Note on the Preparation of Some 4-Amino-3.5-dimethyl- and 2-Amino-3-methylphenylurethanes

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Several amino- and methyl-substituted phenylurethanes (Types I and II) were required as starting material for certain syntheses in the field of local anesthetics.

 $R = C_2H_5$, $n-C_3H_7$, $n-C_4H_9$ and $n-C_5H_{11}$.

benzoic acid a (45 g, 0.25 mole) and thionyl chloride (89 g, 0.75 mole) for 4 h. The thionyl chloride was removed in vacuo and the residue (without purification) was used for further syntheses. An attempt to distill this compound in vacuo led to a violent explosion.

4-Nitro-3,5-dimethylbenzoyl azide. 3.5-dimethylbenzoyl chloride (42.7 g, 0.2 mole) was dissolved in acetone (80 ml) and cooled to 10°. Sodium azide (14.3 g, 0.22 mole) in water (45 ml) was then added dropwise with stirring over a period of 15 min.; the temperature was maintained at 25° by external cooling. After one hour, water (100 ml) was added, and the reaction product which separated was collected and dried in a vacuum desiccator. Yield 40.8 g

 \mathbf{II}

(93 %), m.p. $82-83^{\circ}$ (decomp.). Without purification the product was used for further syntheses.

These compounds were obtained by the Curtius degradation of the appropriate nitro-substituted carboxylic acid.

$$Ar - COCI \xrightarrow{\text{NaN}_3} Ar - CO \cdot N_3 \xrightarrow{\text{R} \cdot \text{OH}} Ar - NH \cdot CO \cdot OR$$

$$CH_3 \qquad CH_3 \qquad Or$$

$$CH_4 \qquad Or$$

Degradation was followed by hydrogenation of the nitro group to the amino compound.

Experimental. 4-Nitro-3,5-dimethylbenzoyl chloride was obtained from 4-nitro-3,5-dimethylbenzoic acid according to the method of Löfgren et al.¹. B.p. 144-145°/9 mm, yield

2-Nitro-3-methylbenzoyl chloride was similarly prepared by refluxing 2-nitro-3-methyl2-Nitro-3-methylbenzoyl azide was prepared

as above. Yield 98 %, m.p. $86-87^{\circ}$ (dec.). 4-Nitro-3,5-dimethyl- and 2-nitro-3-methylphenylurethanes. A solution of the appropriate azide (0.05 mole) in an alcohol (50 ml) containing the desired alkyl group was heated on the water bath until nitrogen evolution had ceased (2 h). Excess of the alcohol was removed in vacuo, leaving an oil which in most instances quickly solidified. The resulting product, obtained in yields of 95-100 %, was practically pure. A sample was recrystallised for analysis.

The nitro compounds are listed in Table 1. The oils of the three compounds which did not

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Ar	R M.	M 90	M.p.°C Recryst. solvent.		Calc. %			Found %		
		м.р. С			С	н	N	C	н	N
CH ₃	[C ₂ H ₅	77.5-78.5	L-B	C ₁₁ H ₁₄ N ₂ O ₄)	1	1	
	n-C ₃ H ₇	90.5 - 91.5	L-B	C ₁₂ H ₁₆ N ₂ O ₄	57.1	6.39	11.1	57.0	6.63	10.9
	$n ext{-}\mathrm{C_4H_9}$	104-105	L	C ₁₃ H ₁₈ N ₂ O ₄	58.6	6.81	10.5	58.6	6.90	10.6
CH ₃	n-C ₅ H ₁₁	83-84	L-B	C ₁₄ H ₂₀ N ₂ O ₄	60.0	7.19	10.0	60.1	7.10	9.90
CH, NO,	C ₂ H ₅	86-87	L	C10H12N2O4	53.6	5.39	12.5	53.5	5.86	12.5
	n-C ₃ H ₇	oil							1	1
	n-C ₃ H ₉	oil								
	n-C ₈ H ₁₁	oil								

Table 1. Nitro-substituted phenylurethanes, Ar-NH \cdot CO \cdot OR

a Aq, water; B, benzene; L, ligroin; M, methanol; P, petroleum ether.

Ar	R M.p	M . 90	Recryst. solvent	Formula	Calc. %			Found %		
		M.p.°C			C	н	N	C	н	N
CH ₃	C ₂ H ₅	97 – 98	M-Aq	C ₁₁ H ₁₆ N ₂ O ₂	63.4	7.75	13.5	63.5	7.78	13.4
NH _s ————————————————————————————————————	n-C ₃ H ₇	128 - 129	M-Aq	${ m C_{12}H_{18}N_2O_2}$	64.8	8.16	12.6	64.9	8.14	12.5
	n-C4H9	117-118	M-Aq	C ₁₃ H ₂₀ N ₂ O ₃	66.1	8.53	11.9	66.1	8.66	11.9
	n-C ₈ H ₁₁	81-82	L-B	C ₁₄ H ₂₂ N ₂ O ₂	67.2	8.86	11.2	67.7	9.10	10.9
CH ₈ NH ₂	C ₂ H ₅	75 – 76	L-B	C ₁₀ H ₁₄ N ₂ O ₂	61.8	7.27	14.4	62.0	7.41	14.1
	n-C ₃ H ₇	69 - 70	L-B	${ m C_{11}H_{16}N_2O_2}$	63.4	7.75	13.5	63.5	7.59	13.2
(n-C ₄ H ₉	71 - 72	L-B	${ m C_{12}H_{18}N_2O_3}$	64.8	8.16	12.6	64.9	7.71	12.7
	n-C5H11	72 - 73	L-P	${ m C_{13}H_{20}N_2O_2}$	66.1	8.53	11.9	66.1	8.60	11.8

Table 2. Amino-substituted phenylurethanes, Ar-NH · CO · OR

a See Table 1.

crystallise were used directly in the next synthetic step.

4-Amino-3,5-dimethyl- and 2-Amino-3-methylphenylurethanes. The appropriate nitro compund (0.05 mole) was dissolved in ethanol (50 ml). It was then hydrogenated over Raney nickel in a Parr hydrogenation apparatus for 2 h at about 55°C with a pressure of about 5 atm. The solvent was evaporated in vacuo and the crystalline residue (95-100 %

of practically pure product) was recrystal-lised.

The amino compounds are listed in Table 2.

- Löfgren, N., Tegnér, C., Willman, N. and Dahlbom, R. Unpublished results.
- Tomisek, A., Graham, B., Griffith, A., Pease, C. S. and Christensen, B. E. J. Am. Chem. Soc. 68 (1946) 1587.
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