690	692	0	3	7	408	407
-6	2	5	158	-113	-11	2	10	549	559	-3	3	3	550	541	1	3	7	50	48
-5	2	5	110	-119	-19	2	10	363	-371	-1	3	3	343	-163	2	3	7	50	47
-4	2	5	190	-185	-1	2	10	351	-351	1	3	3	340	-128	3	3	7	317	309
-3	2	5	322	-224	0	2	10	164	-150	1	3	3	343	-76	3	3	7	190	190
-2	2	5	231	-243	-7	2	10	298	-306	1	3	3	197	-195	5	3	7	54	-26
-1	2	5	637	617	-6	2	10	180	-173	2	3	3	813	813	6	3	7	204	201
0	2	5	247	-252	-5	2	10	590	-579	3	3	3	273	-253	7	3	7	451	454
1	2	5	311	-314	-4	2	10	535	-533	4	3	3	527	-525	-14	3	8	174	168
2	2	5	-46	-16	-3	2	10	604	-605	5	3	3	505	513	-13	3	8	280	277
3	2	5	169	-172	-1	2	10	500	-500	6	3	3	-48	11	-12	3	8	334	331
4	2	5	190	-185	-1	2	10	351	-351	7	3	3	210	-213	-11	3	8	311	309
5	2	5	322	-224	0	2	10	469	-466	1	3	3	303	-838	-10	3	8	420	-426
6	2	5	74	-82	-8	2	10	412	-412	9	3	3	303	-171	-9	3	8	316	-319
7	2	5	188	-185	2	2	10	159	-159	10	3	3	344	-528	-8	3	8	528	-535
8	2	5	-53	78	-1	2	10	406	-406	11	3	3	121	-106	-7	3	8	397	388
9	2	5	149	-147	-10	2	11	138	131	12	3	3	71	-31	-6	3	8	423	429
10	2	5	135	-149	-9	2	11	-58	11	-14	3	3	392	381	-5	3	8	105	99
-15	2	6	511	-497	-8	2	11	116	-121	-13	3	4	160	162	-4	3	8	672	661
-14	2	6	176	-187	-7	2	11	153	153	-12	3	4	210	208	-3	3	8	372	371
-13	2	6	285	-282	-6	2	11	-56	45	-11	3	4	454	449	-2	3	8	262	265
-12	2	6	351	-352	-5	2	11	-56	-37	-10	3	4	629	632	-1	3	8	492	499
-11	2	6	246	-245	-4	2	11	-54	-19	-9	3	4	306	310	0	3	8	392	342
-10	2	6	466	-464	-3	2	11	-54	-19	-8	3	4	604	593	1	3	8	292	291
-9	2	6	823	-820	-2	2	11	140	126	-7	3	4	184	188	2	3	8	472	469
-8	2	6	568	-578	-1	2	11	-56	-35	-6	3	4	536	546	2	3	8	313	316
-7	2	6	713	-705	0	2	11	94	-94	-5	3	4	395	403	3	3	8	532	532
-6	2	6	438	-440	0	3	0	2075	2095	-4	3	4	395	396	4	3	8	191	183
-5	2	6	418	-421	1	3	0	103	-55	-3	3	4	686	691	6	3	8	311	309
-4	2	6	393	-395	2	3	0	504	505	6	3	4	463	471	-5	3	8	336	-330
-3	2	6	1215	-1219	3	3	0	732	725	-1	3	4	752	757	-12	3	8	279	-753
-2	2	6	634	-634	4	3	0	529	533	0	3	4	868	877	-11	3	8	190	-304
-1	2	6	1014	-1014	5	3	0	624	624	-1	3	4	497	500	-10	3	8	39	-54
0	2	6	592	-593															

## POTASSIUM BARIUM HEXATHIONATE

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Table 3. Continued.

H	K	L	F(O)	F(C)	H	K	L	F(O)	F(C)	H	K	L	F(O)	F(C)	H	K	L	F(O)	F(C)
6	4	3	195	-188	-7	4	8	271	270	-2	5	2	-442	-38	2	5	6	590	-593
7	4	3	228	237	-6	4	8	131	129	-1	5	2	1938	-1922	3	5	6	343	-342
8	4	3	-53	-44	-5	4	8	435	424	0	5	2	962	-950	4	5	6	125	-126
9	4	3	-53	-16	-4	4	8	840	837	1	5	2	252	-220	5	5	6	556	-561
10	4	3	211	208	-3	4	8	529	516	2	5	2	273	-285	6	5	6	642	-636
11	4	3	97	-7	-2	4	8	565	545	3	5	2	501	-476	5	5	6	375	-355
-14	4	4	421	424	-1	4	8	513	504	4	5	2	570	-574	8	5	6	64	61
-13	4	4	-56	30	0	4	8	635	635	5	5	2	823	-818	-13	5	7	154	157
-12	4	4	284	284	1	4	8	60	-59	6	5	2	282	-297	5	5	7	198	195
-11	4	4	414	419	2	4	8	217	216	7	5	2	851	-856	-11	5	7	388	381
-10	4	4	422	420	3	4	8	425	413	8	5	2	655	-653	-10	5	7	215	208
-9	4	4	352	354	4	4	8	457	444	9	5	2	53	-53	17	9	5	7	225
-8	4	4	988	952	5	4	8	409	419	10	5	2	136	-135	-8	5	7	94	84
-7	4	4	995	855	6	4	8	251	244	11	5	2	531	-533	-7	5	7	300	296
-6	4	4	526	526	-12	4	8	87	-77	12	5	2	270	-268	-6	5	6	65	64
-5	4	4	511	492	-11	4	8	96	86	-4	5	3	132	-132	3	5	7	334	334
-4	4	4	793	751	-10	4	8	58	-37	-13	5	3	74	82	-4	5	7	781	777
-3	4	4	375	379	-9	4	8	240	-240	-12	5	3	52	-52	7	3	5	7	376
-2	4	4	1294	1292	-8	4	9	129	-115	-11	5	3	389	385	-2	5	7	353	358
-1	4	4	181	183	-7	4	9	141	-129	-10	5	3	376	371	-1	5	7	325	326
3	4	4	710	730	-6	4	9	70	-79	-9	5	3	562	564	0	5	7	145	-143
1	4	4	519	536	-5	4	9	228	236	-8	5	3	229	236	1	5	7	409	399
2	4	4	123	117	-4	4	9	248	-255	-7	5	3	212	217	2	5	7	180	186
3	4	4	337	354	-3	4	9	52	-45	-6	5	3	634	630	3	5	7	228	235
4	4	4	929	944	-2	4	9	97	-95	-5	5	3	724	726	4	5	8	398	395
5	4	4	601	552	1	4	9	216	-217	-4	5	3	355	350	7	5	8	277	274
6	4	4	649	649	0	4	9	251	-258	-3	5	3	702	700	6	5	7	121	117
7	4	4	293	264	1	4	9	110	118	-2	5	3	397	409	7	5	7	59	48
8	4	4	444	458	2	4	9	-54	19	-1	5	3	859	869	-13	5	8	294	283
9	4	4	270	261	3	4	9	-56	-24	0	5	3	171	-171	-12	5	8	473	470
10	4	4	134	188	4	4	9	-56	33	1	5	3	335	341	-11	5	8	178	165
-14	4	5	319	-329	-11	4	10	337	-337	2	5	3	763	759	-10	5	8	153	154
-13	4	5	-52	-1	-10	4	10	182	-170	3	5	3	321	315	-9	5	8	466	469
-12	4	4	-4	-9	-9	4	10	317	-315	4	5	3	764	762	-8	5	8	580	587
-11	4	5	342	-364	-8	4	10	246	-233	5	5	3	403	401	-7	5	8	141	141
-10	4	5	145	133	-7	4	10	443	-439	-14	5	4	141	143	0	5	8	645	648
-9	4	5	155	-153	0	4	10	385	-380	-13	5	4	270	269	3	5	8	302	307
-8	4	5	76	-76	1	4	10	556	-553	-12	5	4	677	675	2	5	8	186	184
-7	4	5	207	206	2	4	10	553	-552	1	5	4	773	770	3	5	8	351	353
0	4	5	145	-136	-9	4	11	84	-89	-10	5	4	252	242	4	5	8	447	439
1	4	5	820	-817	-8	4	11	57	-39	-9	5	4	183	179	5	5	8	56	552
2	4	5	197	-197	-7	4	11	218	225	-8	5	4	389	380	-12	5	9	56	572
3	4	5	331	226	-6	4	11	218	212	-7	5	4	887	890	-11	5	9	106	91
4	4	5	-49	-18	-5	4	11	55	23	-6	5	4	805	797	-10	5	9	295	-303
5	4	5	72	59	-4	4	11	84	67	5	5	4	71	-66	-9	5	9	165	-167
6	4	5	158	-155	-3	4	11	176	172	-4	5	4	1382	1374	-8	5	9	336	-326
7	4	5	-52	21	-2	4	11	185	-195	-3	5	4	717	700	-7	5	9	187	-181
8	4	5	118	-124	-1	4	11	181	-191	-2	5	4	435	-434	-6	5	9	323	-322
9	4	5	237	-237	0	4	11	82	-82	-1	5	4	703	703	-1	5	9	156	-157
-14	4	6	416	416	1	5	0	710	700	0	5	4	969	963	-4	5	9	1	10
-13	4	6	559	-561	-2	4	5	183	1826	1	5	4	745	743	-3	5	9	292	-292
-12	4	6	169	-113	-3	5	0	532	554	2	5	4	1021	1028	-2	5	9	398	-402
-11	4	6	516	-509	-4	5	0	63	74	-14	5	4	302	306	-1	5	9	377	-367
-10	4	6	491	-482	-5	5	0	481	484	4	5	4	552	555	0	5	9	295	-296
-9	4	6	327	-326	6	5	0	914	909	5	5	4	648	644	1	5	9	109	-107
-8	4	6	277	-274	7	5	0	93	86	6	5	4	449	-449	-7	5	9	292	-293
-7	4	6	467	-456	8	5	0	181	171	7	5	4	91	-74	3	5	9	54	-376
-6	4	6	840	-840	10	5	0	257	271	8	5	4	677	670	-9	5	9	307	309
-5	4	6	128	-125	9	5	0	257	257	-1	5	4	435	-434	-5	5	9	323	-325
-4	4	6	313	-212	11	5	0	511	506	10	5	4	430	429	-10	5	9	156	-157
-3	4	6	547	-547	12	5	0	81	75	-14	5	4	214	-215	-8	5	9	398	-392
-2	4	6	425	-420	13	5	0	278	275	-13	5	5	275	-267	-7	5	10	328	-320
-1	4	6	703	-703	-13	5	1	69	-54	-12	5	5	203	-203	-6	5	10	121	-113
0	4	6	113	-124	-12	5	1	533	-527	-11	5	5	55	-55	27	5	10	391	-384
1	4	6	710	-711	-11	5	1	149	-153	-11	5	5	242	-234	-4	5	10	389	-376
2	4	6	920	-928	-17	5	1	117	-111	-9	5	5	53	-47	-4	5	10	539	-528
3	4	6	518	-518	-9	5	1	374	-374	-8	5	5	509	-506	-5	5	10	507	-505
4	4	6	203	-202	-5	5	1	374	-374	-7	5	5	648	-645	-1	5	10	510	-505
5	4	6	311	-303	-7	5	1	493	-473	-6	5	5	344	-336	-9	5	10	232	-216
6	4	6	279	-281	-6	5	1	375	-377	-5	5	5	623	-624	1	5	10	444	-445
7	4	6	601	-601	-5	5	1	304	-299	-4	5	5	433	-424	-8	5	10	102	-97
8	4	6	141	-144	-4	5	1	593	-586	-3	5	5	444	-444	15	5	11	183	-174
-3	4	7	-58	-28	-3	5	1	361	-368	-2	5	5	200	-198	-6	5	11	173	-179
-13	4	7	155	152	-2	5	1	456	-443	-1	5	5	417	-402	-5	5	11	337	331
-12	4	7	77	64	9	5	1	357	-354	-14	5	5	680	-685	-4	5	11	202	200
-11	4	7	291	289	10	5	1	202	-204	-13	5	5	289	-286	7	6	0	483	486
-10	4	7	307	307	11	5	1	118	-121	-12	5	5	692	-687	9	6	0	483	486
-9	4	7	99	-99	3	5	1	538	-543	-11	5	5	687	-682	10	6	0	201	205
-8	4	7	299	-296	4	5	1	198	-117	5	5	5	389	-401	2	6	0	1004	997
-7	4	7	64	49	5	5	2	315	-318	-9	5	5	605	-600	12	6	0	248	251
-6	4	7	93	-75	-12	5	2	346	-335	-8	5	5	667	-662	4	6	0	936	933
-5	4	7	139	148	-11	5	2	-54	-16	-7	5	5	565	-554	-12	6	1	556	-557
-4	4	7	124	116	8	5	1	436	-438	9	5	5	57	-57	14	6	0	266	270
-3	4	7	53	1	-10	5	2	310	-308	-6	5	5	821	-820	-11	6	1	422	422
-2	4	7	59	42	9	5</													

Table 3. Continued.

H	K	L	F(O)	F(C)	H	K	L	F(O)	F(C)	H	K	L	F(O)	F(C)	H	K	L	F(O)	F(C)
3	6	5	616	-626	-11	7	1	236	-241	-2	7	5	520	-517	-5	8	1	393	-386
4	6	5	474	-469	-10	7	1	424	-418	-1	7	5	488	-485	-4	8	1	115	-106
5	6	5	-54	58	-9	7	1	62	-58	0	7	5	405	-395	-3	8	1	304	-300
6	6	5	134	-137	-8	7	1	300	-309	1	7	5	166	-158	-2	8	1	60	-15
7	6	5	123	-129	-7	7	1	479	-472	2	7	5	183	-176	-1	8	1	436	-425
8	6	5	-51	-52	-6	7	1	322	-322	3	7	5	456	-459	-2	8	1	622	-574
9	6	5	-57	-57	-5	7	1	266	-252	4	7	5	397	-391	1	8	1	562	-557
-13	6	6	245	-187	-4	7	1	609	-594	5	7	5	188	-183	2	8	1	310	-315
-12	6	6	213	-221	-3	7	1	777	-761	6	7	5	296	-293	3	8	1	616	-632
-11	6	6	316	-321	-2	7	1	627	-622	7	7	5	143	-145	4	8	1	169	-162
-10	6	6	353	-357	-1	7	1	511	-506	8	7	5	251	-259	5	8	1	180	-169
-9	6	6	739	-745	0	7	1	267	-264	-13	7	6	519	-503	6	8	1	232	-231
-8	6	6	190	-187	1	7	1	455	-465	-12	7	6	270	-271	7	8	1	57	-45
-7	6	6	216	-226	2	7	1	291	-298	-11	7	6	428	-425	8	8	1	468	-462
-6	6	6	334	-329	3	7	1	131	-144	-10	7	6	277	-279	9	8	1	425	-423
-5	6	6	296	-244	4	7	1	656	-656	-9	7	6	166	-160	5	8	1	206	-195
-4	6	6	58	-57	5	7	1	437	-437	-8	7	6	317	-310	10	8	1	147	-145
-3	6	6	542	-573	6	7	1	379	-373	-7	7	6	421	-421	-12	8	2	134	-129
-2	6	6	457	-457	7	7	1	340	-339	-6	7	6	347	-336	-11	8	2	290	-291
-1	6	6	418	-433	8	7	1	249	-252	-5	7	6	676	-674	-10	8	2	301	-296
0	6	6	436	-458	9	7	1	415	-424	-4	7	6	718	-709	-8	8	2	225	-215
1	6	6	379	-314	10	7	1	130	-123	-3	7	6	242	-244	-8	8	2	147	-155
2	6	6	159	-160	11	7	1	170	-164	-2	7	6	55	-65	-7	8	2	402	-405
3	6	6	625	-614	-13	7	1	153	-147	-1	7	6	685	-674	-8	8	2	161	-147
4	6	6	-54	-15	-12	7	1	311	-307	0	7	6	119	-111	-5	8	2	420	-432
5	6	6	453	-453	-11	7	1	507	-507	1	7	6	248	-248	-4	8	2	302	-300
6	6	6	419	-426	-10	7	1	268	-276	2	7	6	740	-734	3	8	2	129	-133
7	6	6	105	-106	-9	7	1	260	-259	3	7	6	237	-219	4	8	2	316	-319
-13	6	7	-60	-7	-8	7	2	253	-240	4	7	6	513	-510	-1	8	2	284	-293
-12	6	7	176	-173	-7	7	2	390	-395	5	7	6	324	-315	0	8	2	110	-107
-11	6	7	108	-154	-6	7	2	650	-656	6	7	6	100	-107	1	8	2	553	-568
-10	6	7	166	-160	-5	7	2	527	-526	7	7	6	396	-390	2	8	2	276	-276
-9	6	7	36	81	-4	7	2	436	-438	-12	7	7	114	-123	3	8	2	266	-257
-8	6	7	328	327	-3	7	2	1468	-1454	-11	7	7	315	-310	4	8	2	466	-468
-7	6	7	284	286	-2	7	2	333	-333	-10	7	7	195	-197	5	8	2	241	-233
-6	6	7	119	123	-1	7	2	124	-124	-9	7	7	317	-315	6	8	2	210	-206
-5	6	7	101	-123	0	7	2	658	-657	-8	7	7	330	-329	7	8	2	287	-291
-4	6	7	50	-23	1	7	2	513	-505	-7	7	7	541	-551	8	8	2	118	-119
-3	6	7	579	573	2	7	2	650	-654	-6	7	7	377	-378	9	8	2	322	-326
-2	6	7	58	-87	3	7	2	547	-542	-5	7	7	259	-255	10	8	2	178	-178
-1	6	7	51	0	4	7	2	395	-403	-4	7	7	51	-51	-12	8	3	111	-126
0	6	7	526	531	5	7	2	762	-776	-3	7	7	416	-404	-11	8	3	445	-417
1	6	7	243	237	6	7	2	207	-203	-2	7	7	232	-230	-10	8	3	163	-157
2	6	7	349	-155	7	7	2	298	-256	-1	7	7	150	-159	-9	8	3	369	-369
3	6	7	81	-79	8	7	2	243	-241	0	7	7	447	-454	-8	8	3	139	-143
4	6	7	15	-15	9	7	2	411	-411	1	7	7	452	-453	9	8	3	236	-234
5	6	7	237	237	10	7	2	424	-419	2	7	7	317	-311	-10	8	3	207	-217
6	6	7	100	100	11	7	2	169	-165	3	7	7	193	-193	5	8	3	210	-205
-12	6	6	375	374	-13	7	3	311	-302	4	7	7	150	-158	-4	8	3	377	-376
-11	6	6	296	296	-12	7	3	140	-151	5	7	7	356	-349	3	8	3	762	-769
-10	6	6	48	56	-11	7	3	79	78	-11	7	7	317	-315	-8	8	3	245	-244
-9	6	6	282	-197	12	7	3	235	-242	-10	7	7	564	-556	-1	8	3	517	-520
-8	6	6	509	506	-9	7	3	344	-340	-9	7	7	207	-206	0	8	3	517	-515
-7	6	6	219	216	-8	7	3	171	-175	-8	7	7	450	-439	1	8	3	139	-143
-6	6	6	535	532	-7	7	3	518	-519	-7	7	7	502	-498	2	8	3	81	-73
-5	6	6	146	-52	6	7	3	459	-469	-6	7	7	319	-317	3	8	3	231	-228
-4	6	6	472	393	5	7	3	616	-620	-5	7	7	512	-511	4	8	3	471	-467
-3	6	6	424	428	-4	7	3	379	-315	-4	7	7	491	-500	5	8	3	227	-230
-2	6	6	52	10	-3	7	3	99	94	-3	7	7	415	-418	6	8	3	311	-306
-1	6	6	311	302	-2	7	3	465	-467	-2	7	7	556	-552	7	8	3	234	-226
0	6	6	515	-51	-1	7	3	463	-464	-1	7	7	225	-227	8	8	3	57	-47
1	6	6	191	21	2	7	3	571	-564	0	7	7	329	-322	9	8	3	112	-112
2	6	6	341	346	1	7	3	612	-613	1	7	7	267	-265	-12	8	4	158	-155
3	6	6	274	236	17	7	3	503	-503	2	7	7	255	-255	-11	8	4	133	-132
4	6	6	217	241	17	7	3	419	-414	3	7	8	317	-317	-10	8	4	116	-116
5	6	6	364	-54	4	7	3	110	-107	4	7	8	364	-363	5	8	4	400	-399
-11	6	7	138	-136	5	7	3	270	-275	-10	7	7	241	-238	-8	8	4	531	-537
-10	6	7	221	-227	6	7	3	320	-334	-9	7	7	302	-304	-7	8	4	99	-110
-9	6	7	57	-3	7	7	3	336	-344	-8	7	7	158	-159	-6	8	4	326	-324
-8	6	6	97	97	8	7	3	229	-221	-7	7	7	557	-550	5	8	4	171	-167
-7	6	6	69	-87	9	7	3	270	-267	-6	7	7	402	-407	-4	8	4	177	-191
-6	6	6	350	-359	16	7	3	282	-298	-5	7	9	309	-308	-3	8	4	182	-180
-5	6	6	214	-208	-13	7	4	175	-171	-4	7	9	261	-261	-2	8	4	253	-259
-4	6	6	52	-52	-12	7	4	253	-253	-3	7	9	169	-169	-3	8	4	356	-362
-3	6	6	329	-229	-10	7	4	434	-426	-2	7	9	330	-227	-2	8	4	404	-401
-2	6	6	514	101	-9	7	4	611	615	0	7	9	201	-182	2	8	4	177	-177
-1	6	6	55	57	-8	7	4	414	415	1	7	9	143	-151	3	8	4	218	-214
0	6	6	211	218	-7	7	4	452	447	2	7	9	232	-204	4	8	4	290	-286
1	6	6	274	278	-6	7	4	391	388	-7	7	9	241	-246	5	8	4	178	-163
2	6	6	181	176	-5	7	4	311	303	-7	7	9	396	-384	-7	8	4	301	-298
3	6	6	177	177	4	7	4	290	293	2	8	0	556	-558	-8	8	4	312	-303
4	6	6	105	105	5	7	4	293	293	3	8	0	151	-144	-6	8	4	132	-137
5	6	6	321	322	-9	7	4	305	-305	-12	8	1	343	-344	-4	8	5	135	-136
6	6	6	451	446	-8	7	4	462	-455	-11	8	1	185	-167	-5	8	5	165	-165
7	6	6	614	609	-7	7	5	75	75										

Table 3. Continued.

H	K	L	F(O)	F(C)	H	K	L	F(O)	F(C)	H	K	L	F(O)	F(C)	H	K	L	F(O)	F(C)	H	K	L	F(O)	F(C)	
2	9	2	519	-514	-6	9	8	592	595	7	10	4	205	201	-3	11	3	369	365	-3	12	3	469	476	
3	9	2	95	-108	-5	9	8	315	305	-10	10	5	421	-413	-2	11	3	100	101	-2	12	3	471	467	
4	9	2	-50	-10	-4	9	8	5	7	-10	10	5	215	-22	-1	11	3	261	256	-1	12	3	242	251	
5	9	2	591	-511	-3	9	8	320	334	-8	10	5	136	-126	-	11	3	148	149	-	12	3	108	139	
6	9	2	566	-559	-2	9	8	332	338	-7	10	5	235	-230	1	11	3	363	358	1	12	3	189	199	
7	9	2	184	-189	-1	9	8	149	146	-6	10	5	335	-325	2	11	3	135	-135	2	12	3	150	156	
8	9	2	323	-326	0	9	8	266	266	-5	10	5	133	-123	3	11	3	232	231	3	12	3	421	419	
9	9	2	243	-245	1	9	8	440	433	-4	10	5	400	-396	4	11	3	599	621	4	12	3	346	347	
-11	9	2	292	285	2	9	8	352	358	-3	10	5	514	-506	5	11	3	-57	41	5	12	3	350	351	
-10	9	2	426	436	-7	9	9	290	-154	-2	10	5	366	-355	6	11	3	81	68	-8	12	4	-59	42	
-9	9	2	128	130	-6	9	9	265	-259	-1	10	5	232	-235	7	11	3	278	289	-7	12	4	226	219	
-8	9	3	254	265	-5	9	9	162	-171	0	10	5	240	-242	-9	11	4	162	169	-6	12	4	-59	64	
-7	9	3	344	347	-4	9	9	200	-216	1	10	5	196	-140	-8	11	4	241	241	-4	12	4	-40	40	
-6	9	3	176	179	-3	9	9	264	-256	2	10	5	418	-415	11	11	4	141	141	4	12	4	217	225	
-5	9	3	565	566	-2	9	9	307	-305	3	10	5	237	-249	-6	11	4	110	111	-3	12	4	265	260	
-4	9	3	177	151	-1	9	9	174	-183	4	10	5	209	-210	-5	11	4	212	207	-2	12	4	92	-90	
-3	9	3	396	402	0	10	0	560	554	5	10	5	385	-392	4	11	4	167	173	-1	12	4	74	59	
-2	9	3	674	667	1	10	0	262	260	-10	10	5	57	-44	-3	11	4	362	359	0	12	4	174	184	
-1	9	3	68	49	2	10	0	176	-177	-9	10	5	147	-142	-2	11	4	481	489	1	12	4	88	79	
0	9	3	175	171	3	10	0	161	161	-8	10	5	328	-335	-1	11	4	294	290	2	12	4	158	150	
1	9	2	370	368	4	10	0	436	440	-10	10	5	144	-144	0	11	4	349	355	3	12	4	-57	-36	
2	9	2	335	335	5	10	0	67	-27	-6	10	5	240	-242	4	11	4	242	244	4	12	4	120	120	
3	9	2	679	679	6	10	0	1	10	-1	10	5	112	-115	-5	11	4	323	324	-1	12	5	333	-331	
4	9	3	319	317	7	10	0	79	83	-4	10	5	362	-353	3	11	4	89	88	-6	12	5	302	311	
5	9	3	314	310	8	10	0	398	411	-3	10	5	170	-173	4	11	4	258	255	-5	12	5	211	-211	
6	9	3	313	383	9	10	0	249	238	-2	10	5	179	-179	5	11	4	287	283	-4	12	5	292	-192	
7	9	3	99	108	10	10	0	82	-76	-1	10	5	299	-303	-9	11	5	96	-90	-3	12	5	118	-126	
8	9	3	86	87	-10	10	1	456	-441	0	10	6	271	-266	-8	11	5	234	-239	-2	12	5	423	-426	
9	9	2	165	166	-9	10	1	383	-381	1	10	6	121	-113	-7	11	5	336	-347	-1	12	5	453	-459	
-11	9	4	-555	28	-8	10	1	259	-258	2	10	5	54	-8	-6	11	5	272	-270	0	12	5	306	-309	
-10	9	4	393	356	-7	10	1	167	-160	3	10	5	57	-36	-5	11	5	153	-158	1	12	5	413	-425	
-9	9	4	229	224	-6	10	1	260	-262	4	10	5	364	-342	-4	11	5	253	-257	-2	12	5	257	-255	
-8	9	4	136	135	-5	10	1	146	-153	-9	10	5	255	-255	2	11	5	247	-251	6	12	5	136	-134	
-7	9	4	257	257	-4	10	1	103	-87	-8	10	5	254	-253	-2	11	5	471	-477	-5	12	6	-55	17	
-6	9	4	676	558	-3	10	1	425	-428	-7	10	5	367	-368	1	11	5	54	35	-6	12	6	8	76	
-5	9	4	698	705	-2	10	1	511	-514	-6	10	5	247	-243	0	11	5	242	-248	-3	12	6	96	-95	
-4	9	4	514	556	-1	10	1	627	-628	-5	10	5	315	-326	1	11	5	391	-397	-2	12	6	100	-167	
-3	9	4	263	265	0	10	1	356	-362	-4	10	5	59	-58	2	11	5	252	-254	-1	12	6	122	-107	
-2	9	4	78	85	1	10	1	500	-520	-3	10	7	69	-65	3	11	5	77	-79	0	12	6	-60	2	
-1	9	4	427	428	2	10	1	345	-356	-2	10	7	386	-391	-4	11	5	134	-132	0	13	0	-83	-33	
0	9	4	286	286	3	10	1	404	-404	-1	10	7	413	-404	-8	11	6	235	-234	1	13	0	217	216	
1	9	4	54	54	4	10	1	220	-218	0	10	7	348	-347	-7	11	6	207	-207	0	12	6	207	203	
2	9	4	771	784	5	10	1	394	-389	1	10	7	237	-239	-6	11	6	216	-215	3	13	0	141	149	
3	9	4	631	631	6	10	1	434	-438	2	10	7	112	-108	-5	11	6	342	-343	4	13	0	351	355	
4	9	4	295	312	7	10	1	172	-169	-7	10	8	289	-297	-6	11	6	121	-103	5	13	0	120	114	
5	9	4	178	175	8	10	1	195	-191	-6	10	8	117	-117	-3	11	6	135	-141	6	13	0	-57	17	
6	9	4	239	240	9	10	1	65	-73	-5	10	8	55	-58	-2	11	6	186	-177	-6	13	1	129	-116	
7	9	4	273	274	-10	10	2	151	-153	-4	10	8	82	-81	-1	11	6	140	-143	-5	13	1	255	-256	
8	9	4	244	244	-9	10	2	54	-36	-3	10	8	182	-173	0	11	6	354	-357	-4	13	1	176	-175	
-11	9	5	136	139	-8	10	2	126	-102	-2	10	8	251	-244	244	1	11	6	409	-409	-3	13	1	195	-193
-10	9	5	117	128	-7	10	2	234	-231	0	10	8	153	-150	-2	11	7	291	-291	-2	13	1	241	-240	
-9	9	5	420	420	-6	10	2	366	-362	-1	10	8	148	-146	-6	11	7	106	-105	-1	12	7	239	-240	
-8	9	5	349	349	-5	10	2	315	-315	0	10	8	502	-499	-6	11	7	242	-249	0	12	7	-18	-192	
-7	9	5	74	97	-4	10	2	119	120	1	11	0	376	371	-5	11	7	292	279	1	13	1	152	-149	
-6	9	5	428	430	-3	10	2	659	-662	2	11	0	226	239	-4	11	7	163	170	2	13	1	225	-227	
-5	9	5	448	452	-2	10	2	283	-281	3	11	0	160	153	-3	11	7	111	103	3	13	1	226	-226	
-4	9	5	256	241	-1	10	2	173	-176	4	11	0	112	125	-2	11	7	266	264	4	13	1	209	-221	
-3	9	5	103	103	-1	10	2	56	-55	5	11	0	199	198	-1	11	7	346	345	5	13	1	248	-249	
-2	9	5	369	366	1	10	2	223	-222	6	11	0	397	398	0	11	7	93	92	-6	12	5	-57	-56	
-1	9	5	422	425	-4	10	2	523	-534	-1	10	2	237	-231	-5	11	7	232	235	-3	13	1	242	-242	
0	9	5	354	355	-3	10	2	323	-322	-7	11	2	111	-104	-2	12	1	464	-462	-2	13	2	224	-224	
-6	9	6	70	72	-14	10	3	586	599	2	11	2	398	-404	-2	12	1	94	-96	-2	13	3	137	142	
-5	9	6	160	161	-13	10	3	175	164	3	11	2	575	-582	-1	12	1	268	-264	-1	13	3	253	249	
-4	9	6	470	472	-2	10	3	253	-253	4	11	2	136	-119	0	12	1	408	-399	0	13	3	264	265	
-3	9	6	554	554	-1	10	3	359	356	5	11	2	64	-53	1	12	1	386	-388	1	13	2	221	224	
-2	9	6	362	363	0	10	3	582	587	6	11	2	290	-294	-2	12	1	459	-467	2	13	3	225	221	
-1	9	6	342	343	1	10	3	247	300	7	11	2	350	-356	-3</										

## RESULTS

Fig. 1. is a stereoscopic view of the hexathionate ion as seen along an approximate twofold axis, and Fig. 2 is a view normal to this axis, with principal bond lengths and angles as calculated from the atomic coordinates of Table 1.

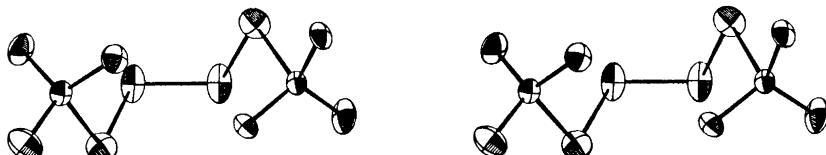


Fig. 1. The *cis-cis* form of the hexathionate ion in  $K_2Ba(S_6O_6)_2$  as seen along an approximate twofold axis. The ellipsoids represent 50 % probability.

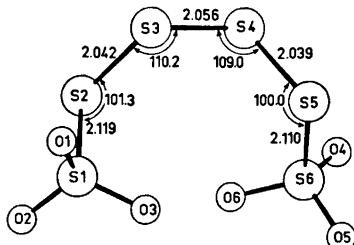


Fig. 2. The hexathionate ion as seen normal to an approximate twofold axis.

The differences between the dimensions of the hexathionate ion listed in Table 4, and those arrived at by the refinement based on two-dimensional film data,<sup>1</sup> are within the errors estimated for the latter. The standard deviations given in parentheses in Table 4 are about one tenth of those estimated for the film data.

The terminal S–S bonds, 2.119 and 2.110 Å, are between sulphonate sulphur atoms and divalent sulphur atoms, and are considerably longer than the three middle S–S bonds, 2.041, 2.056 and 2.039 Å, which are all between two divalent sulphur atoms. Of these three middle bonds the central one is 0.015–0.017 Å longer than the two others.

The dimensions of the hexathionate ion in the *cis-cis* form in the present salt will be further discussed in the next paper<sup>3</sup> and there compared to those of the hexathionate ion in the *trans-trans* form.

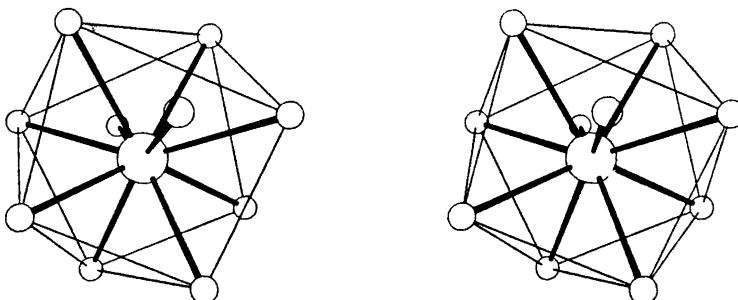
The packing in the crystal has been described in detail earlier,<sup>1</sup> and only a few aspects will be repeated here in order to present the redetermined dimensions.

Each barium ion, situated on a twofold axis, has ten oxygen neighbours within 2.960 Å. Eight of these are in a distorted square antiprism arrangement with Ba–O distances of 2.704(2)–2.894(2) Å. The O–O edges of the square faces are in the range 3.243(3)–4.117(3) Å, and the O–O edges between the square faces are in the range 3.007(2)–3.455(3) Å. The O–O–O angles of

*Table 4.* Dimensions of the *cis-cis* form of the hexathionate ion. Standard deviations, given in parentheses, include estimated uncertainties in unit cell dimensions.

Bond lengths and angles	
$S(1) - S(2) = 2.1186(11)$ Å	$\angle S(1) - S(2) - S(3) = 101.25(5)^\circ$
$S(2) - S(3) = 2.0415(14)$	$\angle S(2) - S(3) - S(4) = 110.20(5)^\circ$
$S(3) - S(4) = 2.0561(13)$	$\angle S(3) - S(4) - S(5) = 108.99(5)^\circ$
$S(4) - S(5) = 2.0394(16)$	$\angle S(4) - S(5) - S(6) = 99.95(5)^\circ$
$S(5) - S(6) = 2.1098(11)$	
$S(1) - O(1) = 1.443(2)$ Å	$S(6) - O(4) = 1.435(2)$ Å
$S(1) - O(2) = 1.441(2)$	$S(6) - O(5) = 1.459(2)$
$S(1) - O(3) = 1.448(3)$	$S(6) - O(6) = 1.449(2)$
$\angle S(2) - S(1) - O(1) = 105.1(1)^\circ$	$\angle S(5) - S(6) - O(4) = 107.3(1)^\circ$
$\angle S(2) - S(1) - O(2) = 100.7(1)^\circ$	$\angle S(5) - S(6) - O(5) = 100.6(1)^\circ$
$\angle S(2) - S(1) - O(3) = 107.4(1)^\circ$	$\angle S(5) - S(6) - O(6) = 107.5(1)^\circ$
$\angle O(1) - S(1) - O(2) = 115.5(2)^\circ$	$\angle O(4) - S(6) - O(5) = 115.8(2)^\circ$
$\angle O(1) - S(1) - O(3) = 111.6(2)^\circ$	$\angle O(4) - S(6) - O(6) = 112.8(2)^\circ$
$\angle O(2) - S(1) - O(3) = 115.0(2)^\circ$	$\angle O(5) - S(6) - O(6) = 111.6(2)^\circ$
Dihedral angles	
$S(1)S(2)S(3)/S(2)S(3)S(4) = 109.4^\circ$	$S(3)S(4)S(5)/S(4)S(5)S(6) = 106.3^\circ$
$S(2)S(3)S(4)/S(3)S(4)S(5) = 89.0^\circ$	
$S(3)S(2)S(1)/S(2)S(1)O(1) = 52.6^\circ$	$S(4)S(5)S(6)/S(5)S(6)O(4) = 66.5^\circ$
$S(3)S(2)S(1)/S(2)S(1)O(2) = 173.0^\circ$	$S(4)S(5)S(6)/S(5)S(6)O(5) = 172.0^\circ$
$S(3)S(2)S(1)/S(2)S(1)O(3) = 66.4^\circ$	$S(4)S(5)S(6)/S(5)S(6)O(6) = 55.1^\circ$
$S(2)S(1)O(1)/S(2)S(1)O(2) = 120.4^\circ$	$S(5)S(6)O(4)/S(5)S(6)O(5) = 121.5^\circ$
$S(2)S(1)O(1)/S(2)S(1)O(3) = 119.0^\circ$	$S(5)S(6)O(4)/S(5)S(6)O(6) = 121.6^\circ$
$S(2)S(1)O(2)/S(2)S(1)O(3) = 120.6^\circ$	$S(5)S(6)O(5)/S(5)S(6)O(6) = 116.9^\circ$
Non-bonded distances	
$S(1) - S(4) = 4.5525(11)$ Å	$S(3) - S(6) = 4.4505(13)$ Å
$S(1) - S(5) = 4.7099(13)$	$S(2) - S(6) = 4.6100(14)$

the faces are  $78.05(6)^\circ$ ,  $87.51(7)^\circ$ ,  $94.53(8)^\circ$ , and  $98.52(7)^\circ$ . The two oxygen atoms not involved in the antiprism arrangement are at  $2.960(2)$  Å from barium, and lie outside the square faces with distances to the corners in the range  $2.406(3) - 3.338(2)$  Å. Fig. 3 is a stereoscopic view of the Ba – O coordination.



*Fig. 3.* A stereoscopic view where the barium-oxygen coordination is indicated by the thick lines, and the thin lines are between the oxygen atoms of the square antiprism.

Each potassium ion is surrounded by six oxygen atoms in much distorted octahedral arrangements. The distances from K(1), situated in a symmetry centre, to the oxygen atoms are 2.664(2), 2.781(2), and 3.008(3) Å. The *trans* O—K(1)—O angles are 180.0° and the *cis* O—K(1)—O angles are in the range 48.8(1)°–131.2(1)°. The distances from K(2), situated on a twofold axis, to the oxygen atoms are 2.668(2), 2.840(2), and 3.028(3) Å. The *trans* O—K(2)—O angles are in the range 132.6(1)°–159.0(1)° and the *cis* O—K(2)—O angles are in the range 48.0(1)°–125.6(1)°.

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