The Diterpenoids of Some Solidago Species

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We have continued our studies of the diterpenoids in the genus Solidago. From extracts of the roots of S. Shortii Torr. & Gray we have isolated two new epoxybutenolides closely related to the elongatolides. The β -substituted $\Delta^{\alpha,\beta}$ -butenolide ring is clearly visible in the UV [λ_{\max} (EtOH) 226 nm], IR (1780, 1750 cm⁻¹) and NMR [4.16 (H-14), 5.27 (2 H-16) τ] spectra.

Resonances in the NMR spectrum at 7.11 (1 H d, J=5 Hz) and 8.78 (3 H s) are indicative of a trisubstituted epoxy grouping. This together with the presence of two additional tertiary (8.89, 9.06) and one secondary (9.12, J=6 Hz) methyl group shows that the new compounds belong to the clerodane class of rearranged labdanes. One of them (1) is not further oxygenated, the other is a secondary angelate (2) [IR 1710 cm⁻¹, MS loss of $C_5H_8O_4$, $C_5H_7O^+ \rightarrow C_4H_7^+ + CO$ m* 36.5, NMR 3.96 (1 H qq, J = 7 and 1.5 Hz) 8.0 - 8.2 (6 H multiplet)]. The carbinyl proton in the NMR spectrum gives rise to a doublet of triplets centred at 7 5.02 (see Fig. 1) diagnostic of an axial proton coupled to one neighbouring equatorial (J=6.0) and two axial (J=10.5)protons, thus suggesting the oxygen function to be located either at C-1 or C-7. Evidence which lead to choice of the latter was found in the mass spectral process: $C_{35}H_{36}O_5$ $^++\rightarrow C_{30}H_{36}O_3$ $^++^2\rightarrow C_{14}H_{31}O^{+2}\rightarrow C_{11}H_{15}^{+}$ (3). Loss of the epoxy function as C_3H_6O to furnish ion 3 was found to be a more likely process with the oxygen function in ring B.

Hydrolysis (KOH/MeOH, room temp., 12 h) of the angelate furnished an aldehyde acid (4) purified on TLC as its methyl ester (5) and characterized by its spectroscopic properties [NMR 0.29 (1 H), 6.31 (3 H) M + $C_{36}H_{40}O_{4}$].

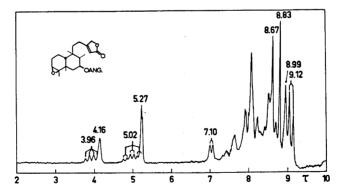


Fig. 1. The NMR spectrum at 60 Mc/s of the butenolide 2.

Solidago virgaurea L. did not yield any diterpenoids. Besides acetylenic compounds 3 we have found an aromatic ester (6) which has been identified by spectroscopic methods [NMR 6.28 (6 H s), 4.72 (2 H s), 3.54 (2 H), 2.84 (1 H) (A_2B , J=8.5 Hz); MS $C_{16}H_{16}O_4^{+}$. (40 %) $\stackrel{?}{=}$ $C_9H_9O_3^{+}$ (100 %)] and by direct comparison with an authentic sample isolated from Aster ptarmicoides Torr. & Gray by Hauge and Sørensen,4 and also synthesized by the same authors. A related, further oxygenated ester, was isolated from S. rigida L. Its spectroscopic properties are similar to those given for an ester (7) also found in Aster ptarmicoides.5 However, our m.p. 94-96°C is different from Bohlmann's 35°C. The chemical relationship between the genera Aster and Solidago is not surprising since they both belong to the tribe Astereae in the Compositae family.

In addition to this, S. rigida L gave a diterpenoid acid which is identical with (NMR, MS, IR, TLC) ent-16-kauren-19-oic acid (8).

We did not succeed in isolating any diterpenoids from S. flexicaulis L.; however, four well known acetylenic methyl esters were identified, 2-trans, 8-cis- and 2-cis, 8cis-matricaria ester and 2-cis- and 2-transdehydromatricaria ester.

Acknowledgements. We want to thank Curator Jan Tengnér, Bergianska Trädgården, Stockholm, for plant material, Dr. Jørgen Lam, Aarhus Universitet, for a copy of his manuscript prior to publication, and Mrs. J. Stene Sørensen, this university, for identifying the acetylenes. This work was supported by a grant from Norges tekniske høgskole to G. B., for which we are grateful.

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Received June 18, 1971.

Isolation of ent-16-Kauren-19-oic Acid and ent-16-Kauren-19-ol from Abrotanella nivigena Muell.

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he biologically important diterpenoid ent-16-kauren-19-oic acid (1) has been isolated from a number of natural sources. Originally it was found in Ricinocarpus stylosus,1 but later it has been shown to be present also in Gibberella fujicuroi,2 Aralia cordata and, using a bioassay technique, in barley together with the corresponding alcohol ent-16-kauren-19-ol

The acid and the alcohol which both show gibberellin-like activity,5 are intermediates in the biosynthesis of the gibberellins 6-9 and they are synthesized in wild cucumber, Echinocystis macrocarpa,

from kaurene.10

We now wish to report the occurrence of these two diterpenoids in the Australian member of the Compositae family Abrotanella nivigena Muell.

ent-16-Kauren-19-oic acididentified by its physical and spectroscopic properties. Like earlier workers 2 we were, in spite of several recrystallizations, unable to increase the melting point above 160-163° to reach the originally reported 1 value of 179 - 181°.

The IR spectrum had bands at 1690, 1660, and 875 cm⁻¹ due to the carboxylic acid and exocyclic methylene groups. The NMR spectrum exhibited two sharp singlets at τ 9.05 (C-10 Me) and 8.76 (C-4 β Me) and three broad signals at τ 5.21 (2H C-17), 7.37 (1H C-13) and 7.91 (2H C-15).

A prominent peak at m/e 302 (78 %) in the mass spectrum revealed the presence of a stable molecular ion, the base peak being at m/e 91 ($C_7H_7^+$).

A peak at m/e 259 (38%) due to the transition $302^{+*} 259^{+} + 43$ (°C₃H₇) has its origin in extrusion of a part of ring A. The presence of the carboxylic acid group is confirmed by the processes $302^+ \rightarrow 257^+ + 45$ (·CO₂H) (14 %), $287^+ \rightarrow 241^+ + 46$ (HCO₂H) (21 %) and $259^{+2} \rightarrow 213^+ + 46$ (HCO_2H) (19 %).