

The Crystal and Molecular Structure of 3,4-T trimethylene-6a-selenaselenophthene

ASBJØRN HORDVIK and JAN A. PORTEN

Chemical Institute, University of Bergen, N-5000 Bergen, Norway

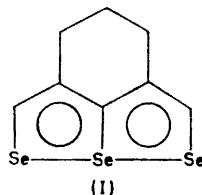
3,4-T trimethylene-6a-selenaselenophthene crystallizes in the orthorhombic space group $Pn2_1a$, with unit cell dimensions: $a = 11.437(3)$ Å, $b = 4.860(5)$ Å, and $c = 16.589(4)$ Å. There are four molecules per unit cell.

The structure has been solved by three-dimensional Patterson synthesis, and refined by full-matrix least squares. The refinement comprises 1004 $h0l-h4l$ reflections, including 101 unobserved.

The atoms of the 6a-selenaselenophthene system and the methylene-carbons C(6) and C(8), bonded to C(3) and C(4), respectively, lie almost in the same plane. The Se-Se bond lengths are Se(1)–Se(6a) = 2.568(3) Å and Se(6a)–Se(6) = 2.554(3) Å, with the angle Se(1)–Se(6a)–Se(6) = 175.2(2)°. The other bond lengths in the selenaselenophthene system are: Se(1)–C(2) = 1.777(23) Å, Se(6a)–C(3a) = 1.906(20) Å, Se(6)–C(5) = 1.817(22) Å, C(2)–C(3) = 1.379(24) Å, C(3)–C(3a) = 1.435(23) Å, C(3a)–C(4) = 1.418(22) Å, and C(4)–C(5) = 1.400(23) Å.

In the crystals, Se(1) of the reference molecule approaches Se(6a) of a screw axis-related molecule at a distance of 3.57 Å.

The present structure investigation of 3,4-trimethylene-6a-selenaselenophthene (I) has been carried out in order to obtain further experimental evidence for the bonding in 6a-selenaselenophthenes.



STRUCTURE DETERMINATION

Crystals of 3,4-trimethylene-6a-selenaselenophthene (I) were generously supplied by Reid.¹ The crystals are very dark purple and belong to the orthorhombic space group $Pn2_1a$.

The structure study is based on photographic data, taken with Weissenberg camera and $\text{CuK}\alpha$ radiation. The data comprise 1004 $h0l-h4l$ reflections, including 101 unobserved.

Approximate selenium positions were found from a three-dimensional Patterson map, and the carbon atoms revealed themselves during a subsequent Fourier synthesis.

The structure was refined by a full-matrix least squares procedure (*cf.* Ref. 2). The constants a_1 and a_2 in the weighting scheme were in the present case set equal to 1.0. Unobserved reflections with $K|F_c|$ greater than $F_{\text{threshold}}$ were included in the refinement with $F_o = F_{\text{threshold}}$. Anisotropic temperature factors were applied to selenium and carbon, and isotropic to hydrogen. Four low order reflections, supposed to be affected by secondary extinction, were excluded from the least squares refinement. The final R factor is 0.076 when unobserved reflections are included, and 0.075 when they are omitted. For further details with respect to the structure determination, see Experimental.

DISCUSSION

Molecular shape and dimensions. Bond lengths and angles in the 3,4-trimethylene-6a-selenaselenophthene molecule, together with their standard deviations, are listed in Tables 1 and 2, and shown in Fig. 1, a and b.

The molecule is presented in Fig. 1, a and b, in a projection on to the least squares plane of the atoms of the 6a-selenaselenophthene system and the methylene carbons C(6) and C(8). The equation for this plane, with weights equal to four for selenium, and one for carbon, is,

$$-0.63748 X + 0.76127 Y - 0.11871 Z = -0.78216$$

with X , Y , and Z in Å units. Deviations from the plane for the selenium and the carbon atoms are given in Fig. 1a. It is seen that the methylene carbons C(6) and C(8) and the atoms of the 6a-selenaselenophthene system lie almost in the same plane.

Table 1. Bond lengths (l) and standard deviations in bond lengths $\sigma(l)$ in 3,4-trimethylene-6a-selenaselenophthene. Bond lengths (l') include corrections for rigid-body libration.

Bond	l' (Å)	l (Å)	$\sigma(l)$ (Å)
Se(1) – Se(6a)	2.568	2.563	0.003
Se(6a) – Se(6)	2.554	2.548	0.003
Se(1) – C(2)	1.777	1.771	0.023
Se(6a) – C(3a)	1.906	1.899	0.020
Se(6) – C(5)	1.817	1.811	0.022
C(2) – C(3)	1.379	1.375	0.024
C(3) – C(3a)	1.435	1.432	0.023
C(3a) – C(4)	1.418	1.414	0.022
C(4) – C(5)	1.400	1.397	0.023
C(3) – C(6)	1.499	1.494	0.028
C(6) – C(7)	1.507	1.503	0.024
C(7) – C(8)	1.564	1.561	0.024
C(8) – C(4)	1.549	1.544	0.030

Table 2. Bond angles $\angle(ijk)$ in 3,4-trimethylene-6a-selenaselenophthene. The standard deviations given in parentheses refer to the last digits of the respective values.

i	j	k	$\angle(ijk)^\circ$
Se(6a)	Se(1)	C(2)	87.7(6)
Se(1)	Se(6a)	C(3a)	87.5(5)
Se(1)	Se(6a)	Se(6)	175.2(2)
Se(6)	Se(6a)	C(3a)	88.0(5)
Se(6a)	Se(6)	C(5)	89.0(5)
Se(1)	C(2)	C(3)	126.7(1.5)
C(2)	C(3)	C(3a)	118.6(1.7)
C(2)	C(3)	C(6)	121.5(1.5)
C(6)	C(3)	C(3a)	119.9(1.4)
Se(6a)	C(3a)	C(3)	119.5(1.2)
Se(6a)	C(3a)	C(4)	119.1(1.3)
C(3)	C(3a)	C(4)	121.4(1.7)
C(3a)	C(4)	C(5)	121.4(1.7)
C(3a)	C(4)	C(8)	119.6(1.4)
C(5)	C(4)	C(8)	118.9(1.4)
Se(6)	C(5)	C(4)	122.5(1.3)
C(3)	C(6)	C(7)	108.8(1.9)
C(6)	C(7)	C(8)	109.8(1.5)
C(4)	C(8)	C(7)	105.8(1.9)

Comparison with related molecules. Fig. 1a shows that the 3,4-trimethylene-6a-selenaselenophthene molecule is nearly symmetric about a plane perpendicular to the molecular plane and passing through Se(6a), C(3a), and C(7).

If one takes the average of corresponding bond lengths (l) in the two halves of 3,4-trimethylene-6a-selenaselenophthene, one gets the values given in Ia.

There are no significant deviations between the C-C and C-Se bond lengths in structure Ia and the corresponding C-C and C-Se bond lengths in the average structure of 6a-selenaselenophthene (II).² However, the Se-Se bonds in Ia, 2.556(3) Å, are significantly shorter than the Se-Se bonds in II, 2.583(3) Å, and the shortening of the Se-Se bonds seems to be due to the presence of the 3,4-trimethylene bridge.

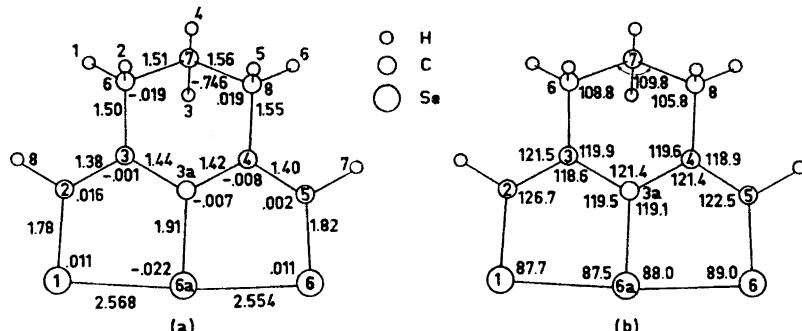
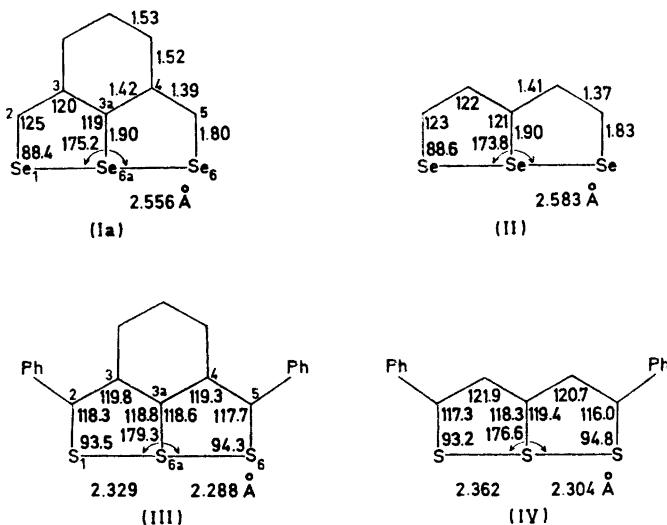


Fig. 1. (a) Bond lengths (Å) in the 3,4-trimethylene-6a-selenaselenophthene molecule, and atomic distances (Å) from the least squares plane of the atoms of the 6a-selenaselenophthene system and carbons C(6) and C(8). (b) Bond angles (°).

Support for this idea derives from a comparison of the structures Ia and II with the structures of 2,5-diphenyl-3,4-trimethylene-6a-thiathiphene (III)³ and 2,5-diphenyl-6a-thiathiphene (IV).⁴ The sum of the Se-Se bond lengths in Ia, 5.112 Å, is 0.054 Å shorter than the sum of the Se-Se bond lengths



in II, 5.166 Å, and similarly the sum of the S-S bond lengths in III, 4.617 Å, is 0.049 Å shorter than the sum of the S-S bond lengths in IV, 4.666 Å.

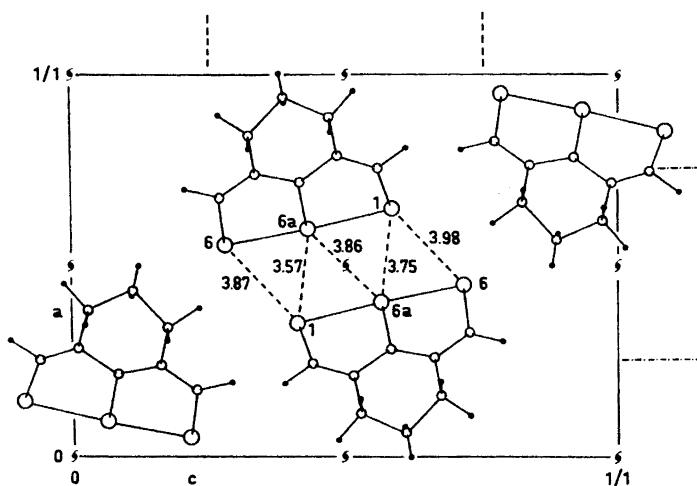


Fig. 2. The arrangement of 3,4-trimethylene-6a-selenaselenophthene molecules in the crystal as seen along the *b*-axis. Atomic distances are given in Å units.

A comparison of bond angles in Ia and II shows that the bond angle at Se(1) is almost the same in the two compounds, while the bond angle at C(2) is greater, and the bond angles at C(3) and C(3a) are smaller in Ia than in II. Similar equalities and differences in bond angles are found for structures III and IV.

The arrangement of 3,4-trimethylene-6a-selenaselenophthene molecules in the unit cell, as seen along the *b* axis, is shown in Fig. 2. Intermolecular atomic distances shorter than corresponding van der Waals distances are given in the figure.

Table 3. Atomic coordinates in fractions of corresponding cell edges. The standard deviations given in parentheses refer to the last digits of the respective values.

Atom	<i>x</i>	<i>y</i>	<i>z</i>
Se(1)	0.14332(17)	0.02769(87)	-0.08708(9)
Se(6a)	0.09320(13)	0.00000	0.06329(9)
Se(6)	0.05284(18)	0.00969(90)	0.2143(11)
C(2)	0.2553(17)	0.2633(55)	-0.0613(11)
C(3)	0.2832(13)	0.3543(43)	0.0148(8)
C(3a)	0.2159(16)	0.2557(47)	0.0816(10)
C(4)	0.2392(14)	0.3435(43)	0.1613(8)
C(5)	0.1726(16)	0.2498(52)	0.2264(9)
C(6)	0.3816(15)	0.5508(56)	0.0289(11)
C(7)	0.4323(15)	0.4980(72)	0.1110(9)
C(8)	0.3378(15)	0.5536(64)	0.1770(10)
H(1)	0.454	0.499	-0.010
H(2)	0.348	0.737	0.025
H(3)	0.430	0.265	0.113
H(4)	0.496	0.520	0.123
H(5)	0.303	0.728	0.176
H(6)	0.384	0.487	0.231
H(7)	0.196	0.317	0.288
H(8)	0.308	0.329	-0.110

Table 4. Temperature parameters U_{ij} (\AA^2) for selenium and carbon. The expression used is $\exp[-2\pi^2(h^2a^{*2}U_{11} + \dots + 2hka^*b^*U_{12} + \dots)]$. All values are multiplied by 10^4 . Standard deviations in parentheses refer to the last digits of the respective values. Isotropic temperature factors $\exp[-8\pi^2U(\sin^2\theta/\lambda^2)]$ with $U=0.0375 \text{ \AA}^2$ were used for the hydrogen atoms.

	U_{11}	U_{22}	U_{33}	U_{12}	U_{23}	U_{13}
Se(1)	446(9)	432(15)	271(6)	82(16)	-92(14)	-68(6)
Se(6a)	300(6)	232(12)	340(7)	-20(12)	-7(12)	-13(6)
Se(6)	500(10)	437(15)	373(8)	-11(18)	28(19)	150(7)
C(2)	415(95)	343(146)	340(95)	133(103)	-19(101)	33(79)
C(3)	299(68)	281(116)	91(56)	32(72)	24(63)	-21(54)
C(3a)	475(94)	74(114)	212(75)	-75(90)	-53(77)	19(66)
C(4)	315(74)	203(119)	104(64)	-99(72)	-37(65)	-52(55)
C(5)	490(95)	402(135)	186(66)	34(100)	-36(84)	-8(72)
C(6)	340(72)	152(147)	484(81)	-49(89)	-8(102)	142(65)
C(7)	403(80)	599(176)	335(77)	-397(142)	252(133)	-58(65)
C(8)	418(86)	445(191)	350(71)	1(121)	-17(120)	76(67)

Table 5. Observed and calculated structure factors for 3,4-trimethylene-6a-selenaselenophene. The values given are ten times the absolute values. Unobserved reflections are marked with a minus sign.

H	K	L	F(O)	F(C)	H	K	L	F(O)	F(C)	H	K	L	F(O)	F(C)	H	K	L	F(O)	F(C)
-2	0	0	214	451	0	0	4	215	419	10	0	0	70	69	3	1	0	1030	1080
4	0	0	1151	1165	5	0	5	114	500	11	0	4	117	93	3	1	9	229	214
6	0	0	752	763	5	0	6	405	420	11	0	4	118	99	3	1	10	103	101
8	0	0	131	114	5	0	7	603	711	11	0	5	916	505	3	1	11	224	230
10	0	0	370	341	5	0	6	1033	1159	11	0	4	493	508	3	1	12	147	136
12	0	0	61	63	5	0	6	502	493	11	0	5	104	77	3	1	13	491	505
14	0	0	64	67	5	0	5	505	506	11	0	6	104	55	3	1	14	360	303
16	0	0	743	611	5	0	5	111	111	11	0	7	106	80	3	1	15	425	377
18	0	0	112	112	5	0	4	115	115	11	0	6	104	49	3	1	16	17	311
20	0	0	1553	1491	5	0	3	357	350	11	0	9	400	344	3	1	17	105	117
22	0	0	1313	1227	5	0	4	316	303	11	0	10	124	177	3	1	18	348	316
24	0	0	340	329	5	0	3	316	303	11	0	11	118	203	3	1	19	163	145
26	0	0	277	259	5	0	2	306	300	11	0	12	113	189	3	1	20	149	141
28	0	0	673	592	5	0	1	300	244	12	0	4	104	50	4	1	21	517	439
30	0	0	374	355	5	0	0	360	293	12	0	3	210	100	4	1	2	488	448
32	0	0	103	127	6	0	1	212	244	12	0	3	104	504	4	1	3	191	169
34	0	0	177	176	6	0	0	202	255	12	0	3	103	119	4	1	4	608	605
36	0	0	437	463	6	0	0	169	166	12	0	3	114	144	4	1	5	193	130
38	0	0	426	387	6	0	0	145	157	12	0	3	106	129	4	1	6	384	413
40	0	0	657	611	6	0	0	155	190	12	0	3	107	71	4	1	7	754	661
42	0	0	771	771	6	0	0	171	171	12	0	3	108	129	4	1	8	530	475
44	0	0	511	511	6	0	0	173	173	12	0	3	109	129	4	1	9	514	477
46	0	0	523	516	6	0	0	169	151	12	0	3	110	129	4	1	10	153	156
48	0	0	124	124	6	0	0	149	149	12	0	3	111	127	4	1	11	192	192
50	0	0	374	355	5	0	0	160	210	12	0	3	112	129	4	1	12	246	246
52	0	0	147	147	5	0	0	174	244	12	0	3	113	129	4	1	13	220	210
54	0	0	177	176	5	0	0	160	202	12	0	3	114	129	4	1	14	191	191
56	0	0	109	109	5	0	0	144	144	12	0	3	115	129	4	1	15	192	192
58	0	0	111	111	5	0	0	145	145	12	0	3	116	129	4	1	16	193	193
60	0	0	124	124	5	0	0	146	146	12	0	3	117	129	4	1	17	194	194
62	0	0	124	124	5	0	0	147	147	12	0	3	118	129	4	1	18	195	195
64	0	0	124	124	5	0	0	148	148	12	0	3	119	129	4	1	19	196	196
66	0	0	124	124	5	0	0	149	149	12	0	3	120	129	4	1	20	197	197
68	0	0	124	124	5	0	0	150	150	12	0	3	121	129	4	1	21	198	198
70	0	0	124	124	5	0	0	151	151	12	0	3	122	129	4	1	22	199	199
72	0	0	124	124	5	0	0	152	152	12	0	3	123	129	4	1	23	200	200
74	0	0	124	124	5	0	0	153	153	12	0	3	124	129	4	1	24	201	201
76	0	0	124	124	5	0	0	154	154	12	0	3	125	129	4	1	25	202	202
78	0	0	124	124	5	0	0	155	155	12	0	3	126	129	4	1	26	203	203
80	0	0	124	124	5	0	0	156	156	12	0	3	127	129	4	1	27	204	204
82	0	0	124	124	5	0	0	157	157	12	0	3	128	129	4	1	28	205	205
84	0	0	124	124	5	0	0	158	158	12	0	3	129	129	4	1	29	206	206
86	0	0	124	124	5	0	0	159	159	12	0	3	130	129	4	1	30	207	207
88	0	0	124	124	5	0	0	160	160	12	0	3	131	129	4	1	31	208	208
90	0	0	124	124	5	0	0	161	161	12	0	3	132	129	4	1	32	209	209
92	0	0	124	124	5	0	0	162	162	12	0	3	133	129	4	1	33	210	210
94	0	0	124	124	5	0	0	163	163	12	0	3	134	129	4	1	34	211	211
96	0	0	124	124	5	0	0	164	164	12	0	3	135	129	4	1	35	212	212
98	0	0	124	124	5	0	0	165	165	12	0	3	136	129	4	1	36	213	213
100	0	0	124	124	5	0	0	166	166	12	0	3	137	129	4	1	37	214	214
102	0	0	124	124	5	0	0	167	167	12	0	3	138	129	4	1	38	215	215
104	0	0	124	124	5	0	0	168	168	12	0	3	139	129	4	1	39	216	216
106	0	0	124	124	5	0	0	169	169	12	0	3	140	129	4	1	40	217	217
108	0	0	124	124	5	0	0	170	170	12	0	3	141	129	4	1	41	218	218
110	0	0	124	124	5	0	0	171	171	12	0	3	142	129	4	1	42	219	219
112	0	0	124	124	5	0	0	172	172	12	0	3	143	129	4	1	43	220	220
114	0	0	124	124	5	0	0	173	173	12	0	3	144	129	4	1	44	221	221
116	0	0	124	124	5	0	0	174	174	12	0	3	145	129	4	1	45	222	222
118	0	0	124	124	5	0	0	175	175	12	0	3	146	129	4	1	46	223	223
120	0	0	124	124	5	0	0	176	176	12	0	3	147	129	4	1	47	224	224
122	0	0	124	124	5	0	0	177	177	12	0	3	148	129	4	1	48	225	225
124	0	0	124	124	5	0	0	178	178	12	0	3	149	129	4	1	49	226	226
126	0	0	124	124	5	0	0	179	179	12	0	3	150	129	4	1	50	227	227
128	0	0	124	124	5	0	0	180	180	12	0	3	151	129	4	1	51	228	228
130	0	0	124	124	5	0	0	181	181	12	0	3	152	129	4	1	52	229	229
132	0	0	124	124	5	0	0	182	182	12	0	3	153	129	4	1	53	230	230
134	0	0	124	124	5	0	0	183	183	12	0	3	154	129	4	1	54	231	231
136	0	0	124	124	5	0	0	184	184	12	0	3	155	129	4	1	55	232	232
138	0	0	124	124	5	0	0	185	185	12	0	3	156	129	4	1	56	233	233
140	0	0	124	124	5	0	0	186	186	12	0	3	157	129	4	1	57	234	234
142	0	0	124	124	5	0	0	187	187	12	0	3	158	129	4	1	58	235	235
144	0	0	124	124	5	0	0	188	188	12	0	3	159	129	4	1	59	236	236
146	0	0	124	124	5	0	0	189	189	12	0	3	160	129	4	1	60	237	237
148	0	0	124	124	5	0	0	190	190	12	0	3	161	129	4	1	61	238	238
150	0	0	124	124	5	0	0	191	191	12	0	3	162	129	4	1	62	239	239
152	0	0	124	124	5	0	0	192	192	12	0	3	163	129	4	1	63	240	240
154	0	0	124	124	5	0	0	193	193	12	0	3	164	129	4	1	64	241	241
156	0	0	124	124	5	0	0	194	194	12	0	3	165	129	4	1	65	242	242
158	0	0	124	124	5	0	0	195	195	12	0	3	166	129	4	1	66	243	243
160	0	0	124	124	5	0	0	196	196	12	0	3	167	129	4	1	67	244	244
162	0	0	124	124	5	0	0	197	197	12	0	3	168	129	4	1	68	245	245
164	0	0	124	124	5	0	0	198	198	12	0	3	169	129	4	1	69	246	246
166	0	0	124</																

Table 5. Continued.

H	K	L	F(0)	F(1)	H	K	L	F(0)	F(1)	H	K	L	F(0)	F(1)	H	K	L	F(0)	F(1)	
1	2	4	665	655	6	2	13	-65	85	0	3	15	945	855	7	3	1	365	372	
1	2	7	1410	1299	6	2	14	334	494	0	3	15	142	110	7	3	2	624	658	
1	2	8	417	367	6	2	15	405	446	0	3	17	102	79	7	3	3	255	252	
1	2	9	151	141	6	2	16	233	236	1	3	4	595	861	7	3	4	308	376	
1	2	10	217	249	0	2	17	97	88	1	3	4	547	409	7	3	5	247	371	
1	2	11	145	145	6	2	18	207	340	1	3	5	54	30	7	3	6	267	308	
1	2	12	421	393	7	2	1	54	79	1	3	5	365	341	7	3	7	152	143	
1	2	13	471	466	7	2	2	206	195	1	3	5	365	341	7	3	8	503	545	
1	2	14	610	622	7	2	3	149	119	1	3	6	679	867	7	3	9	494	492	
1	2	15	380	353	7	2	4	795	631	1	3	7	705	697	7	3	10	215	212	
1	2	16	195	185	7	2	5	629	595	1	3	8	234	255	7	3	11	213	14	
1	2	17	123	153	7	2	6	11	102	1	3	9	187	178	7	3	12	271	262	
1	2	18	121	121	7	2	7	172	226	1	3	10	205	159	7	3	13	506	518	
1	2	19	119	149	7	2	8	170	140	1	3	11	234	62	7	3	14	166	167	
1	2	20	249	263	7	2	9	66	70	1	3	12	405	481	7	3	15	107	76	
1	2	21	342	455	7	2	10	343	350	1	3	13	403	392	8	3	1	342	339	
1	2	22	557	549	7	2	11	625	584	1	3	14	619	588	8	3	2	308	291	
1	2	23	506	446	7	2	12	383	70	1	3	15	169	142	8	3	3	62	75	
1	2	24	563	543	7	2	13	-63	45	1	3	17	668	45	8	3	4	62	303	
1	2	25	374	363	7	2	14	120	113	1	3	17	136	143	8	3	5	340	303	
1	2	26	620	620	7	2	15	215	206	1	3	18	600	571	8	3	6	196	194	
1	2	27	432	383	7	2	16	111	109	2	3	2	414	392	8	3	7	111	100	
1	2	28	545	257	7	2	17	261	280	2	3	3	62	47	8	3	8	207	248	
2	1	10	29	39	6	2	1	123	269	2	3	3	4	202	173	8	3	9	205	243
2	1	11	103	103	6	2	2	44	436	2	3	3	5	202	174	8	3	10	446	418
2	1	12	307	351	6	2	3	705	346	2	3	3	5	202	174	8	3	11	134	130
2	1	13	516	516	8	2	4	270	260	2	3	3	7	149	142	8	3	12	110	111
2	1	14	472	474	8	2	5	544	480	2	3	8	604	727	8	3	13	145	125	
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2	1	19	375	400	8	2	10	93	93	2	3	13	378	376	9	3	18	441	424	
2	1	20	578	562	8	2	11	501	476	2	3	14	490	453	9	3	19	111	109	
2	1	21	808	745	8	2	12	338	319	2	3	15	192	170	9	3	20	126	106	
2	1	22	344	366	8	2	13	145	154	2	3	16	359	132	9	3	21	132	140	
2	1	23	147	147	8	2	14	125	121	2	3	17	72	72	9	3	22	204	194	
2	1	24	160	159	9	2	15	103	105	2	3	18	241	172	9	3	23	255	255	
2	1	25	134	147	9	2	16	521	543	2	3	19	524	501	9	3	24	291	421	
2	1	26	1041	1041	9	2	17	111	111	2	3	20	519	510	9	3	25	204	197	
2	1	27	526	526	9	2	18	105	105	2	3	21	519	510	9	3	26	249	249	
2	1	28	121	121	9	2	19	105	105	2	3	22	519	510	9	3	27	136	136	
2	1	29	112	119	9	2	20	525	525	2	3	23	519	517	9	3	28	174	169	
2	1	30	236	261	9	2	21	127	138	3	3	7	982	1109	10	3	8	610	533	
2	1	31	134	151	9	2	22	436	496	3	3	9	615	624	10	3	9	128	108	
2	1	32	404	520	9	2	23	441	399	3	3	10	149	104	10	3	11	260	272	
2	1	33	523	523	9	2	24	103	370	3	3	11	220	223	10	3	12	61	59	
2	1	34	220	216	9	2	25	-62	17	3	3	12	90	214	10	3	13	605	605	
2	1	35	261	263	9	2	26	445	466	3	3	13	342	359	10	3	14	105	105	
2	1	36	256	256	9	2	27	525	525	3	3	14	342	348	10	3	15	105	105	
2	1	37	164	164	9	2	28	145	142	3	3	15	342	342	10	3	16	74	74	
2	1	38	113	112	9	2	29	404	274	3	3	16	145	104	10	3	17	61	59	
2	1	39	146	582	10	2	3	43	8	4	3	1	414	307	10	3	4	82	86	
2	1	40	217	213	10	2	4	54	495	4	3	2	480	170	11	3	5	105	132	
2	1	41	470	470	10	2	5	111	111	4	3	3	181	105	11	3	6	85	85	
2	1	42	805	900	10	2	6	7	60	4	3	4	455	462	11	3	7	95	84	
2	1	43	825	825	10	2	7	204	195	4	3	5	209	254	11	3	8	50	58	
2	1	44	77	77	10	2	8	9	152	4	3	6	255	245	11	3	9	88	88	
2	1	45	616	616	10	2	9	244	240	4	3	7	304	303	11	3	10	155	145	
2	1	46	129	132	10	2	10	-50	54	4	3	8	351	240	12	3	11	155	150	
2	1	47	146	143	10	2	11	56	59	4	3	9	108	705	12	3	12	259	270	
2	1	48	525	523	10	2	12	140	142	4	3	10	311	274	12	3	13	149	147	
2	1	49	474	474	10	2	13	64	81	4	3	11	374	396	12	3	14	89	74	
2	1	50	210	210	10	2	14	263	264	4	3	12	393	393	12	3	15	474	472	
2	1	51	153	153	11	2	1	124	156	4	3	13	401	401	12	3	16	503	503	
2	1	52	107	112	11	2	2	178	148	4	3	14	310	356	12	3	17	227	240	
2	1	53	100	144	11	2	3	97	98	5	3	1	245	197	10	4	0	126	141	
2	1	54	154	154	11	2	4	61	62	5	3	2	472	249	10	4	1	123	121	
2	1	55	764	764	11	2	5	29	37	5	3	3	272	245	10	4	2	111	126	
2	1	56	924	924	11	2	6	10	140	142	5	3	4	304	255	10	4	3	101	134
2	1	57	131	131	11	2	7	147	142	5	3	5	732	707	10	4	4	147	147	
2	1	58	282	284	11	2	8	64	68	5	3	6	556	410	10	4	5	89	89	
2	1	59	345	341	12	2	9	71	83	5	3	7	174	137	10	4	6	121	137	
2	1	60	518	547	12	2	10	455	445	5	3	8	345	345	10	4	7	477	477	
2	1	61	719	743	12	2	11	433	514	5	3	9	108	90	10	4	8	458	303	
2	1	62	600	1024	12	2	12	94	93	5	3	10	500	475	10	4	9	677	677	
2	1	63	539	447	12	2	13	151	443	5	3	11	500	475	10	4	10	130	130	
2	1	64	54	25	12	2	14	113	103	5	3	12	377	346	10	4	11	123	111	
2	1	65	-54	28	12	2	15	125	159	5	3	13	373	347	10	4	12	260	244	
2	1	66	176	196	12	2	16	-78	44	5	3	14	466	544	10	4	13	138	135	
2	1	67	389	389	12	2	17	30	460	5	3	15	305	305	10	4	14	260	244	
2	1	68	-53	12	2	18	200	1708	6	3	16	464	522	10	4	15	126	126		
2	1	69	372	409	6	3	0	248	240	6	3	7	88	102						

EXPERIMENTAL

The unit cell dimensions for crystals of 3,4-trimethylene-6a-selenaselenophthene were determined from high-order reflections on $h0l$ and $0kl$ Weissenberg photographs, where lead nitrate powder lines had been superimposed for reference. A least squares procedure on 41 measured 2θ -values gave $a = 11.437(3)$ Å, $b = 4.860(5)$ Å, and $c = 16.589(4)$ Å.

Four molecules per unit cell give a calculated density of 2.456 g/cm³ as compared with the density 2.45 g/cm³, found by flotation.

The intensities of the $h0l-h4l$ and $0kl$ reflections were estimated visually from Weissenberg photographs taken with Ni-filtered CuK α radiation ($\mu = 159.4$ cm⁻¹). $0kl$ reflections from the zero layer about a were used for scaling only.

Lp corrections and absorption corrections were applied, the latter according to a procedure of Coppens *et al.*⁵

The hydrogen positions were estimated and included in the structure factor calculations, but not refined.

The scattering factors used for selenium, carbon, and hydrogen in the structure factor calculations were those given in the International Tables.⁶ The selenium scattering curve was corrected for anomalous dispersion, using the $\Delta f'$ and $\Delta f''$ values given by Cromer.⁷

The final atomic coordinates are listed in Table 3, the temperature parameters in Table 4. The final list of structure factors is given in Table 5.

An analysis of the thermal parameters of the S and C atoms, assuming the molecule a rigid body was carried out according to the method of Schomaker and Trueblood.⁸ The Se—Se, Se—C, and C—C bond lengths have been corrected for rigid body libration according to Cruickshank's formula.⁹ The corrected bond lengths are listed in the first column of Table 1.

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REFERENCES

1. Jackson, M. G. and Reid, D. H. *Unpublished data*; Jackson, M. G., Ph. D. Thesis, University of St. Andrews 1973.
2. Hordvik, A. and Julshamn, K. *Acta Chem. Scand.* **25** (1971) 2507.
3. Birknes, B., Hordvik, A. and Sæthre, L. J. *Acta Chem. Scand.* **26** (1972) 2140.
4. Hordvik, A. *Acta Chem. Scand.* **25** (1971) 1583.
5. Coppens, P., Leiserowitz, L. and Rabinovich, D. *Acta Cryst.* **18** (1965) 1035.
6. *International Tables for X-Ray Crystallography*, Kynoch Press, Birmingham 1962, Vol. III, p. 202.
7. Cromer, D. T. *Acta Cryst.* **18** (1965) 17.
8. Schomaker, V. and Trueblood, K. N. *Acta Cryst.* **B 24** (1968) 63.
9. Cruickshank, D. W. J. *Acta Cryst.* **9** (1956) 757; **14** (1961) 896.

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