tion products on Sephadex LH-20.8 In the present study a few experiments with lignin model compounds in the system Sephadex LH-20/dimethylformamide were performed. The results (I, $K_{\rm d}'=0.72$; II, $K_{\rm d}'=0.54$; IX, $K_{\rm d}'=0.46$) indicate that the gel filtration properties of this system in certain respects resemble those of the systems polystyrene gel/tetrahydrofuran and Sephadex G-25/dimethyl sulfoxide.

Experimental. A Waters Gel Permeation Chromatograph Model 200 (column set B1: 3×10^3 Å, 5×10^2 Å, 10^2 Å, and 60 Å) was used for gel filtration on polystyrene gel with tetrahydrofuran as an eluent. For the experiments with Sephadex gels the following columns and solvents were used:

Sephadex G-25 (fine)/water. Amount of gel, 27 g; bed volume, 133 ml; column length, 35 cm. Sephadex G-25 (fine)/dioxane-water (1:1). Amount of gel, 47 g; bed volume, 199 ml; column length, 57 cm. Sephadex G-25 (fine)/dimethyl sulfoxide. Amount of gel, 32 g; bed volume, 198 ml; column length, 56 cm. Sephadex LH-20/dioxane-water (9:1). Amount of gel, 8 g; bed volume, 32 ml; column length, 18 cm. Sephadex LH-20/dimethylformamide. Amount of gel, 8 g; bed volume, 36 ml; column length, 20 cm.

Gel filtration. The samples (1-3 mg) were dissolved in 1-2 ml of the eluent and applied to the columns. The UV absorbance at 280 nm of the effluent from the columns was continuously recorded by means of a Beckman DB spectrophotometer equipped with a 1 mm flow cell and a recorder. In most experiments the flow rate was 5-10 ml/h.

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Fungal Carotenoids

VII.* Synthesis of β , γ - and γ , γ Carotene with Terminal Methylene
Groups

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Recently Arpin et al.¹ reported the isolation and identification of β , γ -carotene (10) from the discomycete Caloschypha fulgens. The new IUPAC nomenclature rules for carotenoids ² are used here. The γ -terminus features an exocyclic methylene group common to terpenes, but previously unreported in carotenes. We now report the synthesis of β , γ -carotene (10) according to the route outlined in Scheme 1 and confirm the assigned structure. Symmetrical γ , γ -carotene (8) was also synthesized according to the same basic route.

Synthetic β,γ -carotene (10), purified by column chromatography on deactivated alumina and silica gel plates and crystallized from benzene-methanol solution, melted at 174°C, undepressed on admixture with a natural specimen.

The synthetic and natural pigments could not be separated chromatographi-

Synthetic β , γ -carotene (10) had $\lambda_{\rm max}$ 421, 444 [$E(1~\%, 1~{\rm cm}) = 2600$] and 472 [$E(1~\%, 1~{\rm cm}) = 2370$] nm in petroleum ether and 425, 447.5, [$E(1~\%, 1~{\rm cm}) = 2520$] and 475 [$E(1~\%, 1~{\rm cm}) = 2520$] and 475 [$E(1~\%, 1~{\rm cm}) = 2300$] nm in acetone. The IR spectrum (KBr) resembled those of common carotenes except for a strong absorption at 889 cm⁻¹ ascribed to $R_1R_2C = CH_2$. The NMR spectrum (60 Mc) in CDCl₃ solution exhibited singlets at τ 9.17 (3H) and 9.10 (3H) due to gem. dimethyl on the γ -ring, broad singlets at τ 5.31 (1H) and 5.46 (1H) attributed to a $R_1R_2C = CH_2$ grouping with magnetically non-equivalent protons and a doublet at τ 7.57 (1H, J = 8 cps) ascribed to the methne proton at C-6'. Signals at τ 8.98 (singlet, 6H, gem. dimethyl) and τ 8.30 (singlet, 3H, end-of-chain methyl) were caused by the β -end group. Four in-chain methyl groups gave a singlet at τ 8.03

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(12H); non-allylic and allylic methylene groups gave rise to multiplets centered at τ 8.5 (8H) and 7.9 (4H), respectively, and the olefinic protons of the polyene chain (14H) produced signals in the τ 3.4–4.2 region. The mass spectrum exhibited the molecular ion at m/e 536. Fragment ions were observed at M–92, M–106, and M–158.3 No fragmentation within the rings was observed. Ions corresponding to in-chain cleavages were recorded, the strongest ion arising from fragmentation of the 15,15' bond.

The physical properties of synthetic β , γ -carotene (10) correspond to those reported for the natural compound.

Symmetrical γ,γ -carotene (8) has not been reported to occur in nature, although it is suspected to be present in the aphid $Macrosiphum\ liriodendri.^4$ Crystallized from benzene-methanol, synthetic γ,γ -carotene (8) melted at 183°C. The IR spectrum again exhibited strong absorption at 889 cm⁻¹ attributed to the exocyclic methylene groups. The absorption spectrum of γ,γ -carotene (8) in petroleum ether showed $\lambda_{\rm max}$ 414, 438 [$E(1\%, 1\ {\rm cm}) = 2960$] and 468 nm; in acetone at 416, 440 [$E(1\%, 1\ {\rm cm}) = 2870$], and 469 nm. The NMR spectrum in CDCl₃ displayed the expected signals: singlets at τ 9.17 (6H) and 9.10 (6H) due to gem. dimethyl of the γ -ring, broad signals at τ 5.31 (2H) and 5.46 (2H) ascribed to terminal methylene, doublet at τ 7.57 (2H, J=8 cps) from methine at C-6, C-6′, singlet at t 8.03 (12H)

due to in-chain methyl groups, multiplets centered at τ 7.9 and 8.5 from allylic and non-allylic methylene groups, and signals at τ 3.4–4.2 (olefinic protons). The mass spectrum of γ, γ -carotene (8) showed the molecular ion at m/e 536 and the usual M–92, M–106, and M–158 peaks.³ Other ions noted were accomodated by fragmentation within the polyene chain. On comparative chromatography (alumina paper) the symmetrical γ, γ -carotene (8) was adsorbed between β, γ -carotene (10) and β, γ -carotene.

Further details will be published.

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