m.p. 144° alone or in admixture with an authentic specimen.

We thank Professor H. Erdtman for his interest in this work and Dr. I Wellings for checking the English.

- Fieser, L. F. and Oxford, A. E. J. Am. Chem. Soc. 64 (1942) 2060.
- Fieser, L. F., Berliner, E., Bondhus, F. J., Chang, F. C., Dauben, W. G., Ettlinger, M. G., Fawaz, G., Fields, M., Heidelberger, C., Heymann, H., Vaughan, W. R., Wilson, A. G., Wilson, E., Wu, M.-I., Leffler, M. T., Hamlin, K. E., Matson, E. J., Moore, E. E., Moore, M. B., and Zaugg, H. E. J. Am. Chem. Soc. 70 (1948) 3174, 3175.
- Erdtman, H. and Nilsson, M. Acta Chem. Scand. 10 (1956) 735.
- Erdtman, H. Proc. Roy. Soc. (London) A 143 (1933) 177, p. 186.
- Anderson, C. L., Horton, W. J., Walker, F. E. and Weiler, M. R. J. Am. Chem. Soc. 77 (1955) 598.

Received November 24, 1960.

The Crystal Structure of Hf₂Al₃

LARS-ERIK EDSHAMMAR

Institute of Inorganic and Physical Chemistry, University of Stockholm, Stockholm, Sweden

In the course of phase analyses and crystal structure determinations in the binary metal systems hafnium-aluminium 1-3 and zirconium-aluminium 1, a phase with the composition Hf₂Al₃ has been studied.

The structure has been determined from complete single crystal data using Patterson and electron density maps. The alloy was prepared by arc-melting and single crystals could be obtained from the crushed melt. The single crystal and Guinier powder data showed the structure to be orthorhombic with the following unit cell dimensions:

$$a = 9.52_{3} \text{ Å}$$
 $b = 13.76_{3} \text{ Å}$ $c = 5.52_{2} \text{ Å}$

The following structure was derived: Unit cell content: 8 Hf₂Al₃ (observed density 8.00, calculated density 8.04).

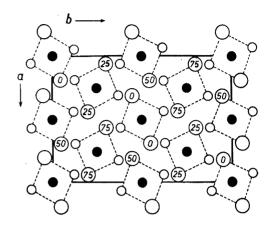


Fig. 1. The structure of $\mathrm{Hf_2Al_3}$ seen along the c-axis. The figures in the large circles indicate the height of the Hf-atoms in percent of c. Small open circles: Al at c/4 above the Hf atoms in the pyramid. Small black circles: Al at 0.39 c below the Hf atoms in the pyramid.

Space-group: Fdd2 (No. 43).

16 Hf in 16(b): x = 0.185 y = 0.052 z = 0.000 16 Al in 16(b): x = 0.185 y = 0.133 z = 0.500 8 Al in 8(a): x = 0 y = 0 z = 0.61

The structure may be described as built up by triangular-bipyramids, Hf₂Al₃, with the hafnium-atoms forming the apices of the pyramids (cf. Fig. 1).

An examination of the zirconium-aluminium system shows the existence of an isotypic structure Zr₂Al₃.

Full details on these investigations and a discussion of the structures in the hafniumaluminium system will be given elsewhere.

The author wishes to express his thanks to Dr. Arne Magnéli for his kind interest in this work.

- Edshammar, L.-E. and Andersson, S. Acta Chem. Scand. 14 (1960) 223.
- Edshammar, L.-E. Acta Chem. Scand. 14 (1960) 1220.
- 3. Edshammar, L.-E. Acta Chem. Scand. 15 (1961). In press.

Received November 26, 1960.