## A Ternary W<sub>5</sub>Si<sub>3</sub>-type Phase in the Ni-Si-B System ARSIN AYDIN URAZ\* and STIG RUNDOVIST

Institute of Chemistry, University of Uppsala, Box 531, S-751 21 Uppsala 1, Sweden

In the course of studies of ternary Ni-Si-B alloys by one of the authors (A.A.U.) the occurrence of a new ternary phase was observed. Preliminary X-ray diffraction data indicated that the new phase might be isostructural with Co<sub>4.7</sub>Si<sub>2</sub>B and Fe<sub>4.86</sub>Si<sub>2</sub>B. Aronsson and Lundgren <sup>1</sup> have shown that Co4,7Si4B crystallizes with a structure closely related to the W.Si. type. The W<sub>5</sub>Si<sub>3</sub> structure is tetragonal and the space group is I4/mcm. The unit cell contains twenty tungsten atoms situated on one 16k and one 4b position, and twelve silicon atoms on one 8h and one 4a position. In Co4., Si<sub>2</sub>B, the silicon atoms occupy the 8h position and the boron atoms the 4a position. The 16k position is fully occupied by cobalt atoms, while the 4b position is only partially occupied by cobalt atoms. Aronsson and Engström 2 found an analogous situation in the structure of Fe<sub>4.86</sub>Si<sub>2</sub>B. It seemed possible that the Ni-Si-B phase might exhibit the same type of defect structure, and it was decided to analyze the structure in some detail.

A single crystal was selected from an alloy, which according to powder diffraction analysis contained only faint traces of Ni<sub>2</sub>Si in addition to the ternary phase. The unit cell dimensions were determined

from Guinier-Hägg type powder films, using  $\mathrm{Cu}K\alpha_1$  radiation and silicon (a=5.4305 Å) as the internal calibration standard. The crystal was mounted with the tetragonal axis as the rotation axis, and photographic intensity data for the layer lines (hk0) through (hk3) were recorded in a Weissenberg camera with filtered MoK radiation. The intensities of the reflexions were estimated visually. The absorption in the crystal was calculated to be very small, and no correction was considered necessary.

Preliminary calculations showed that the structure is closely similar to that of  $\text{Co}_4$ ,  $\text{Si}_2\text{B}$ . The structure data reported <sup>1</sup> for  $\text{Co}_4$ ,  $\text{Si}_2\text{B}$  were therefore taken as the starting values in a structure refinement by the least squares method. The calculations were made on a CDC 3600 computer using the program ORFLS.<sup>3</sup> Atomic scattering factors were interpolated from tables given in Ref. 4, and weights of the reflexions were applied according to the formula  $w=1/(a+|F_0|+c|F_0|^2)$ , with a=2.5 and c=0.1.

In the first stage of refinement, scale factors (one for each of the four layer lines), positional parameters and individual isotropic temperature factors were varied. The R value for the 117 observed reflexions did not drop below 0.15 and the temperature factor for the 4b nickel position was very large. In the case of Co<sub>4.7</sub>Si<sub>2</sub>B and Fe<sub>4.8</sub>Si<sub>2</sub>B strong evidence was presented <sup>1,2</sup> to show that the low scattering power of the 4b position is associated with metal vacancies rather than metal/nonmetal substitution. Accordingly, the degree of occupation of the 4b nickel position was included among the variable parameters in the next step of refinement. The R value now dropped to 0.085. The results of this refinement are presented in Table 1.

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As seen in Table 1, the temperature factor associated with the 4b position is negative. There is a strong correlation

Table 1. Structure data for Ni<sub>4.8</sub>Si<sub>2</sub>B. Space group I4/mcm. a=8.632 (2) Å, c=4.290 (1) Å.

	x	$oldsymbol{y}$	B(Å 2)	Occupation parameter
Ni 1 in 16k	0.0787(2)	0.2033(2)	0.34 (6)	
Ni 2 in 4b	<del>-</del> ' '	` ′	-0.08(10)	0.58(3)
Si in $8h$	0.1588(6)		0.58(11)	- '
B in $4a$		_	1.5 (8)	_

Standard deviations are given in parenthesis and refer to the last decimal places of the respective values.

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<sup>\*</sup> On leave from the Department of Physics, Faculty of Science, University of Ankara, Ankara, Turkey.

between the temperature factor and the occupational parameter, and it was found that a positive value for the temperature factor was obtained in a refinement, where the occupational parameter was held fixed at a value exceeding that given in Table 1

by one standard deviation.

With the fractional occupation of the 4bnickel position as obtained from the refinement, the composition of the crystal is calculated to be Ni<sub>4.8</sub>Si<sub>2</sub>B. While the phase relationships in the Ni—Si—B system still remain to be clarified in detail, the result of the present structure analysis is taken as a strong indication that the ternary W<sub>5</sub>Si<sub>3</sub>-type phase in this system exhibits a similar nonstoichiometric behaviour as the corresponding phases in the Co-Si-B and Fe-Si-B systems.

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## Equilibration of cis and trans 6-Methyl-2-oxo-2-ethoxy-1,2oxaphosphorinane

KNUT BERGESEN and ARNE BERGE

Chemical Institute, University of Bergen, Bergen, Norway

The conformational energies of substi-The conformational energies of tuents on cyclohexane rings have been widely studied, both because of the wide occurrence and importance of this ring system, and also because of the simplicity of the theoretical treatment of the data so obtained.1,2 In contrast to the abundance of information in the cyclohexane area, there has been, until very recently, a paucity of data concerned with conformational preferences in heterocyclic systems, with the result that relatively little is known regarding the conformational consequences of introducing one or more heteroatoms into a six-membered ring. 9,8 This paper reports the conformational equilibria of the cis and trans isomers of 6-methyl-2-ethoxy-2-oxo-1,2-oxaphosphorinane (II). The preparation of the geometric isomers has been reported in a previous paper.4

In analogy with cyclohexanes it has been found by X-ray diffraction method, and proton magnetic resonance studies that the 1,3,2-dioxaphosphorinane (I) ring 5-7 exists in a chair-like structure where the phosphoryl oxygen atom occupies the equatorial position.

$$\begin{array}{c}
R_3 \\
P=0 \\
R_1 \\
R_2
\end{array}$$

Proton magnetic resonance and infrared spectra of cis and trans isomers of 6methyl-2-oxo-2-alkoxy-1,2-oxaphosphorinanes indicate difference in the configuration around the phosphorus atom. The 6-methyl group is found to occupy the equatorial position in both isomers.