Table 1. Endogenous phosphorylation. Each flask contained: 80 μ moles of glycylglycine buffer, pH 7.4; 20 or 30 μ moles of orthophosphate, pH 7.4; 5 μ moles of ADP; 5 μ moles of DPN; 0.03 μ moles of cytochrome c; 50 μ moles of glucose; 0.5 mg of hexokinase; 10 μ moles of MgCl₂, and 0.5 ml of mitochondria in a final volume of 2 ml. 50 μ moles of DL- β -hydroxybutyrate were added only where indicated. Incubated in air for 20 min at 30°. Other details given in Ref.².

$egin{array}{c} {f Added} \ {f substrate} \end{array}$	Oxygen uptake µatoms/mg N	Acetoacetate formed μ moles/mg N	Phosphate uptake $\mu \mathrm{moles/mg}$ N	P/O
None β -Hydroxybutyrate Substrate minus endogenous	3.1 8.5 5.4	$egin{array}{c} 0 \ 5.3 \ 5.3 \end{array}$	5.7 22.0 16.3	1.8 2.6 3.0

highest rate of endogenous phosphorylation were most stable to storage at 4°. Furthermore, certain pretreatments of liver mitochondria, such as freezing and thawing, or preincubation with low concentrations of phosphate which, among other detrimental changes (cf. Ref.7), abolished the endogenous activity, also decreased the stability of the stored preparations. The implication of these findings with regard to a possible correlation between endogenous phosphorylation and mitochondrial stability will be discussed.

- Weinbach, E. C. and Garbus, J. J. Biol. Chem. 234 (1959) 412.
- Weinbach, E. C. J. Biol. Chem. 234 (1959) 1580.
- Siekevitz, P. and Potter, V. R. J. Biol. Chem. 200 (1953) 187.
 Borgstrom, B., Sudduth, H. C. and Lehn.
- Borgstrom, B., Sudduth, H. C. and Lehninger, A. L. J. Biol. Chem. 215 (1955) 571.
- Chance, B. and Williams, G. R. J. Biol. Chem. 217 (1955) 383.
- Ernster, L., Jalling, O., Löw, H. and Lindberg, O. Exptl. Cell Research, Suppl. 3 (1955) 124.
- Hunter, F. E. Jr. and Ford, L. J. Biol. Chem. 216 (1955) 357.

Particulate and Soluble D-Lactic Cytochrome c Reductase of Yeast

Agnar P. Nygaard

Johan Throne Holst's Institutt for Ernæringsforskning, Blindern, Oslo, Norway

The main fraction of the D-lactic cytochrome c reductase activity (D-LDH) of aerobic yeast was associated with respiratory particles. By means of the extraction of lipids with n-butanol, acetone, and ether, 40 % of the enzyme

was solubilized. The solubilized enzyme has been purified one thousand fold. The difference spectrum of oxidized/reduced enzyme had absorption maximum at 450 m μ . At pH 3, flavin fluorescence appeared. The flavin concentration, calculated from the spectral difference, agreed well with that obtained from fluorescence intensity. In contrast to the flavohaemoprotein L-lactic cytochrome c reductase (L-LDH), there was no haem associated with the D-enzyme, and the D-enzyme did not reduce the haem of L-LDH. The turnover number of D-LDH was 20 000 min⁻¹ mole⁻¹, compared with 15 000 for L-LDH.

Particulate as well as solubilized D-LDH were absolutely specific for D-lactate. A preparation of D-LDH, which was obtained directly in a soluble form in the cell homogenate¹, has erroneously been determined to have relative stereospecificity². The apparent Michaelis constant of D-lactate for D-LDH was under certain conditions one thousand fold lower than that of L-lactate for L-LDH.

The oxidation of D-lactate to pyruvate went to completion with excess of cytochrome c. At a relatively high buffer concentration (μ 0.08), the initial rate of reaction has been determined for a series of cytochrome c concentrations, each with a series of D-lactate concentrations. The results can be summarized as follows: The apparent Michaelis constant of D-lactate was proportional to the cytochrome c concentration, and the rate, obtained with excess of lactate, was proportional to the cytochrome c concentration. These criteria. which were the same for soluble and particulate enzyme, are fullfilled in the following reaction scheme (where E = enzyme, SH2= lactate, and A = cytochrome c:

The fresh preparations of soluble D-LDH reduced ferricyanide at a rate which was 5 % of that with cytochrome c. On storage, the ferricyanide reduction was lost more rapidly than the cytochrome c reductase activity. This could either be ascribed to a modification of D-LDH, or to the presence of another enzyme. The aerobic "D-lactic ferricyanide reductase" had reaction properties strikingly different from those of the corresponding anaerobic enzyme.

- Nygaard, A. P. Biochim. et Biophys. Acta 40 (1960) 85.
- Nygaard, A. P. Arch. Biochem. Biophys. 88 (1960) 178.

Purification of Yeast Hexokinase by Cellulose Ion Exchanger Chromatography

Gunnar Ägren, Lorentz Engström and Sten Eklund

Institute of Medical Chemistry, University of Uppsala, Sweden

In an earlier communication from this laboratory, it was reported that radioactive phosphorylserine could be isolated from a hydrolysate of a crude yeast hexokinase preparation which had been incubated with its radioactive substrate, AT³²P or glucose-6-³²Pl. It seemed necessary to perform further studies with a more purified enzyme. Yeast hexokinase was isolated in a crystalline form several years ago ^{2,3}. By using cellulose ion exchanger chromatography, it now appears possible to obtain a more active preparation than the crystalline enzyme in an easy way.

As starting material, a commercial preparation (Sigma hexokinase, Type III) was used having a specific activity of 100-150 units per mg of preparation (estimated according to Kunitz et al.3). The activity was raised by two consecutive chromatographic separations using carboxymethyl cellulose columns (Serva Entwicklungslab., Germany) to between 1 200 and 1900 units/mg protein, respectively. The elution was performed at pH 5 with a gradient increase in the concentration of the acetate buffer $(0.02 \text{ N} \rightarrow 0.30 \text{ N})$ with regard to sodium acetate). The enzyme was then chromatographed on a column of diethylaminoethyl cellulose (Eastman Org. Chem., U.S.A.). It was eluted with a potassium phosphate buffer of pH 7.0, increasing gradually from 0.05 M to 0.30 M. The enzyme was eluted as a main peak with nearly constant specific activity from fraction to fraction. No inactive protein could be separated from the enzyme activity by similar rechromatography of the main part of the enzyme peak. The specific activity was about 3 000 units/mg protein (as compared to about 1 400 units/mg protein found by Kunitz et al.³). The average yield in the three step purification was about 10 %.

When a sedimentation analysis was carried out in a Spinco ultracentrifuge, Model E, only one boundary was obtained. A single determination of the molecular weight was performed according to Archibald 4, and it was found to be about 50 000.

- Ågren, G. and Engström, L. Acta Chem. Scand. 10 (1956) 489.
- Berger, L., Colowick, S. R. and Cori, C. F. J. Gen. Physiol. 29 (1946) 379.
- Kunitz, M. and McDonald, M. R. J. Gen. Physiol. 29 (1946). 393.
- Archibald, W. J. Phys. & Colloid Chem. 51 (1947) 1204.

On Electron Transport and Phosphorylation in Plant and Bacterial Light-Induced Phosphorylation

Herrick Baltscheffsky, Margareta Baltscheffsky and Barbro Arwidsson

Wenner-Gren Institute, University of Stockholm, Stockholm, Sweden

The following findings from experiments on light-induced phosphorylation (aerobic conditions, saturating light-intensities) will be discussed:

- (1) Coenzyme Q_o stimulates light-induced phosphorylation in isolated spinach chloroplasts. It is, however, less efficient as stimulatory agent than menadione. At 10^{-4} M concentration, for example, coenzyme Q_o gives only 70 % of the stimulation obtained with the same concentration of menadione.
- (2) Valinomycin, which completely uncouples oxidative phosphorylation in animal mitochondria at low concentrations ¹, inhibits light-induced phosphorylation in isolated chromatophores of the photosynthetic bacterium *Rhodospirillum rubrum* to approximately 50 %. This may indicate that two phos-