# The Disulfide Bonds of Viscotoxin A3 from the European Mistletoe (Viscum album L., Loranthaceae)

GUNNAR SAMUELSSON and BARBRO PETTERSSON

Department of Pharmacognosy, Faculty of Pharmacy, Lindhagensgatan 128, S-112 51 Stockholm, Sweden

Viscotoxin A3 has been digested with pepsin and trypsin, and the resulting peptides have been isolated. Two peptides are key fragments for deduction of the position of the disulfide bonds. The amino acid composition of the two peptides, and partial amino acid sequences of fragments isolated after performic acid oxidation permit the following conclusions. One disulfide bond links residue 16 (Cys) with residue 26 (Cys) in the amino acid sequence of Viscotoxin A3. Two other disulfide bonds link residues 3 (Cys) and 4 (Cys) with residues 32 (Cys) and 40 (Cys), but it has not been possible to show if residue 3 (Cys) is linked to residue 32 (Cys) or to residue 40 (Cys), and vice versa for residue 4 (Cys). This arrangement of the disulfide bonds gives Viscotoxin A3 a very compact structure, which is also reflected in the remarkable stability of the substance, e.g., against heat.

Viscotoxin A3 is one of the pharmacologically active proteins, first isolated from the European mistletoe, Viscum album L., as a mixture called Viscotoxin.<sup>1,2</sup> Following performic acid oxidation of Viscotoxin, the oxidation product of Viscotoxin A3 was isolated <sup>3</sup> and the amino acid sequence determined.<sup>4</sup> A method for isolation of pure native Viscotoxin A3 was later developed.<sup>5</sup> Viscotoxin A3 contains 6 cysteine residues, forming 3 disulfide bonds. The availability of pure native Viscotoxin A3 has now permitted determination of the position of these disulfide bonds, as reported in this paper.

#### EXPERIMENTAL

# Material and apparatus

Viscotoxin A3 was isolated from crude Viscotoxin 2 by chromatography on cellulose phosphate and SE-Sephadex<sup>®</sup>, as described previously.<sup>3,5</sup>

Trypsin, Worthington, 2 × crystallized, salt free. Batch 6118. The enzyme was treated with L-(1-tosylamido-2-phenyl) ethyl-chloromethyl ketone to inhibit chymotryptic activity, as described by Kostka and Carpenter.

Pepsin, Bovine, 3× crystallized. Serva Entwicklungslabor., Heidelberg, DBR.

Cation exchanger Dowex 50 W-X2. Purified and subjected to hydraulic separation of particle sizes as described previously.4

SE-Sephadex<sup>®</sup>. AB Pharmacia, Uppsala, Sweden. Sephadex<sup>®</sup> G-10, G-15, and G-25. AB Pharmacia, Uppsala, Sweden.

pH-stat. Radiometer titrator TTTlc with titrigraph SBR2c. Radiometer, Copenhagen, Denmark.

### Methods

Digestion of Viscotoxin A3 with pepsin and trypsin. 200 mg Viscotoxin A3 was dissolved in 20 ml of water, and the pH of the solution adjusted to 2.0 by addition of 6 N HCl. 8 mg pepsin was added, and the solution was kept at 37°C overnight. The pH was then adjusted to 6.5, and 3 ml 0.15 M CaCl, solution was added. After adjustment of the pH to 6.5, 2 ml of an aqueous trypsin solution (2 mg/ml) was added, and the alkali uptake during digestion at 37°C recorded. Another 2 ml of trypsin solution was added 3.5 h later (final enzyme: protein ratio 1:25), and the digestion continued overnight. The mixture was concentrated in vacuo, diluted with 2% acetic acid, desalted by gel filtra-

tion in 2 % HOAc on Sephadex G-25, and lyophilized.

Separation of peptides from the pepsin-trypsin digestion. The desalted, dry mixture of peptides was dissolved in 2 ml 1/30 M KH<sub>2</sub>PO<sub>4</sub>, and the solution applied to a column  $(10 \times 300 \text{ mm})$  of SE-Sephadex, previously equilibrated with  $1/15 \text{ M PO}_4^{3-}$  buffer of pH 5.0. Elution was performed with a straight line buffer gradient, starting with 1/15 M PO<sub>4</sub><sup>3-</sup> buffer of pH 5.0, and ending with 1/15 M PO<sub>4</sub><sup>3-</sup> buffer, pH 8.0, containing NaCl to a concentration of 0.125 M. The gradient was obtained by the arrangement described by Parr, and the total gradient volume was 1200 ml. Fractions of 5 ml were collected, and their optical density at 232 m $\mu$  determined. Fractions corresponding to UV-absorbing peaks were pooled, concentrated in vacuo, and desalted on Sephadex G-15. Isolated

peptides were dissolved in 2 % HOAc, and the solutions stored in the refrigerator.

Amino acid analysis. Hydrolysis of peptides for amino acid analysis was performed as described by Hirs et al., with modifications according to Crestfield et al. The amino acids were determined with an automatic amino acid analyzer, according to Spackman

et al.,10 as modified by Samuelsson.11

Oxidation of peptides. Oxidation was performed with performic acid at 0°C, according

to Hirs.12

Separation of oxidized peptides. The procedure of Schroeder et al. 13 was used. The column of Dowex 50W-X2 had the dimensions 9 x 1200 mm, and the total volume of the buffer gradient was 1935 ml. Fractions of 5 ml were collected, and aliquots of 0.20 or 0.40 ml were taken from every second fraction for hydrolysis and analysis by the ninhydrin method, as described previously. Fractions corresponding to ninhydrin positive peaks were pooled, desalted on Sephadex G-10, concentrated to dryness in vacuo, and stored in the refrigerator as solutions in 2 % HOAc, until further investigated.

Determination of amino acid sequences. Subtractive Edman degradation was used,

as previously described.4 Amino terminal amino acids were also identified by treatment of the peptides with dansyl chloride,\* hydrolysis of the reaction product, and identification of the liberated dansyl amino acids by thin layer chromatography, as described by

Stehelin and Duranton.<sup>15</sup>

## RESULTS AND DISCUSSION

The conditions used for the enzymatic degradation of native Viscotoxin A3 are essentially the same as those used by Spackman et al. in the studies of the disulfide bonds of ribonuclease. 4 However, digestion with chymotrypsin was omitted, as better results were obtained by digestion with pepsin and trypsin only.

Separation of the digestion products by chromatography on SE-Sephadex is illustrated in Fig. 1. Eight peaks were obtained, numbered 1-8 in the

<sup>\* 1-</sup>Dimethylamino-naphthalene-5-sulfonyl chloride.

figure. Amino acid analysis showed peaks 3, 4, 6, and 8 to contain cystine (Table 1). Except for the analysis of the material from peak 8, the figures of Table 1 have not been corrected for the destruction of serine, threonine,

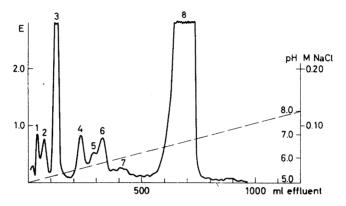


Fig. 1. Separation of peptides from Viscotoxin A3 after digestion with pepsin and trypsin. Column: SE-Sephadex ( $10 \times 300$  mm). Gradient elution with PO<sub>4</sub><sup>3-</sup> buffer. –, optical density at 232 m $\mu$  (left ordinate). – – , calculated pH and NaCl concentration of gradient (right ordinate). For details: see text.

Table 1. Amino acid analysis of cystine-containing peptides, isolated from a pepsintrypsin hydrolysate of Viscotoxin A3 (cf. Fig. 1).

Amino acid	Peak 3 <sup>a</sup>	Peak 4 <sup>a</sup>	Peak 6 <sup>a</sup>	Peak $8^{t}$		
Lysine	2.1	1.2	2.6	4.1		
Arginine	0.9	1.0	0.9	2.9		
Aspartic acid	1.9	1.0	2.3	4.0		
Threonine	2.6	0.8	3.1	$4.9^c$		
Serine	3.7	0.4	5.2	$4.9^c$		
Proline	2.8	1.0	3.4	5.0		
Glycine	2.6	0.3	3.0	3.9		
Alanine	_	1.6		2.9		
ł Cystine	3.5	1.9	3.7	$6.0^d$		
Ísoleucine	0.6	0.2	0.9	2.70		
Leucine	0.8	0.1	1.0	1.9€		
Tyrosine	0.7	0.2	0.9	1.8		

<sup>&</sup>lt;sup>a</sup> Single analysis. 24 h hydrolysis.

or tyrosine. The values for isoleucine are probably also too low, as the  $-\mathrm{Ile}-\mathrm{Ile}$ -bond (residues 34 and 35 in Viscotoxin A3) is very resistant to acid hydrolysis. When these facts are taken into account, the results of the amino acid analyses indicate that peak 3 represents a peptide containing residues

<sup>&</sup>lt;sup>b</sup> Mean values of 3 analyses.

<sup>&</sup>lt;sup>c</sup> Corrected for destruction during hydrolysis.

d Determined as cysteic acid.

After 72 h hydrolysis.

2-10 and 29-46 of Viscotoxin A3, with two disulfide bonds (cf. Fig. 4). Peak 6 could be the same peptide as peak 3, but with the amino terminal lysine (residue 1) intact. Peak 4 corresponds to a peptide formed from residues 14-17 and 24-28, with one disulfide bond (cf. Fig. 4). Peak 8 represents undigested Viscotoxin A3, as seen from the amino acid composition. The peptides obtained from peaks 3, 4, and 6 are in the following denoted as peptides PT3, PT4, and PT6.

Peptide PT4. This peptide was oxidized with performic acid, and the fragments separated on Dowex 50W-X2, as illustrated in Fig. 2. Two peptides

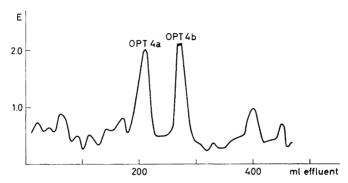


Fig. 2. Separation of oxidation products of peptide PT4 on Dowex 50W-X2 (9 × 1200 mm). Gradient elution with pyridine-HOAc buffer, as described in the text.

were obtained, denoted OPT4a, and OPT4b. Edman degradation of these peptides, as presented in Table 2, gave the partial amino acid sequence  $Pro-Thr-Cys(SO_3H)$ -(Ala, Lys) for OPT4a, and  $Asx-Ala-(Cys(SO_3H), Arg)$ \* for OPT4b, thereby demonstrating a disulfide bond between residues 16 and 26 in the amino acid sequence of Viscotoxin A3 (cf. Fig. 4).

Table 2. Amino acid composition and Edman degradation of peptides OPT4a and OPT4b.

Amino acid			<i>T4a</i> ep	OPT4b Step				
	0	1	2	3	0	1	2	
Lysine Arginine	1.0	a 	a 	a 	_ 1.0	_ a	_ a	
Cysteic acid Aspartic acid	0.9	1.1 —	1.1	0.6	1.2 1.0	$\begin{array}{c} 1.2 \\ \textbf{0.2} \end{array}$	1.0 0.1	
Threonine Proline Alanine	$0.7 \\ 0.9 \\ 1.0$	$\begin{array}{c} 0.8 \\ 0 \\ 1.0 \end{array}$	$\begin{array}{c} 0.5 \\ 0 \\ 1.0 \end{array}$	$\begin{array}{c} 0.4 \\ 0 \\ 1.0 \end{array}$	- - 1.0	- - 1.0	- 0.2	

a Not determined.

Acta Chem. Scand. 25 (1971) No. 6

<sup>\*</sup> Asx=Aspartic acid or asparagine.

Peptide PT6. Chromatographic separation on Dowex 50W-X2 of the oxidation products of peptide PT6 yielded 4 peptides, denoted OPT6a, OPT6b, OPT6c, and OPT6d (Fig. 3). Table 3 shows the amino acid composition of these peptides (step 0). The isoleucine values of peptides OPT6a and OPT6b are too low, as they were obtained from a 24 h hydrolysate only.

Dansylation of the peptides showed that the amino terminal amino acids of the peptides were the following: OPT6a and OPT6b: isoleucine; OPT6c: lysine; and OPT6d: leucine. Three steps of Edman degradation of peptides

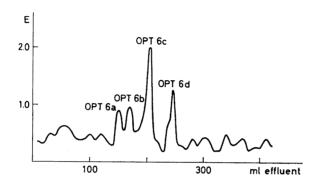


Fig. 3. Separation of oxidation products of peptide PT6 on Dowex 50W-X2 (9 × 1200 mm). Gradient elution with pyridine-HOAc buffers, as described in the text.

Table 3. A	mino acid	composition a	nd Edman	degradation	of peptide	s OPT6a - OPT6d.

Amino acid		OPT6a Step			<i>OPT6b</i> Step			OPT6c Step			OPT6d Step			
	0	1	2	3	0	1	2	3	0	1	2	0	1	2
Lysine	0.9	×	×	×	1.1	×	×	×	0.9	0	×	0.8	×	×
Arginine	_			_				_	0.7	0.6	×	_	_	
Cysteic acid	1.2	1.0	1.0	1.0	1.0	1.0	1.1	0.7	2.1	2.0	1.7	0.8	1.0	1.0
Aspartic acid	1.0	1.0	1.0	1.2	1.0	a	1.1	1.0	1.0	1.0	1.0	-	_	<b>–</b>
Threonine	0.9	0.8	0.8	1.0	1.0	1.0	1.0	0.9	1.8	1.7	1.6	<b>-</b> .	<b>-</b> .	
Serine	$3.0^b$	$2.8^{b}$	$2.9^b$	$2.2^b$	$2.9^b$	$3.1^b$	$2.7^b$	$2.1^b$	$1.3^b$	$1.3^b$	$0.4^b$	$1.2^b$	$1.1^{b}$	$0.2^b$
Proline	2.1	1.7	2.0	1.9	1.9	2.0	2.0	1.9	1.0	1.1	1.0	_	_	_
Glycine	1.3	1.1	1.0	1.0	1.4	1.0	1.2	1.3	1.2	1.0	1.0	1.0	1.1	0.9
Isoleucine	$1.2^{c}$ (2)	$0.9^d$	$0.2^d$	0	$\begin{array}{c} 0.6^c \\ (2) \end{array}$	0.8	0.1	0	_	_	_		-	<b>-</b> .
Leucine	_	<b>—</b> .	<b>–</b> .			-			-	_	_	0.6	0	0
Tyrosine	0.70	$0.4^d$	$0.5^d$	+	0.2	_		_	<b> </b>	-	-	-	-	
3-Chlorotyrosine	-	-	-	-	+	+	+	+	-	-	-	-		-

<sup>×</sup> Not determined.

<sup>a</sup> Not determined, due to technical error.

<sup>d</sup> 72 h hydrolysis.

<sup>&</sup>lt;sup>b</sup> Values corrected for destruction during hydrolysis, assuming 10 % loss in 24 h and 30 % in 72 h.

<sup>&</sup>lt;sup>c</sup> 24 h hydrolysis.

<sup>+</sup> Present, but not determined.

OPT6a and OPT6b, and two degradation steps of peptides OPT6c and OPT6d are presented in Table 3. These results give the following partial amino acid sequences for the four peptides: OPT6a: Ile-Ile-Ser- $(Gly,2Ser,Thr,Cys(SO_3H),2Pro,Tyr,Asx,Lys)$ . OPT6b: Ile-Ile-Ser- $(Gly,2Ser,Thr,Cys(SO_3H),2Pro,TyrCl,Asx,Lys)$ . OPT6c: Lys-Ser- $(2Cys(SO_3H),Pro,Asx,2Thr,Gly,Arg)$ . OPT6d: Leu-Ser- $(Gly,Cys(SO_3H),Lys)$ .

These results can be interpreted as follows: OPT6a consists of residues 34-46 of Viscotoxin A3. OPT6b is an artefact, formed by chlorination of the tyrosine residue in OPT6a. OPT6c and OPT6d correspond to peptides containing residues 1-10 and 29-33, respectively (cf. Fig. 4). Thus, peptide PT6 contains residues 1-10 and 29-46 of the Viscotoxin A3 sequence, and two disulfide bonds connect residues 3 (Cys) and 4 (Cys) with residues 32 (Cys) and 40 (Cys).

Peptide PT3. The amino acid composition of this peptide (Table 1) indicates that it has been formed from peptide PT6 by tryptic cleavage of the amino terminal lysine residue. This peptide was therefore subjected to partial acid hydrolysis with the aim of breaking the peptide bond between the two cysteine residues Nos. 3 and 4, and isolating peptide fragments containing the disulfide bridges and a sufficient number of connected amino acids, to give an answer to the question if residue 3 is linked to residue 32 or to residue 40, and vice versa for residue 4. However, despite several attempts, these experiments were unsuccessful, and this question must therefore remain open until better methods are developed.

The arrangement of the disulfide bridges in Viscotoxin A3. The results of this investigation are summarized in Fig. 4. One disulfide bond links residue 16 (Cys) to residue 26 (Cys), forming a loop of 9 amino acids in the chain. The amino terminal and carboxyl terminal ends of the molecule are brought close to each other by two disulfide bridges, connecting residues 3 (Cys) and

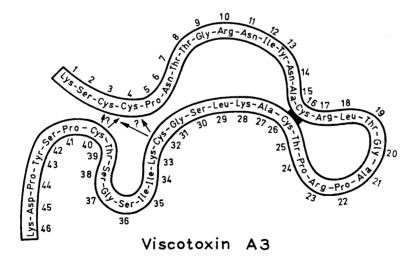


Fig. 4. The arrangement of the disulfide bonds in Viscotoxin A3.

Acta Chem. Scand. 25 (1971) No. 6

4 (Cys) with residues 32 (Cys) and 40 (Cys) also forming a second loop of 7 amino acids (residues 33 - 39). This arrangement of the disulfide bonds gives the molecule a very compact structure, which explains the unusual stability of Viscotoxin A3. Thus heating an aqueous solution to 100°C for 30 min has no influence on the toxicity of the substance, and aqueous solutions have also been kept in the refrigator for several years with no loss of biological activity.

Acknowledgements. This investigation has been supported by grants from the Swedish Medical Research Council (Projects Nos. B69-13X-2084-03A, B70-13X-2084-04B, and B71 - 13X - 2084 - 05C).

#### REFERENCES

- 1. Winterfeld, K. and Bijl, M. Ann. 561 (1948) 107.
- 2. Samuelsson, G. Svensk Farm. Tidskr. 65 (1961) 481.
- 3. Samuelsson, G. Acta Chem. Scand. 20 (1966) 1546.
- 4. Samuelsson, G., Seger, L. and Olson, T. Acta Chem. Scand. 22 (1968) 2624.
- Samuelsson, G. and Pettersson, B. Acta Chem. Scand. 24 (1970) 2751.
   Kostka, V. and Carpenter, F. H. J. Biol. Chem. 239 (1964) 1799.
   Parr, C. W. Proc. Biochem. Soc. 324th Meeting XXVII.

- 8. Hirs, C. H. W., Stein, W. H. and Moore, S. J. Biol. Chem. 211 (1954) 941.
  9. Crestfield, A. M., Moore, S. and Stein, W. H. J. Biol. Chem. 238 (1963) 622.
  10. Spackman, D. H., Stein, W. H. and Moore, S. Anal. Chem. 30 (1958) 1190.
  11. Samuelsson, G. Svensk Kem. Tidskr. 80 (1967) 98.
  12. Hirs, C. H. W. J. Biol. Chem. 219 (1956) 611.

- 13. Schroeder, W. A., Jones, R. T., Cormick, J. and McCalla, K. Anal. Chem. 34 (1962) 1570.
- 14. Spackman, D. H., Stein, W. H. and Moore, S. J. Biol. Chem. 235 (1960) 648.
- 15. Stehelin, D. and Duranton, H. J. Chromatog. 43 (1969) 93.

Received November 14, 1970.