

Crystal Structure of DL-Tyrosine

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The crystal structure of DL-tyrosine has been determined by X-ray methods using 1180 observed reflections collected by counter methods. The crystals are orthorhombic, space group $Pna2_1$, with unit cell dimensions $a = 20.83$ Å, $b = 6.81$ Å, and $c = 5.90$ Å. The structure was solved by direct methods and refined to a conventional R -factor of 0.037; estimated standard deviations in bond lengths not involving hydrogen are 0.003 Å and in angles 0.2°. The molecular dimensions are found to be nearly identical with those found for L-tyrosine. The difference in crystal packing between the L and the DL form is discussed.

The preferred conformation of tyrosine is of interest in several biochemical relations and has frequently been discussed. Information on the structure of the molecule in solution is difficult to obtain; one approach to the problem is to examine the molecular structure in various crystal modifications in order to study the influence of the environment on the conformation. For an optically active compound crystals of the active and racemic forms may both be examined.

The conformation of tyrosine in the crystals of L-tyrosine¹ is different from that in L-tyrosine hydrochloride² but is the same as the one reported for L-phenylalanine hydrochloride³ and potassium L-tyrosine-o-sulphate dihydrate.⁴ We have now investigated the crystal structure of DL-tyrosine in which the molecular packing necessarily is different from that in L-tyrosine. Furthermore, as the crystal data on L-tyrosine were found to be different from those derived from powder data,⁵ we felt that the crystal data on DL-tyrosine arrived at from powder photographs⁶ ought to be checked against single crystal results.

EXPERIMENTAL

DL-Tyrosine was recrystallized following the same procedure as the one worked out for L-tyrosine.⁷ A single crystal with approximate dimensions $0.2 \times 0.25 \times 0.35$ mm³ was mounted with the direction of the largest dimension (c) along the goniometer head axis and used in all the X-ray experiments.

Precession photographs indicate orthorhombic symmetry; space group extinctions are $(0kl)$ for $k+l$ odd and $(h0l)$ for h odd. The absence of centers of symmetry was proved by Wilson statistics; the space group is thus $Pna2_1$.

Unit cell dimensions were determined from diffractometer measurements on 30 general reflections using $MoK(\alpha_1 + \alpha_2)$ radiation ($\lambda = 0.71069 \text{ \AA}$). It may be noted that one axis (6.810 \AA) has half the length of that derived from powder data (13.68 \AA).⁶

The intensity data were recorded with the use of a SYNTEX PI diffractometer with graphite crystal monochromated MoK radiation for 1369 independent reflections with $2\theta < 60^\circ$ using the $\omega - 2\theta$ scanning mode with scan speed varying from 2 to $12^\circ \text{ min}^{-1}$ dependent on the intensity. The scan range was from 1° below $2\theta (\alpha_1)$ to 1° above $2\theta (\alpha_2)$ and background counts were taken for half the scan time at each of the scan range limits. Three standard reflections measured after every 50 reflections showed no systematic fluctuation during the experiment. The 1180 reflections greater than $2.5\sigma(I)$ were considered to be observed; the remaining 189 were excluded from the structure refinement procedure.

The structure determination was based on the program MULTAN written by P. Main, M. M. Woolfson and G. Germain. All other computer programs applied are described in Ref. 10.

Atomic form factors used were those of Doyle and Turner⁸ for oxygen, nitrogen, and carbon atoms and of Stewart, Davidson and Simpson⁹ for hydrogen.

CRYSTAL DATA

DL-4-Hydroxyphenylalanine (DL-tyrosine) $C_9H_{11}O_3N$, orthorhombic. $a = 20.836$ (0.0008) \AA ; $b = 6.810$ (0.002) \AA ; $c = 5.905$ (0.002) \AA . $V = 837.9 \text{ \AA}^3$, $M = 181.19$, $F(000) = 384$, $Z = 4$, $D_{\text{calc}} = 1.435 \text{ g cm}^{-3}$. Absent reflections: $(0kl)$ for $k+l$ odd, $(h0l)$ for h odd. Space group $Pna2_1$.

STRUCTURE DETERMINATION

The intensity data were put on an absolute scale by Wilson's statistical method and normalized structure amplitudes were calculated. The 154 of these with E -values larger than 1.50 were used as input in the program assembly MULTAN.¹¹ Of the solutions obtained the one with the highest figure of merit was used as the basis for an E -map which readily indicated the positions of all non-hydrogen atoms. Initial least-squares refinement included the thirteen heavy atoms with isotropic, then anisotropic temperature factors. When the R -factor reached 0.05 all hydrogen atoms were assigned coordinates from stereochemical considerations and were included in the further least-squares calculations with individual isotropic thermal parameters. After five cycles the shifts in the parameters were negligible compared to the corresponding standard deviations. The final conventional R -factor was 0.037 and the weighted R was 0.055. A comparison of observed and calculated structure factors is given in Table 1; the final parameters for non-hydrogen atoms are listed in Table 2 and for hydrogen atoms in Table 3.

Magnitudes and directions of the principal axes of the ellipsoids of vibrations are given in Table 4. An analysis of the thermal parameters showed that the phenol and alanine parts may each be regarded as rigid bodies. The phenol part has the axis of the largest oscillation (4.6°) nearly parallel to the O1—C4—C1—C7 direction; the alanine part has the corresponding axis (7.9°) parallel to the C8—O3 direction. The bond lengths were corrected for librational effects according to this model.

Table 1. Observed and calculated structure factors. The columns are h , k , l , $10|F_o|$ and $10|F_c|$.

2 0 0	55 55	11 4 0	33 29	24 0 1	23 20	18 4 1	53 52	18 0 2	61 62	12 4 2	102 101
4 0 0	353 353	12 4 0	110 111	26 0 1	79 76	19 4 1	29 29	20 0 2	29 21	13 4 2	291 292
6 0 0	292 286	13 4 0	183 190	28 0 1	82 78	28 4 1	36 36	22 0 2	27 26	14 4 2	144 146
8 0 0	555 558	14 4 0	141 135	0 1 1	1022 1008	21 1 1	47 47	24 0 2	71 70	15 4 2	2 4
10 0 0	616 603	16 4 0	43 43	1 1 1	144 131	23 1 1	65 62	26 0 2	67 67	16 4 2	114 113
12 0 0	18 16	17 4 0	68 67	2 1 1	47 47	24 1 1	32 26	28 0 2	83 80	17 4 2	126 129
13 0 0	22 8	18 4 0	36 36	3 1 1	24 24	24 4 1	40 40	1 1 2	110 179	14 4 2	16 32
14 0 0	36	26 4 0	36 38	4 1 1	23 23	25 4 1	40 40	2 1 2	396 375	19 4 2	35 35
15 0 0	74 67	28 4 0	121 118	5 1 1	23 23	0 5 1	37 39	3 1 2	399 394	20 4 2	29 33
16 0 0	159 178	21 4 0	68 58	6 1 1	257 248	1 5 1	182 181	4 1 2	478 455	21 4 2	53 53
20 0 0	145 196	22 4 0	41 42	7 1 1	145 146	2 5 1	93 98	5 1 2	325 321	22 4 2	66 64
22 0 0	57 53	23 4 0	62 43	8 1 1	126 123	3 5 1	93 99	6 1 2	96 88	23 4 2	31 28
24 0 0	89 87	24 4 0	44 50	9 1 1	72 73	4 5 1	44 43	7 1 2	150 161	25 4 2	41 42
26 0 0	73 73	25 4 0	22 24	10 1 1	297 246	5 5 1	20 19	8 1 2	188 182	1 5 2	77 77
28 0 0	25 18	26 4 0	35 37	11 1 1	73 72	6 5 1	114 110	9 1 2	287 272	2 5 2	49 49
1 1 0	203 202	2 5 0	190 194	12 1 1	156 161	7 5 1	142 138	10 1 2	36 37	3 5 2	45 43
2 1 0	327 338	3 5 0	96 89	13 1 1	176 170	8 5 1	30 30	11 1 2	150 149	4 5 2	229 228
3 1 0	648 666	4 5 0	29 31	14 1 1	204 201	9 5 1	114 110	12 1 2	73 71	5 5 2	204 200
4 1 0	266 277	6 5 0	99 100	15 1 1	29 34	10 5 1	62 62	13 1 2	107 106	6 5 2	114 124
5 1 0	667 722	12 4 0	127 129	16 1 1	95 96	11 5 1	48 48	14 1 2	160 170	7 5 2	85 84
6 1 0	443 455	8 5 0	98 92	17 1 1	65 69	12 5 1	66 69	15 1 2	132 137	8 5 2	25 25
7 1 0	363 340	9 5 0	90 95	18 1 1	51 57	13 5 1	95 96	16 1 2	155 154	9 5 2	44 43
8 1 0	399 343	10 5 0	37 31	19 1 1	74 78	14 5 1	143 145	18 1 2	191 195	10 5 2	73 71
9 1 0	111 114	11 5 0	125 125	20 1 1	121 123	15 5 1	120 124	19 1 2	40 39	11 5 2	27 31
10 1 0	285 188	12 5 0	48 46	21 1 1	42 46	16 5 1	65 64	20 1 2	50 52	12 5 2	50 51
12 1 0	351 355	13 5 0	79 79	23 1 1	26 28	17 5 1	75 74	21 1 2	27 31	14 5 2	37 37
13 1 0	206 207	14 5 0	34 33	24 1 1	22 25	18 5 1	59 74	22 1 2	65 63	15 6 2	204 200
14 1 0	151 145	15 5 0	99 101	25 1 1	43 39	19 1 1	34 35	23 1 2	36 36	15 6 2	47 46
15 1 0	107 104	16 5 0	46 42	26 1 1	74 71	20 5 1	44 41	24 1 2	77 76	16 5 2	91 89
16 1 0	144 136	17 5 0	40 42	28 1 1	73 64	21 5 1	28 29	25 1 2	29 38	16 5 2	29 38
17 1 0	112 112	18 5 0	40 41	1 2 1	443 427	2 5 1	54 57	26 1 2	61 59	21 5 2	27 25
18 1 0	56 51	19 5 0	31 32	2 2 1	293 245	2 4 1	34 34	27 1 2	54 54	22 5 2	52 46
19 1 0	56 52	20 5 0	74 74	3 2 1	382 377	1 6 1	25 19	28 1 2	55 56	23 2 2	24 22
20 1 0	143 133	21 5 0	24 12	4 2 1	216 214	2 6 1	109 109	0 2 2	466 431	0 2 2	170 176
22 1 0	27 24	22 5 0	51 85	5 2 1	36 30	3 6 1	149 149	1 2 2	198 190	1 6 2	117 121
23 1 0	31 28	23 5 0	53 52	6 2 1	193 146	4 6 1	187 184	2 2 2	303 300	2 6 2	195 200
25 1 0	40 40	24 5 0	131 136	7 2 1	221 200	6 6 1	38 38	3 2 2	121 120	3 6 2	75 76
26 1 0	32 35	26 5 0	33 34	8 2 1	98 101	7 5 1	70 69	4 2 2	223 247	4 6 2	70 73
27 1 0	51 51	3 6 0	49 45	9 2 1	207 206	8 6 1	46 48	5 2 2	145 144	5 6 2	45 42
28 1 0	74 71	4 6 0	13 15	10 2 1	237 223	9 6 1	45 45	6 2 2	183 187	6 7 2	75 75
1 2 0 0	75 72	5 6 0	25 22	12 2 1	140 132	10 6 1	84 79	7 2 2	126 122	7 6 2	35 40
2 2 0 0	721 732	6 6 0	47 43	12 2 1	137 134	11 6 1	47 46	8 2 2	145 142	1 6 2	69 72
3 2 0 0	544 543	7 6 0	104 104	13 2 1	171 175	12 6 1	102 102	9 2 2	72 73	9 6 2	75 41
4 2 0 0	58 58	17 6 0	84 98	14 2 1	140 140	13 6 1	46 53	10 2 2	195 195	10 6 2	77 76
5 2 0 0	320 386	18 6 0	96 95	15 2 1	47 47	14 6 1	40 38	11 2 2	31 26	12 6 2	51 51
6 2 0 0	319 319	19 6 0	25 25	16 2 1	156 159	15 6 1	19 19	12 2 2	99 96	13 6 2	24 32
7 2 0 0	164 135	13 6 0	97 100	17 2 1	145 148	16 6 1	41 41	13 2 2	141 148	14 6 2	31 31
8 2 0 0	231 223	14 6 0	203 200	18 2 1	64 68	17 6 1	67 70	14 2 2	195 195	15 6 2	24 32
9 2 0 0	72 64	15 6 0	77 81	20 2 1	30 34	16 6 1	67 66	15 2 2	42 44	16 6 2	40 41
10 2 0 0	44 41	16 6 0	50 51	21 2 1	32 31	16 6 1	35 35	16 2 2	161 167	17 6 2	57 61
11 2 0 0	323 313	17 6 0	45 46	23 2 1	62 43	20 6 1	35 34	17 2 2	44 45	19 5 2	24 26
12 2 0 0	41 47	18 6 0	65 65	25 2 1	37 37	21 6 1	27 27	18 2 2	124 122	20 5 2	45 41
13 2 0 0	416 408	19 6 0	49 47	25 2 1	26 23	0 7 1	58 61	19 2 2	103 102	1 6 2	23 30
15 2 0 0	67 70	22 6 0	47 47	26 2 1	29 33	1 7 1	85 85	2 2 2	72 71	2 7 2	22 22
16 2 0 0	142 143	1 7 0	75 76	28 2 1	42 43	2 7 1	49 49	22 2 2	25 24	3 7 2	52 54
17 2 0 0	195 192	2 7 0	145 150	0 3 1	354 342	3 7 1	49 49	23 2 2	37 36	4 7 2	75 76
19 2 0 0	131 135	3 7 0	47 46	1 3 1	327 323	4 7 1	29 26	24 2 2	43 42	4 7 2	99 100
20 2 0 0	52 52	4 7 0	36 35	2 3 1	259 242	5 7 1	28 25	25 2 2	39 37	5 7 2	27 25
21 2 0 0	63 36	7 0	22 25	3 3 1	330 320	6 7 1	63 61	26 2 2	44 47	9 7 2	34 34
22 2 0 0	78 72	7 0	77 78	4 3 1	156 152	7 7 1	61 59	27 2 2	81 82	10 7 2	53 53
23 2 0 0	61 46	9 7 0	112 112	5 3 1	165 158	8 7 1	45 44	1 3 2	130 123	11 7 2	32 33
26 2 0 0	73 72	10 7 0	36 33	6 3 1	162 162	9 7 1	88 85	2 3 2	260 253	12 7 2	56 55
28 2 0 0	77 77	14 7 0	24 14	7 3 1	127 133	10 7 1	31 29	3 3 2	204 193	13 7 2	25 25
1 3 0 0	56 61	16 7 0	79 75	8 3 1	87 81	11 7 1	97 93	4 3 2	94 94	15 7 2	37 37
2 3 0 0	17 23	17 7 0	79 80	8 3 1	79 81	12 7 1	92 90	5 3 2	265 247	17 7 2	44 46
3 3 0 0	48 54	17 7 0	57 56	10 3 1	46 49	13 7 1	25 25	6 3 2	188 185	16 7 2	58 55
4 3 0 0	127 111	18 8 0	63 68	11 3 1	177 176	14 7 1	46 46	7 3 2	182 185	1 8 2	34 34
5 3 0 0	89 79	18 8 0	40 47	12 3 1	108 109	15 7 1	36 36	8 3 2	192 185	1 8 2	15 15
6 3 0 0	47 39	18 8 0	30 28	13 3 1	122 124	16 7 1	51 52	9 3 2	201 203	2 8 2	71 72
7 3 0 0	130 128	14 8 0	36 36	14 3 1	159 190	17 7 1	73 73	10 3 2	17 19	3 8 2	45 49
8 3 0 0	117 117	15 8 0	28 28	15 3 1	107 106	18 7 1	31 31	11 3 2	94 97	4 8 2	65 66
9 3 0 0	142 146	6 6 0	49 49	16 3 1	128 132	19 7 1	60 60	12 3 2	58 58	5 8 2	91 90
10 3 0 0	205 196	7 6 0	71 51	17 3 1	46 46	19 8 1	24 25	13 3 2	115 119	6 8 2	26 26
11 3 0 0	216 220	8 6 0	37 38	18 3 1	68 68	20 8 1	49 48	13 3 2	31 37	7 8 2	33 33
12 3 0 0	150 145	9 8 0	26 23	19 3 1	32 27	21 8 1	31 27	23 3 2	41 41	4 9 2	45 41
13 3 0 0	37 37	10 8 0	66 47	23 3 1	86 86	4 8 1	60 64	25 3 2	67 61	6 9 2	24 22
14 3 0 0	55 67	11 9 0	77 51	5 4 1	67 64	0 9 1	48 48	26 3 2	39 37	7 9 2	38 37
15 3 0 0	41 41	12 8 0	76 74	23 3 1	27 33	3 9 1	32 32	27 3 2	54 49	2 0 3	419 404
16 3 0 0	76 72	13 8 0	55 54	24 3 1	36 40	8 6 1	50 50	19 3 2	60 60	13 8 2	62 59
17 3 0 0	39 41	14 9 0	65 62	25 3 1	57 50	8 9 1	47 43	24 2 2	154 153	6 0 3	197 187
18 3 0 0	64 79	15 8 0	40 37	27 3 1	39 32	10 8 1	44 43	24 2 2	177 173	8 0 3	220 210
19 3 0 0	51 52	16 8 0	28 32	1 4 1	187 186	11 8 1	50 50	20 2 2	491 391	4 4 2	212 233
20 3 0 0	39 39	17 9 0	51 56	2 4 1	245 245	12 8 1	31 27	24 2 2	46 46	12 3	219 246
21 3 0 0	60 57	18 9 0	34 33	3 4 1	239 235	15 8 1	33 32	25 3 2	67 61	6 4 2	24 22
22 3 0 0	38 38	19 8 0	24 54	4 4 1	67 64	0 9 1	48 48	26 3 2	39 37	7 9 2	38 37
23 3 0 0	66 68	19 9 0	82 81	5 4 1	103 100	3 9 1	38 37	27 3 2	54 49	4 0 3	133 127
24 3 0 0	86 82	9 9 0	82 81	6 4 1	176 168	7 9 1	43 43	3 2 2	35 31	2 9 2	45 49
1 4 0 0	48 49	4 8 1	453 449	8 4 1	64 58	8 9 1	33 29	22 2 2	67 67	2 9 2	47 50
2 4 0 0	142 179	6 0 1	200 178	9 4 1	45 46	0 8 2	435 429	3 4 2	177 173	12 3	219 246
3 4 0 0	246 246	8 0 1	75 77	10 4 1	89 88	2 0					

Table 1. Continued.

0	1	3	229	232	5	6	3	46	42	10	2	4	82	79	14	7	4	31	26	1	5	5	57	53	3	4	6	27	29	
1	1	3	133	133	7	6	3	127	127	11	2	4	176	159	8	4	4	76	74	2	5	5	22	26	4	4	6	23	11	
2	1	3	341	324	8	5	3	114	114	12	2	4	54	55	1	4	5	29	31	3	5	5	39	37	5	4	6	51	51	
3	1	3	16	16	9	5	3	92	91	13	2	4	72	71	2	8	4	31	23	4	5	5	39	37	6	4	6	65	67	
4	1	3	103	102	10	5	3	82	81	14	2	4	56	57	3	8	4	30	20	6	5	5	22	25	7	6	6	54	54	
5	1	3	156	152	12	6	3	49	90	15	2	4	45	44	8	8	4	32	26	6	5	5	36	37	8	4	6	40	44	
6	1	3	170	167	14	5	3	50	51	16	2	4	22	20	2	0	5	209	362	7	5	5	47	50	9	4	6	27	22	
7	1	3	125	123	15	5	3	45	44	17	2	4	76	82	4	0	5	171	172	8	9	5	71	74	10	4	6	33	37	
8	1	3	82	84	16	5	3	52	69	18	2	4	27	31	6	0	5	221	224	9	9	5	13	34	12	4	6	14	11	
9	1	3	95	93	17	5	3	31	35	19	2	4	74	78	8	0	5	217	237	10	5	5	62	61	13	4	6	59	60	
10	1	3	68	67	18	5	3	66	62	21	2	4	73	74	10	0	5	82	81	11	5	5	30	30	14	4	6	37	39	
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14	1	3	151	155	2	6	3	76	76	2	1	4	100	103	18	0	5	111	113	16	5	5	34	36	3	5	6	27	21	
15	1	3	95	99	3	6	3	35	35	3	3	4	144	148	20	0	5	80	77	17	5	5	29	25	4	5	6	34	30	
16	1	3	202	202	4	6	3	57	47	4	3	4	74	78	22	0	5	65	64	14	5	5	49	51	5	5	5	121	120	
17	1	3	67	70	5	6	3	51	48	5	3	4	75	75	1	1	5	217	215	2	6	5	44	44	6	5	6	54	55	
18	1	3	95	94	5	6	3	84	83	6	3	4	112	124	1	1	5	215	215	4	5	5	40	43	7	6	6	24	24	
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20	1	3	82	79	8	6	3	54	49	8	3	4	20	20	3	1	5	111	118	6	6	5	87	86	9	5	6	57	52	
21	1	3	50	45	9	6	3	43	43	9	3	4	144	142	4	1	5	201	202	8	6	5	54	53	10	5	6	24	27	
22	1	3	50	47	10	6	3	79	83	10	3	4	91	88	5	1	5	68	65	9	6	5	78	78	11	5	6	33	32	
23	1	3	218	217	12	6	3	71	68	11	3	4	84	85	6	1	5	34	40	10	6	5	53	58	12	5	6	52	46	
24	2	3	173	169	13	6	3	25	26	12	3	4	97	96	7	1	5	125	125	11	6	5	46	45	8	6	6	102	102	
25	4	2	3	327	311	15	6	3	29	25	14	3	4	134	144	9	1	5	129	129	14	6	5	56	51	2	6	6	46	45
26	5	2	3	132	127	16	6	3	59	61	16	3	4	108	99	10	1	5	61	57	6	0	5	87	87	3	6	6	40	40
27	6	2	3	121	124	19	6	3	61	61	17	3	4	114	117	11	1	5	107	113	1	7	5	26	29	4	6	6	27	26
28	7	2	3	177	177	20	5	3	35	35	18	3	4	57	59	12	0	5	99	99	3	7	5	57	52	5	6	6	26	29
29	8	2	3	177	180	7	7	3	134	136	19	3	4	57	59	13	1	5	63	63	4	7	5	63	59	6	6	6	26	30
30	9	2	3	62	65	1	7	3	74	74	20	3	4	52	51	14	1	5	159	169	6	7	5	36	39	8	6	6	55	47
31	11	2	3	101	99	2	7	3	120	120	21	3	4	55	52	15	1	5	52	46	8	7	5	35	35	2	7	7	114	114
32	12	2	3	188	185	3	7	3	67	69	22	3	4	33	33	16	1	5	65	66	9	7	5	33	27	4	8	7	144	142
33	13	2	3	105	112	4	7	3	54	53	23	3	4	25	24	18	1	5	67	64	0	0	5	40	79	6	0	7	161	160
34	14	2	3	99	96	5	7	3	72	74	24	3	4	56	60	19	1	5	51	52	2	0	5	53	83	8	0	7	96	89
35	15	2	3	122	126	6	7	3	91	87	0	4	4	252	264	20	1	5	78	73	4	0	6	167	151	10	0	7	55	55
36	16	2	3	136	139	7	7	3	31	32	1	4	4	62	78	21	1	5	67	47	6	0	6	102	103	12	0	7	36	34
37	17	2	3	84	89	8	7	3	30	29	2	4	4	78	80	22	1	5	58	58	8	0	6	35	42	14	0	7	50	47
38	18	2	3	93	91	9	7	3	45	44	3	4	4	67	70	23	1	5	67	45	10	0	5	31	32	1	1	7	124	117
39	19	2	3	77	77	12	7	3	73	75	4	4	4	33	32	1	2	5	72	70	12	0	6	59	57	1	1	7	116	116
40	20	2	3	35	37	13	7	3	62	60	5	4	4	44	44	2	2	5	145	142	14	6	6	121	123	2	1	7	51	49
41	21	2	3	96	95	14	7	3	55	55	6	4	4	45	61	3	2	5	145	142	14	6	6	63	63	3	1	7	47	47
42	22	2	3	39	41	16	7	3	62	60	14	2	4	40	40	13	2	5	151	156	20	0	6	75	75	1	2	7	32	32
43	23	2	3	12	12	2	8	3	66	68	19	4	4	40	62	12	1	5	59	60	1	2	5	56	57	13	2	7	57	54
44	24	2	3	27	27	9	5	3	54	54	9	4	4	65	65	2	3	5	55	57	7	2	5	55	55	14	2	7	27	25
45	25	3	3	59	59	20	0	4	65	81	11	5	4	124	121	3	3	5	110	108	3	2	5	52	55	0	3	7	72	71
46	26	3	3	101	113	22	0	4	61	40	12	5	4	102	101	4	3	5	41	35	4	2	6	99	97	1	3	7	54	54
47	27	3	3	68	68	24	0	4	65	61	13	5	4	23	22	5	3	5	40	47	5	2	6	76	76	3	3	7	31	35
48	28	3	3	64	66	1	1	4	117	117	13	5	4	23	22	19	2	5	55	55	17	4	5	41	35	1	4	7	41	43
49	29	3	3	66	66	10	1	4	117	117	15	5	4	25	25	18	2	5	55	55	13	4	5	48	45	4	3	7	41	43
50	30	3	3	67	67	11	1	4	117	117	15	5	4	26	25	18	2	5	55	55	13	4	5	48	45	4	3	7	41	43
51	31	3	3	69	69	10	1	4	34	33	3	4	4	26	25	18	3	5	55	55	13	4	5	48	45	4	3	7	41	43
52	32	3	3	66	69	10	1	4	35	34	15	6	4	26	25	18	3	5	55	55	13	4	5	48	45	4	3	7	41	43
53	33	3	3	65	69	10	1	4	35	34	15	6	4	26	25	18	3	5	55	55	13	4	5	48	45	4	3	7	41	43
54	34	3	3	106	109	12	2	4	153	155	6	4	4	49	47	20	3	5	29	32	17	2	6	71	72	6	4	7	53	49
55	35	3	3	151	158	13	1	4	66	63	7	6	4	47	45	21	3	5	40	38	19	2	6	36	36	7	4	7	54	53
56	36	3	3	128	129	14	1	4	96	99	8	6	4	54	52	14	1	5	95	95	1	3	6	24	20	8	4	7	28	27
57	37	3	3	84	84	15	1	4	61	46	10	6	4	55	52	24	2	5												

Table 2. Fractional atomic coordinates and thermal parameters with estimated standard deviations ($\times 10^5$) for non-hydrogen atoms. The temperature factor is given by $\exp - (B_{11}h^2 + B_{22}k^2 + B_{33}l^2 + B_{12}hk + B_{13}hl + B_{23}kl)$.

Atom	<i>x</i>	<i>y</i>	<i>z</i>	<i>B</i> ₁₁	<i>B</i> ₂₂	<i>B</i> ₃₃	<i>B</i> ₁₂	<i>B</i> ₁₃	<i>B</i> ₂₃
O1	-10168 7	73060 27	17400 3	112 39	2000 49	1565 17	-117 22	-20 81	273
O2	17158 8	40200 24	2030 44	169 4	1400 31	1506 44	-219 17	-126 22	-464 68
O3	21463 8	67438 25	-12859 45	161 4	1659 34	1047 43	-103 18	51 21	30 68
N	19866 8	49323 26	44313 45	113 3	1336 35	949 40	36 17	5 21	116 71
C1	9488 9	79124 27	28302 47	108 4	933 34	1319 52	45 18	35 26	-78 76
C2	6665 9	84401 30	7769 50	117 4	1136 37	1185 49	39 20	161 26	313 77
C3	96 10	82104 31	4253 52	122 4	1182 36	1284 57	68 20	-48 28	202 83
C4	-3696 9	74333 29	21334 48	104 4	972 35	1340 55	-15 19	31 26	-252 79
C5	-970 10	68510 31	41867 52	122 4	1119 35	1247 53	-13 20	116 26	314 78
C6	5619 10	71018 30	45145 48	123 4	1189 37	1151 50	117 20	53 27	284 76
C7	16567 9	82769 29	32339 48	108 4	1115 35	1426 61	-30 19	18 24	-331 76
C8	20904 9	65017 31	27242 47	87 4	1181 37	850 45	-76 18	-9 22	-81 74
C9	19741 8	56741 30	3502 47	78 3	1294 39	931 46	54 18	-58 21	-460 77

Standard deviations in bond lengths and angles were computed from the correlation matrix ignoring standard deviations in cell parameters. For distances and angles the standard deviations were calculated to be 0.003 Å and 0.2°, respectively, except for the bond lengths involving hydrogen where the standard deviation varied from 0.03 to 0.06 Å.

DISCUSSION

A drawing of the molecule is shown in Fig. 1 in which the numbering of the atoms is also indicated. The bond lengths (corrected for thermal effects) and valence angles are given in this figure; interatomic distances and bond angles are also listed in Table 5.

The unit cell dimensions for DL-tyrosine are quite close to those found for L-tyrosine,¹ and it is interesting to note that the same relation exists between the space groups of L- and DL-tyrosine as that reported for L- and DL-alanine,¹² viz. $P\bar{2}_12_1\bar{2}_1$ for the optically active crystals and $Pna2_1$ for the racemate. It has recently been pointed out by Pedone and Benedetti¹³ that such relations

Table 3. Fractional coordinates ($\times 10^4$) and isotropic thermal parameters (\AA^2) with estimated standard deviations for hydrogen atoms.

Atom	<i>x</i>	<i>y</i>	<i>z</i>	<i>B</i>
H2	984	8954	-479	5.3
	23	69	112	1.2
H3	-202	8587	-1020	2.5
	13	41	75	0.6
H5	-367	6261	5327	2.1
	15	40	72	0.6
H6	721	6847	5787	4.5
	15	53	87	0.8
H71	1744	8652	4920	2.4
	17	49	91	0.9
H72	1773	9259	2367	1.6
	12	41	65	0.6
H8	2511	6857	3006	1.4
	15	41	60	0.6
HO1	-1170	6871	2893	5.1
	18	45	85	0.8
HN1	1668	4558	4428	5.8
	22	65	112	1.2
HN2	2018	5535	5782	4.8
	14	45	65	0.6
HN3	2284	3906	4282	3.4
	13	43	71	0.7

Table 4. R.m.s. amplitudes of vibration (\bar{u}^2)[‡] and *B*-values (\AA^2) along the principal axes of vibration given by the components of a unit vector *e* in fractional coordinates ($\times 10^3$).

Atom	(\bar{u}^2) [‡]	<i>B</i>	<i>e_x</i>	<i>e_y</i>	<i>e_z</i>
O1	.219	3.80	-9	143	23
	.165	2.15	1	21	-168
	.155	1.89	47	26	6
	.208	3.41	-38	89	3
O2	.181	2.60	20	85	-118
	.143	1.63	21	81	121
	.203	3.26	26	-124	4
	.182	2.63	40	79	17
O3	.135	1.45	5	5	-168
	.178	2.51	9	144	13
	.157	1.95	-47	28	1
	.129	1.32	0	12	-169
C1	.158	1.97	41	52	66
	.154	1.86	8	79	-140
	.143	1.61	-24	112	68
	.176	2.45	31	86	85
C2	.158	1.98	29	-115	23
	.132	1.37	23	32	-145

Table 4. Continued.

	.173	2.36	29	117	18
C3	.163	2.09	-34	68	91
	.145	1.66	18	-58	142
	.162	2.07	14	-89	124
C4	.150	1.77	45	41	-26
	.144	1.63	-5	109	112
	.171	2.32	33	69	93
C5	.164	2.13	-29	117	13
	.137	1.49	19	56	-141
	.181	2.58	31	107	44
C6	.153	1.85	-37	90	28
	.139	1.53	2	44	-161
	.171	2.32	10	-110	107
C7	.153	1.85	46	1	-52
	.149	1.75	11	98	121
	.169	2.25	13	-141	9
C8	.135	1.45	46	39	-24
	.122	1.18	6	13	167
	.179	2.54	8	138	-51
C9	.132	1.38	-43	39	63
	.120	1.14	21	30	149

may indeed be expected between crystals of the optically active and racemic compounds where virtually identical layers of molecules are packed through the operation of different symmetry elements. This appears to be the case in L- and DL-tyrosine, the layer of molecules between the diagonal glide-planes shown in Fig. 3 is nearly the same in both structures. The layers are differently

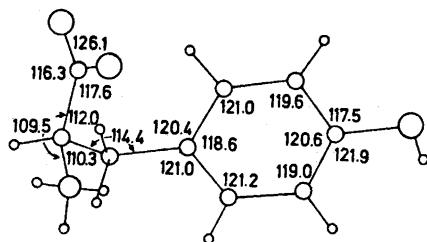
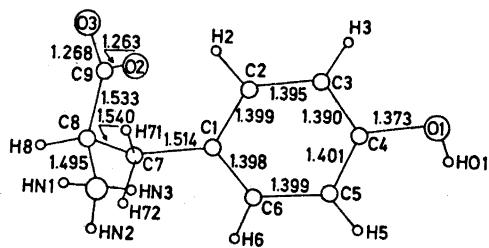


Fig. 1. Bond lengths (\AA) corrected for thermal vibration effects and angles ($^\circ$) in DL-tyrosine.

Table 5. Bond lengths (\AA) and bond angles ($^\circ$) in DL-tyrosine.

	Bond length	Corrected	Bond angle	
C1—C2	1.395	1.399	C1—C2—C3	121.0
C2—C3	1.393	1.395	C2—C3—C4	119.6
C3—C4	1.386	1.390	C3—C4—C5	120.6
C4—C5	1.396	1.401	C4—C5—C6	119.0
C5—C6	1.397	1.399	C1—C6—C5	121.2
C1—C6	1.394	1.398	C2—C1—C6	118.6
C1—C7	1.513	1.514	C7—C1—C2	120.4
C7—C8	1.540		C7—C1—C6	121.0
C8—C9	1.530	1.533	O1—C4—C5	121.9
C8—N	1.485	1.495	O1—C4—C3	117.5
C9—O2	1.251	1.263	C1—C7—C8	114.4
C9—O3	1.262	1.268	C7—C8—C9	112.0
C4—O1	1.371	1.373	C7—C8—N	110.3
O1—HO1	0.81		N—C8—C9	109.5
N—H1N	0.70		C8—C9—O2	117.6
N—H2N	0.90		C8—C9—O3	116.3
N—H3N	0.94		O2—C9—O3	126.1
C2—H2	1.05		Hydrogen bond lengths	
C3—H3	0.99			
C5—H5	0.97			
C6—H6	0.84		O1—O2($-x, 1-y, \frac{1}{2}+z$)	2.668
C7—H71	1.04		N—O1($-x, 1-y, \frac{1}{2}+z$)	2.875
C7—H72	0.87		N—O3($x, y, 1+z$)	2.833
C8—H8	0.92		N—O3($\frac{1}{2}-x, -\frac{1}{2}+y, \frac{1}{2}+z$)	2.856

packed, however, as may be expected since all layers in L-tyrosine are identical whereas the layers in DL-tyrosine consist alternatingly of D and L forms. The difference in packing is visualized in Figs. 2 (A) and (B) which show the coupling of molecules of different layers through the alanine parts; (A) shows the conditions in L-tyrosine as viewed along the *a*-axis (6.913 Å) and (B) represents the DL-tyrosine structure as viewed down the *b*-axis (6.810 Å). The pertinent symmetry elements responsible for the coupling of layers are screw axes and glide-planes, respectively. The molecular packing is somewhat denser in the racemate, the density being 1.435 g cm⁻³ as compared to 1.414 g cm⁻³ for L-tyrosine.

The bond lengths and angles found in the present investigation are nearly identical to those determined from the study of the L-form. Except for the C9—O2 bond in the carboxyl group the differences in corresponding bonds are insignificant, 0.008 Å or less; the C9—O2 bond is found 0.017 Å longer than in L-tyrosine. The differences in corresponding angles are 0.7° or less, the largest dissimilarities are found in the alanine moiety. The conformation in the two structures are also equal; the angle between the phenyl ring plane and the plane defined by the C1—C7—C8 atoms is 85.8° (as compared to 85.9° in L-tyrosine), the dihedral angles C1—C7—C8—C9 and C1—C7—C8—N are 52.5° and 290.3, respectively (53.1° and 290.7° in L-tyrosine) whereas the torsional angle N—C8—C9—O2 is 11.7°, 2.3° less than the corresponding angle in L-tyrosine.

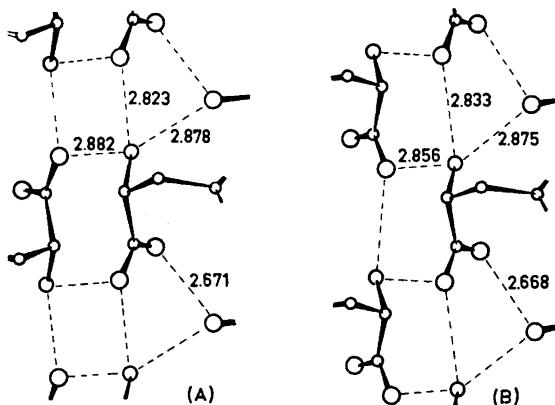


Fig. 2. Molecular arrangement in L-tyrosine (A) and in DL-tyrosine (B) crystals.

The hydrogen bond system in the crystals of DL-tyrosin is shown in Fig. 3. Each molecule is hydrogen donor as well as acceptor in four hydrogen bonds in such a way that one molecule is hydrogen bonded to six neighbouring molecules. The hydrogen bond lengths (given in Table 5) within the layers of molecules of equal chirality are nearly the same as the corresponding bonds in L-tyrosine; the hydrogen bonds linking the layers together are by 0.026 Å shorter in the racemate than in the crystals of the L-form.

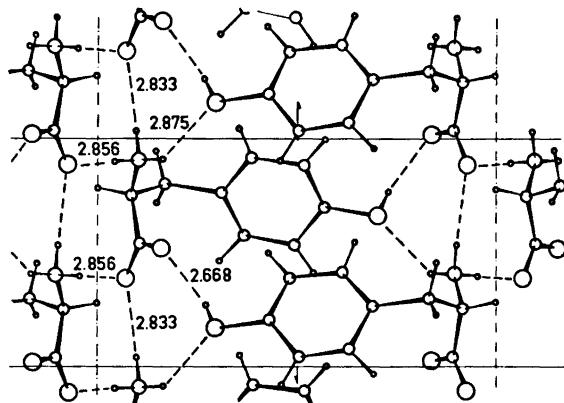


Fig. 3. The crystal structure of DL-tyrosine as viewed along the *b*-axis.

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